

Multiple response optimization of blueberry juice depectinization

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ABSTRACT: To obtain blueberry juice with a high content of antioxidants it is necessary to introduce an enzymatic depectinization step into the process. Due to the importance of this step in the final properties of blueberry juice it is critical that the operation conditions be optimized. The aim of this research was to evaluate the effects of temperature, duration of treatment and enzymatic complex concentration on anthocyanin content and juice yield during enzymatic depectinization. Results indicated that the best factor combination was 50°C during 1.3h and 4mg 100g⁻¹ of LAFASE® CLARIFICATION and 8mg 100g⁻¹ of LAFASE® HE GRAND CRU enzymatic complex concentration. Under these conditions, blueberry juice with 798.41±8.03mg of cyanidin-3-glycoside L⁻¹ and a juice yield of 87% was obtained. The combination of the response surface and desirability function methodologies enabled the optimization of the blueberry juice during the depectinization step, in terms of anthocyanin content and juice yield.

Key words: blueberry juice, enzymatic depectinization, multiple response optimization.

Otimização de múltiplas respostas durante a despectinização do suco de mirtilo

RESUMO: Para obter o suco de mirtilo com um alto teor de antioxidantes, é necessário realizar uma etapa de despectinização enzimática durante o processo. Esta etapa influenciará nas propriedades finais do suco de mirtilo, então, é necessário que as condições de operação sejam otimizadas. O objetivo deste trabalho foi avaliar os efeitos da temperatura, duração do tratamento e concentração do complexo enzimático na concentração de antocianinas e no rendimento do suco durante a despectinização enzimática. Os resultados indicaram que a melhor combinação de parâmetros foi de 50°C, durante 1 à 3h e uma concentração de complexo enzimático de 4mg 100g⁻¹ de LAFASE® CLARIFICATION e 8mg 100g⁻¹ de LAFASE® HE GRAND CRU. Sob estas condições, foi obtido o suco de mirtilo com 798.41±8.03mg de cianidina-3-glicosídeo L⁻¹ e um rendimento de suco de 87%. A combinação das metodologias de superfície de resposta em função da preferência possibilitaram a otimização da despectinização do suco de mirtilo, em termos de teor de antocianinas e rendimento de suco.

Palavras-chave: suco de mirtilo, despectinização enzimática, otimização de múltiplas respostas.

INTRODUCTION

Blueberries (*Vaccinium corymbosum*) are a different phytonutrients rich dietary source. Among the many phenolic compounds, anthocyanins have a great amount of attention since they possess potent antioxidant activity (PRIOR et al., 1998; SEERAM, 2008; VRHOVSEK et al., 2012). Anthocyanins have shown an important role in the prevention of macular degeneration (TREVITHICK & MITTON, 1999), neuronal diseases, cardiovascular diseases, cancer, diabetes (CHAMBERS & CAMIRE, 2003; KRAFT et al., 2005; MARTINEAU et al., 2006; NETO, 2007; SHUKITT-HALE et al., 2008; PATRAS et al., 2010) and urinary tract disorders (KALT & DUFOUR, 1997;

HOWELL et al., 2005; JEPSON & CRAIG, 2007). Because of their beneficial role as micronutrients, it is of utmost importance to measure the changes in polyphenolics during processing to better assess the dietary value of the processed products (RAWSON et al., 2011; HOWARD et al., 2012).

During blueberry juice processing, one of the main problems is gelification due to high pectin concentration. Rapid gelation process prevents the application of subsequent unit operations as filtration, pasteurization, concentration, and others. Fruit juice depectinization through the use of pectinases has been presented as an efficient alternative to reduce turbidity and enhance juice yield (SANDRI et al., 2011, 2013). However, a significant loss of

anthocyanin content was observed during juice processing (SRIVASTAVA et al., 2007). In addition, information on how different processing steps affect the content of bioactive compounds in the final product is limited (BROWNMILLER et al., 2008; SABLANI et al., 2010).

