

Article

Complete ^1H - and ^{13}C - Resonance Assignment of Methyl $2\alpha,3\beta,24$ -Tri-O-acetylurs-12-en-28-oate and Methyl $2\alpha,3\beta,24$ -Tri-O-acetylolean-12-en-28-oate by NMR Spectroscopy

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Received: August 1, 1997

O trabalho descreve o estudo de dois triterpenos isômeros (**1**: $2\alpha,3\beta,24$ -tri-O-acetil-12-eno-28-ursolato de metila e **2**: $2\alpha,3\beta,24$ -tri-O-acetil-12-eno-28-oleanato de metila) efetuando a completa atribuição dos deslocamentos químicos dos hidrogênios e carbonos. Foram utilizadas técnicas de ressonância magnética nuclear (^1H e ^{13}C) uni- e bidimensionais empregando os seguintes passos: a) Análise comparativa dos espectros de RMN ^{13}C -PND e RMN ^{13}C -DEPT para identificação dos carbonos quaternários, metínicos, metilênicos, e metílicos; b) aplicação da técnica $^1\text{H} \times ^{13}\text{C}$ HMBC [acoplamento de hidrogênio e carbono-13 via duas ($^2\text{J}_{\text{CH}}$) e três ($^3\text{J}_{\text{CH}}$) ligações] para atribuição dos deslocamentos químicos dos espectros de ^{13}C ; c) uso dos espectros $^1\text{H} \times ^{13}\text{C}$ HMQC [interação spin-spin de hidrogênio e carbono-13 via uma ($^1\text{J}_{\text{CH}}$) ligação] para determinar os deslocamentos químicos dos átomos de hidrogênio e para confirmar os dos carbonos hidrogenados; d) uso dos espectros bidimensionais de correlação homonuclear de hidrogênio e hidrogênio ($^1\text{H} \times ^1\text{H}$ -COSY), e de efeito nuclear Overhauser homonuclear de hidrogênio e hidrogênio ($^1\text{H} \times ^1\text{H}$ -NOESY) para confirmar os sinais de hidrogênios e para determinações configuracionais (α e β) dos hidrogênios metilênicos e metínicos e, e) análise dos padrões de desdobramento (multiplicidade e constante de acoplamento) nos espectros unidimensionais para confirmar os sinais de vários átomos de hidrogênio. Foram descritas as condições experimentais dos aparelhos utilizados (RMN 500 MHz, para hidrogênio e 125 MHz, para carbonos) e de isolamento dos constituintes **1** e **2**. Todos os resultados são resumidos na forma de tabelas.

The complete ^1H and ^{13}C chemical shift assignments of extended hydrogen spin systems in triterpenoid derivatives (methyl $2\alpha,3\beta,24$ -tri-O-acetylurs-12-en-28-oate and methyl $2\alpha,3\beta,24$ -tri-O-acetylolean-12-en-28-oate) was accomplished making use of one and two dimensional NMR techniques (HMBC, HMQC, COSY and NOESY).

Keywords: triterpenoids, complete ^1H and ^{13}C -NMR signal assignments; 1D NMR; 2D NMR ($^1\text{H} \times ^{13}\text{C}$ -HMBC, $^1\text{H} \times ^{13}\text{C}$ -HMQC, $^1\text{H} \times ^1\text{H}$ -COSY, $^1\text{H} \times ^1\text{H}$ -NOESY)

Introduction

The extraordinary advances made in spectroscopic techniques have enormously accelerated the research in the field of isolation and structure elucidation of complex natural products. Of all the physical methods, the NMR

technique has greatly changed during the last two decades mainly by introduction and development of multiple pulse and 2D NMR. Consequently, a large number of triterpenes have been examined by NMR spectroscopy and much chemical shift data has been accumulated and utilized in further investigations of these natural products. On the

other hand, the use of these secondary metabolites as therapeutic agents has been the subject of extensive exploratory activities during recent years¹. The ¹H and ¹³C triterpene chemical shifts provided useful information concerning conformations and configurations of these complex organic derivatives and are also useful for the better understanding of the correlations between their molecular conformations and their biological activities².