A useful tool for process optimization that enables determination of the optimal conditions for multiple influential factors with a limited number of experiments is response surface methodology (SHI & YU, 2005; VARRONE et al., 2012). This methodology has been successfully used to optimize clarification of carambola fruit (LIEW ABDULLAH et al., 2007), pectin extraction from lemon by-products (MASMOUDI et al., 2008) and enzymatic clarification of banana juice (LEE et al., 2006) among other applications. However, it has not been utilized to optimize the blueberry juice depectinization process.

The aim of this work was to evaluate the effects of temperature, duration of treatment and enzymatic complex concentration on anthocyanins content and juice yield during enzymatic depectinization.

MATERIAL AND METHODS

Sample preparation

Snow chaser blueberries were harvested and placed in polyethylene terephthalate trays and immediately introduced into a cooling chamber at $0\pm 0.5^{\circ}\text{C}$ until use. Blueberries were washed, weighed and then crushed with a food processor (MR 400 Plus, Braun, Spain). An enzymatic depectinization with two commercial enzymatic packs, LAFASE[®] CLARIFICATION (E1) and LAFASE[®] HE GRAND CRU (E2) (Laffort, France) was carried out. Finally, the juice was filtered through a 0.5mm sieve and then centrifuged (ALRESA Digicen, Álvarez Redondo S.A., Spain) at $2706 \times g$ for 20 minutes. The solid residues were discarded and the supernatant is referred to as blueberry juice.

Anthocyanins

The pH differential spectrophotometric method developed by GIUSTI & WROLSTAD (2001) was used to assess the total monomeric anthocyanin (TA) content. Aqueous buffers with pH 1.0 (KCl 0.025M) and 4.5 (CH_3COONa 0.4M) were utilized to dilute the samples. Absorbance measurements were taken at 510 and 700nm against a blank cell filled with distilled water. A $2690\text{m}^2 \text{mol}^{-1}$ molar extinction coefficient was used for the Cyanidin-3-O-glucoside. Results were expressed as milligrams of Cyanidin-3-O-glucoside equivalent per liter of juice.

Juice yield (JY)

Mass relationship between weight of juice and initial blueberry weight was calculated. Juice yield was expressed as grams of juice per 100g blueberries.

Experimental design

An experimental design with STATGRAPHICS Centurion XV package was carried out. Four factors were studied in the following experimental range: temperature (T): $0-100^{\circ}\text{C}$; time (t): $0-2.5\text{h}$; concentration of enzymatic pack LAFASE[®] HE GRAND CRU (E1): $0-8\text{mg } 100\text{g}^{-1}$ blueberries and LAFASE[®] CLARIFICATION (E2): $0-16\text{mg } 100\text{g}^{-1}$ blueberries. Total monomeric anthocyanin content and juice yield were selected as response variables. Using a Draper-Lin small composite design, a total of 34 experiences were carried out, each by triplicate. Results were analyzed by response surface methodology. In order to maximize anthocyanin content and juice yield, the level of each factor was optimized by the desirability function approach.

RESULTS AND DISCUSSION

Effects of experimental factors on total anthocyanins concentration

Results showed a wide variation of total anthocyanin concentration and juice yield as a function of the experimental conditions (Table 1), which indicated the relevance of this optimization study. The highest concentration of total anthocyanins occurred when the experiment was carried out at 50°C during 1.3h and 4 and 8mg 100g^{-1} of blueberry E1 and E2 enzyme concentration respectively (Table 1). The experimental data were analyzed using multiple regression, resulting in Equation 1 where only the significant factors were included.

$$AT = 21933 + 1662.T - 80719.t - 2721.E1 - 1384.E2 - 0.021.T^2 + 0.10.T.t - 207.55.T.E1 - 103.78.T.E2 - 26.95.t^2 + 10097.t.E1 + 5049.t.E2 \quad (1)$$

Statistical analysis revealed that this experimental design described the relationship between total anthocyanins and the experimental factors ($P < 0.05$) adequately. The high value of R^2 (0.975) indicated that only 2.5% of the total variation was not explained. Temperature (T), time (t), enzyme concentration (E1 and E2), T^2 and t^2 , as well as T-t, T-E1, T-E2, t-E1, t-E2 interactions were significant ($P < 0.05$).

In order to visualize the effects of the factors on total anthocyanins response, contour

Table 1 - Blueberry juice depectinization experimental design.