In this paper we report an extensive NMR study of two isomeric (C₃₇H₅₆O₈) triterpenoids derivatives with their complete ¹H and ¹³C signal assignments, by application of 1D and 2D spectral experiments. The compounds investigated were methyl 2 α ,3 β ,24-tri-O-acetyllurs-12-en-28-oate (1) and methyl 2 α ,3 β ,24-tri-O-acetylolean-12-en-28-oate (2) obtained after acetylation and methylation of a mixture isolated from *Mentha villosa* Huds³. This plant is used as a remedy in the treatment of amebiasis, giardiasis⁴ and shistosomiasis⁵.

This is the first report giving the complete assignment of these pentacyclic triterpenoid derivatives of ursolic and oleanolic acids. The following steps were concurrently employed: a) comparative analysis of the ¹³C-NMR-PND and ¹³C-NMR-DEPT for identification of quaternary, methine, methylene and methyl carbon atoms; b) application of the HMBC experiment to chemical shift assignment of the ¹³C spectra; c) use of the HMQC spectra to determine the chemical shifts of the hydrogen atoms and to confirm

those of the hydrogenated carbons; d) use of hydrogen ¹H x ¹H - COSY and ¹H x ¹H - NOESY maps to confirm the ¹H assignments (and, indirectly, also the ¹³C assignments) and to establish the configurational assignment (α and β) of all methylene and methine hydrogens and e) analysis of the splitting patterns (multiplicity and coupling constant) in the 1D NMR spectra to confirm the resonances (including the configurational assignment) of various hydrogen atoms.

Experimental

Plant material

Mentha villosa was collected in the "Horto de Plantas Medicinaias" of the "Universidade Federal do Cear ", Fortaleza, Brazil. A voucher of the plant (N. 16.545) is deposited in the Herbarium "Prisco Bezerra" of the Departamento de Biologia of the Universidade Federal do Cear .

Isolation procedure

The fraction obtained from the ETOH extract by partition with CHCl₃ was successively chromatographed on silica gel column to afford fraction E (eluted with hexane-CHCl₃ 2:8). Fraction E was methylated with CH₂N₂ and then acetylated with Ac₂O/pyridine in the usual manner to yield a product named E-MeAc. Silica gel preparative TLC of E-MeAc lead to a fraction (R_f 0.60, eluted with CH₂Cl₂),

Table 1. NMR data for methyl 2 α ,3 β ,24-tri-O-acetyllurs-12-en-28-oate (1)*.

C	¹ H x ¹³ C-HMQC (¹ J _{CH})		¹ H x ¹³ C-HMBC		¹ H x ¹ H-NOESY
	δ_C	δ_H	² J _{CH}	³ J _{CH}	
4	43.00	---	3H-23, 2H-24		---
8	39.30	---	3H-26	3H-27	---
10	37.50	---	3H-25		---
13	138.60	---		3H-27	---
14	41.80	---	3H-27	3H-26	---
17	48.00	---			---
28	178.00	---		MeO-28	---
AcO-2	170.80	---	H ₃ CCO ₂ -2	H-2	---
AcO-3	171.00	---	H ₃ CCO ₂ -3	H-3	---
AcO-24	171.10	---		2H-24	---
CH					
2	69.28	5.16 (dt, J = 12.0, 4.8 Hz, H-2 β)			H-1 β , 2H-24, 3H-25
3	79.90	4.82 (d, J = 12.0 Hz, H-3 α)		3H-23, 2H-24	H-1 α , H-5 α , 3H-23
5	55.52	1.07 (H-5 α)		2H-24, 3H-25	H-3 α , H-7 α , H-9 α
9	47.60	1.57 (H-9 α)		3H-25, 3H-26	H-1 α , H-5 α , 3H-27
12	125.00	5.24 (t, J = 3.5 Hz)			2H-11, H-18 β

Table 1. (cont.)

18	52.85	2.24 (d, J = 11.4 Hz, H-18β)		3H-29	H-12, H-22β, 3H-29
19	39.10	1.32 (H-19α)	3H-29	3H-30	3H-27, 3H-29, 3H-30
20	38.87	0.98 (H-20β)	3H-30	3H-29	3H-29, 3H-30
CH₂					
1	44.32	2.11 (dd, J = 12.4, 4.8 Hz, H-1β) 1.10 (t, J = 12.4 Hz, H-1α)		3H-25	H-1α, H-2β, 3H-25 H-1β, H-3α, H-9α
6	19.20	1.64 (H-6α) 1.47 (H-6β)			H-6β, H-7α, 3H-23 H-6α
7	32.20	1.48 (H-7α) 1.37 (H-7β)		3H-26	H-5α, H-6α, H-7β, 3H-27 H-6β, H-7α, 3H-26
11	23.30	1.92 (m, H-11α, H-11β)			H-12, 3H-25, 3H-26
15	27.90	1.76 (dt, J = 14.0, 4.9 Hz, H-15β) 1.04 (H-15α)		3H-27	H-15α, 3H-26 H-15β
16	24.20	2.02 (dt, J = 12.4, 4.9 Hz, H-16α) 1.67 (H-16β)			H-16β, H-19α, H-22α H-16α
21	30.65	1.47 (H-21β) 1.30 (H-21α)		3H-30	H-21α, H-22β H-16α, H-21β, 3H-30
22	36.60	1.66 (H-22α) 1.58 (dt, J = 10.4, 4.2 Hz, H-22β)			H-16α, H-22β H-18, H-21β, H-22α
24	65.50	4.20 (s, 2H-84)		3H-23	H-2β, 3H-23, 3H-25
CH₃					
23	23.10	1.03 (s)		2H-24	H-3α, 2H-24
25	16.28	1.07 (s)			H-6β, 2H-11, 2H-24, 3H-26
26	16.67	0.74 (s)			H-6β, H-7β, 2H-11, H-15β, 3H-25
27	23.50	1.06 (s)			H-7α, H-9α
29	16.95	0.84 (d, J = 6.4 Hz)			H-18β, H-19α, H-20β
30	21.15	0.94 (d, J = 6.0 Hz)			H-19α, H-20β, H-21β
AcO-2	21.01	1.97 (s)			
AcO-3	20.80/21.01	2.04/2.05 (s)			
AcO-24	20.80/21.01	2.04/2.05 (s)			
MeO-28	51.47	3.60 (s)			

* Multiplicity of signals of carbon atoms deduced by comparative analysis of PND- and DEPT-¹³C-NMR. Chemical shifts of hydrogen atoms obtained from 1D ¹H-NMR. The 2D ¹H x ¹H - COSY and 2D ¹H x ¹³C - HMQC spectra were also used in these assignments.

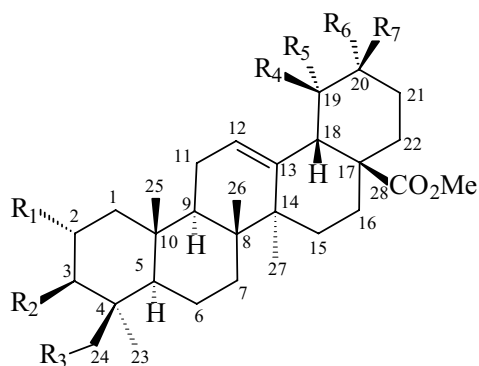
identified as a mixture of the triterpene derivatives (**1** and **2**) which were separated by preparative HPLC Waters model 6000A, detector R - 401 differential refractometer; RP 18 (250 x 9.4 mm) column; mobile phase: methanol-water 98:2, flow-rate of 2 mL min⁻¹.

NMR spectra

¹H- and ¹³C-NMR experiments were performed on a BRUKER ARX 500 spectrometer working at 500.1 MHz

for hydrogen and 125.75 MHz for ¹³C carbon, using CDCl₃ as solvent. Solutions were made from 0.35 ml of CDCl₃ and 2-8 mg of triterpenes with TMS as the internal standard. For all experiments the temperature was stabilised at 298 K. For the NOESY experiments, the samples were degased by bubbling nitrogen through the solution and fitting a teflon serum cap. The 2D experiments were acquired and processed with the software provided by BRUKER on ASPECT X32.

Typical acquisition and processing conditions for COSY and NOESY experiments were: relaxation delay of 1 to 2 seconds, 512 to 1024 t_1 increments; 1024 to 2048 t_2 points; sweep width of 6 ppm. Sine bell squared and shifted ($\pi/4$, $\pi/6$ and $\pi/8$) apodization functions were used for processing. The mixing time in the NOESY experiments, generally set at 1.2-1.5 seconds, was also varied between 0.8 and 2 seconds, without substantial change in the results. For $^1\text{H} \times ^{13}\text{C}$ (^{13}C detected) and $^{13}\text{C} \times ^1\text{H}$ (^1H detected) correlations, the same relaxation delay was used, 512 to 1024 t_1 increments, 1024 to 2048 t_2 points, the sweep width being respectively 7 ppm for ^1H and 180 ppm for ^{13}C . Lorentzian and Gaussian deconvolution were generally used in the processing. The number of scans was set for an overall acquisition time of about 12 h to 16 h.