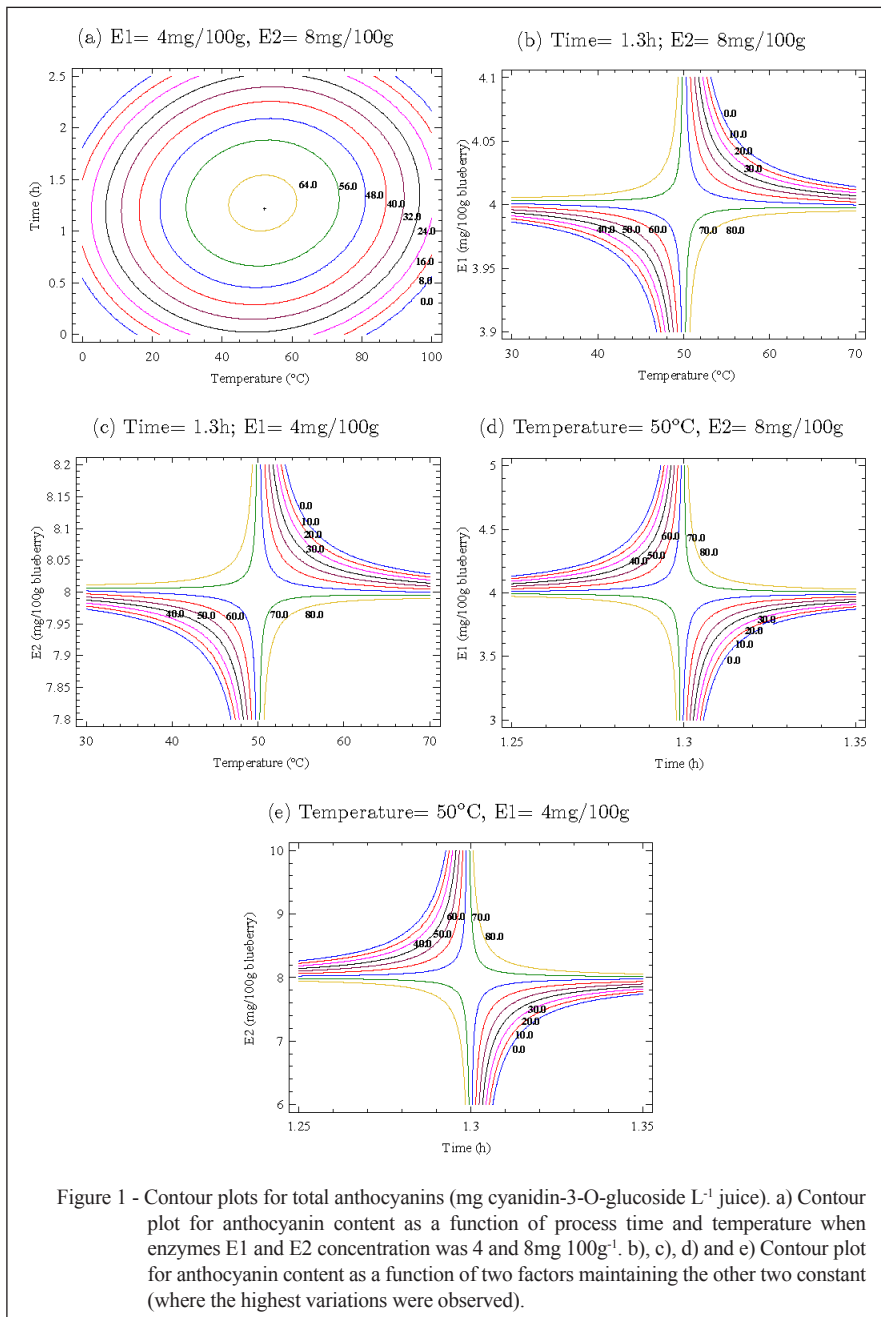
Temperature (°C)	Time (h)	E1 (mg 100g ⁻¹)	E2 (mg 100g ⁻¹)	Total anthocyanins (mg Cyd-3-O-glu L ⁻¹)	Juice Yield (g 100g ⁻¹)
0	1.3	4	8	89.5±0.3 ^a	76.0±2.6 ^{cde}
20	0.5	0	16	360.8±33.0 ^d	79.1±1.2 ^{def}
20	0.5	8	0	330.6±5.5 ^d	46.5±1.5 ^a
20	2	0	16	336.0±24.0 ^d	81.9±1.4 ^{ef}
20	2	8	0	264.5±17.4 ^c	80.1±0.3 ^{def}
50	0	4	8	210.5±2.3 ^b	70.7±0.0 ^c
50	1.3	0	0	319.5±26.0 ^d	69.4±2.2 ^c
50	1.3	4	8	659.2±21.4 ^e	74.6±4.0 ^{cd}
50	1.3	8	4.8	358.2±37.7 ^d	85.3±2.3 ^f
50	2.5	4	8	254.1±25.1 ^{bc}	75.1±1.1 ^{cde}
80	0.5	8	0	337.3±2.4 ^d	56.3±0.3 ^b
80	0.5	0	16	328.3±23.8 ^d	61.1±1.1 ^b
80	2	0	16	357.5±23.9 ^d	47.5±2.4 ^a
80	2	8	0	354.2±25.4 ^d	47.2±4.6 ^a
100	1.3	4	8	244.4±16.6 ^{bc}	70.1±1.9 ^c

plots were generated by varying two factors and holding the other two constant (Figure 1). Figure 1a allowed the identification of the anthocyanin maximum concentration area in this experimental design. Temperature and treatment time were increased to 60°C and 1.4h, respectively, resulting in a higher concentration of total anthocyanins. However, when these factors values were higher than the aforementioned ones, a decrease in total anthocyanins was observed. These results could be explained if a combination of phenomena is considered. Firstly, an increase of anthocyanin solubility and diffusivity coefficient resulted from temperature increase, as suggested by CACACE & MAZZA (2003) and BRAMBILLA et al. (2011). Secondly, a temperature dependent degradation process that becomes significant above 60°C is probably due to the loss of the glycoside molecule from position three of the anthocyanin structure and the polyphenolic ring opening (FENNEMA, 2000). Regarding treatment duration, a longer treatment caused an increase in total anthocyanin concentration until the extraction and degradation rates were even as observed by other authors (IBARZ RIBAS et al., 2000; ROMERO CASCALES, 2008). As for commercial enzyme concentration and temperature, an increase in any factor caused an increase of total anthocyanin content (Figure 1b) until 4mg 100g⁻¹ E1 concentration and 48°C were reached, while higher E1 concentration and temperature resulted in a total anthocyanin

decrease. A similar behavior was observed in enzyme E2 but at 8mg 100g⁻¹ concentration (Figure 1c). Finally, it was observed that the highest anthocyanin content was found within 76-79min treatment range and with 3.6-4.4mg 100g⁻¹ E1 and 7-8mg 100g⁻¹ E2 concentrations respectively (Figure 1d and Figure 1e). With regards to the added enzyme concentration, ROMERO CASCALES (2008) noted that while some studies showed a color increase in the vinification process where pectolytic exogenous enzymes were applied, others have reported no benefits. This may be due to a large heterogeneity among different commercial preparations. These authors also reported that other studies had obtained wines with more color as they worked with minimal enzyme doses (0.3%) and wines with less color when they worked with high enzyme doses (0.7%). They suggested that this effect could be caused by enzyme preparations containing β -glucosidase activity, that apparently had a bleaching effect on pigments extracted from various fruits.

Effect of the experimental factors on juice yield

There is a wide variety of results ranging from 46 to 85%. The lowest juice yield value was obtained with treatments shorter than 30min or longer than 120min and at temperatures higher than 80°C (Table 1). The highest juice yield value was obtained by using the following factor combination: 50°C,



1.3-1.5h treatment and 4 and 8mg 100g⁻¹ enzyme concentration for E1 and E2 respectively.

Multiple regression analysis was carried out resulting in a second order polynomial equation (Equation 2). Only the significant terms were included in the equation.

$$JY = 36350 + 2378.T - 119371.t - 4488.E1 - 2363.E2 - 0.002.T^2 - 0.28.T.t - 297.T.E1 - 145.T.E2 - 14925.t.E1 + 7461.t.E2 - 8.95.E1.E2 \quad (2)$$

The high $R^2 = 0.885$ indicates that the fitting quality was satisfactory. The terms T, t, E1, E2, T², T t, T E1, T E2, t E1, t E2 and E1 E2 were significant ($P < 0.05$).

The contour plots generated for juice yield showed that increasing temperature to 70°C and treatment duration to 1,3h caused an increase in juice yield. A juice yield decrease was observed above these conditions (Figure 2a).

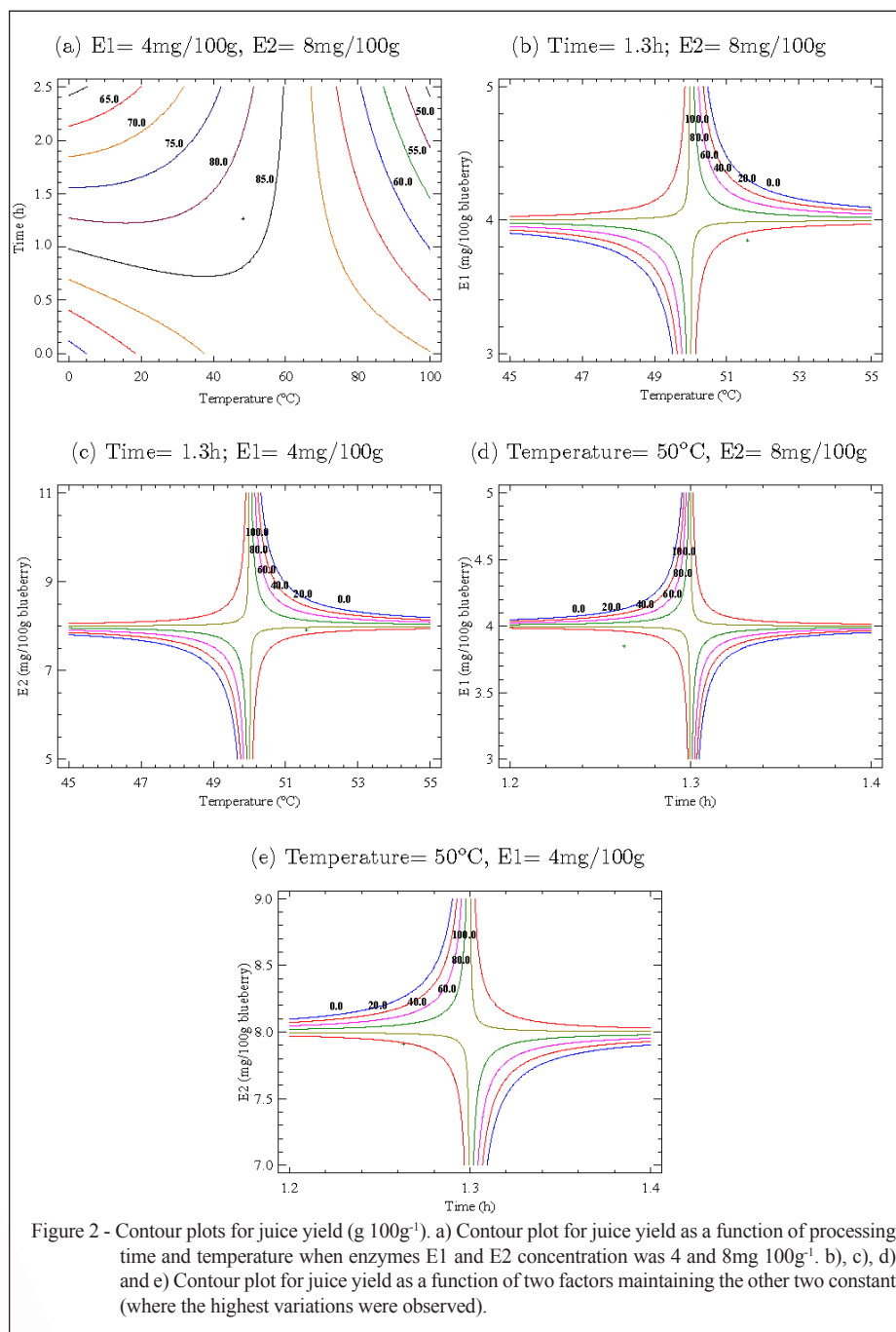
Temperatures within the 49-51°C range and E1 enzyme concentration within 3.6 to 4.4mg 100g⁻¹

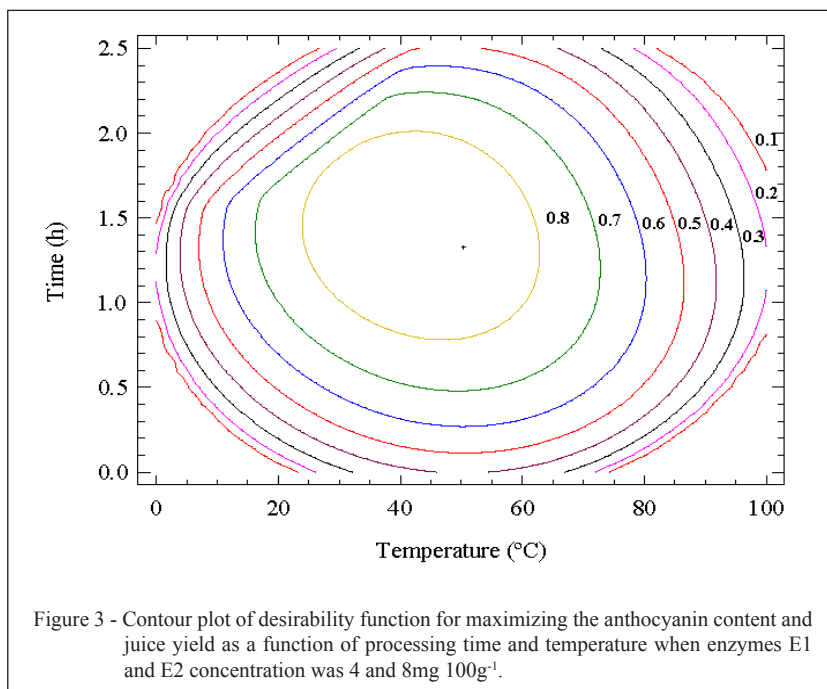
led to an increase in juice yield; while a decrease of the response variable was observed above the aforementioned values (Figure 2b). A similar behavior was observed for E2 between 7 and 9 mg 100g⁻¹ blueberries (Figure 2c). In the case of E1 and E2 as a function of treatment time, the contour plots showed similar results to those mentioned above (Figure 2d and Figure 2e). Neither a short

time treatment (less than 0.5h) nor a long treatment (above 2h) rendered a high yield value.

Optimization of enzymatic depectinization

By applying the desirability function method, the optimum conditions for obtaining blueberry juice were determined (Figure 3). The desirability function value obtained was 0.9583.





Optimum conditions that provided the highest values of total anthocyanins (712mg of cyanidin-3-glucoside 1L⁻¹ of juice) and juice yield (85%) were: temperature = 50°C; duration = 1.3h; E1 concentration = 4mg 100g⁻¹; E2 concentration = 8mg 100g⁻¹.

Optimized conditions obtained through the application of the desirability function were utilized to obtain blueberry juice. In addition, an adjustment to the process was introduced: the vessels head space was reduced to a minimum in order to avoid the contact with oxygen during the depectinization stage.

The blueberry juice obtained had a 798.41±8.03mg cyanidin-3-glucoside 1L⁻¹ concentration and the juice yield was 87%. This anthocyanin concentration was slightly higher than the value predicted by the desirability function and 20% higher than the values obtained during the experimental design. This better result could be related to the smaller head space left in the vessels during the last experiment. Oxygen is well known as a destabilizing agent in processed products containing anthocyanins (FRANCIS, 1989), and it also has a negative effect on the antioxidant capacity of blueberry juice (KALT et al., 2000).

CONCLUSION

Response surface methodology and desirability function enabled the evaluation of a wide range of temperature, duration of treatment

and enzyme concentration. The best combination of process conditions to obtain total anthocyanin and juice yield highest values were: temperature = 50°C; duration = 1.3h; E1 concentration = 4mg/100g; E2 concentration = 8mg/100g within the studied experimental range.

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