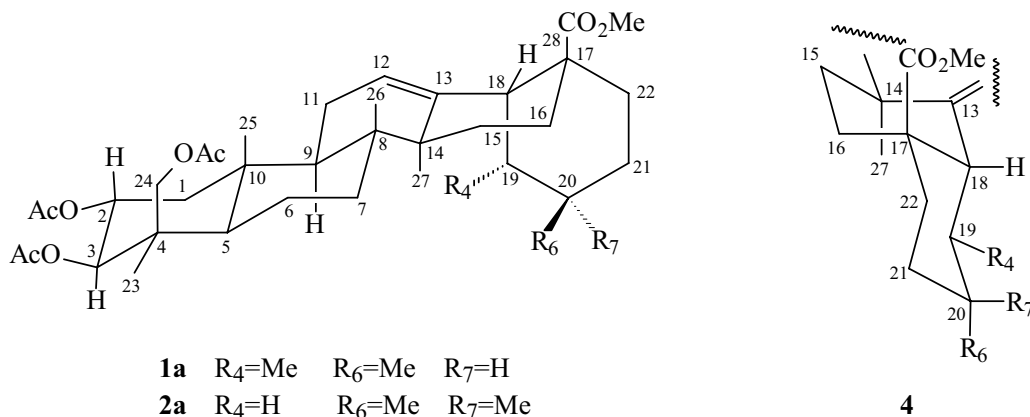
	R ₁	R ₂	R ₃	R ₄	R ₅	R ₆	R ₇
1	OAc	OAc	OAc	Me	H	Me	H
2	OAc	OAc	OAc	H	H	Me	Me
3	OAc	OAc	H	H	H	Me	Me

Figure 1. Structures of 2 α ,3 β ,24-tri-O-acetylsurs-12-en-28-oate (1), 2 α ,3 β ,24-tri-O-acetylolean-12-en-28-oate (2) and 2 α ,3 β -di-O-acetylolean-12-en-28-oate (3).

Results and Discussion

An essential prerequisite to the unambiguous assignment of ^1H chemical shifts from ^{13}C - ^1H shift correlated spectra is to first unambiguously assign the ^{13}C chemical shifts of protonated carbons^{6,7}.

The signals corresponding to quaternary, methine, methylene and methyl carbon atoms were identified by comparative analysis of the ^{13}C -NMR - PND and ^{13}C - NMR DEPT spectra. The $^1\text{H} \times ^{13}\text{C}$ - HMBC Heteronuclear Multiple Connectivity - coupling of hydrogen and carbon-13 via two ($^2J_{\text{CH}}$) and three ($^3J_{\text{CH}}$) bonds spectra were successfully used to attribute the chemical shifts of several protonated as well as, of almost all non-protonated carbons. For example, the singlet at δ_{H} 4.20 in the ^1H -NMR spectrum of **1** (Fig. 1) was correlated with hydrogens attached at an oxygenated carbon. Assuming this resonance assignable to 2H-24, the application of the HMBC technique led to the assignment of CH-3, C-4, CH-5 and CH₃-23 by identifying $^2J_{\text{CH}}$ and $^3J_{\text{CH}}$ connectivities. Thus, the carbon atom C-4 (δ_{C} 43.00) is readily assigned as the only quaternary carbon of the four; C-3 (δ_{C} 79.90) is an oxygenated carbon and C-5 (δ_{C} 55.52) is distinguished by additional 3H-25 (δ_{H} 1.07) cross peak ($^3J_{\text{CH}}$) for the latter carbon, while CH₃-23 (δ_{C} 23.10) was identified as a methyl carbon. Working along the molecule in this fashion, using the hydrogens from only the six methyl resonances, allowed in addition unambiguous assignment of CH₂-1 (δ_{C} 44.32), CH₂-7 (δ_{C} 33.20), C-8 (δ_{C} 39.30), CH-9 (δ_{C} 47.60), C-10 (δ_{C} 37.50), C-13 (δ_{C} 138.60), C-14 (δ_{C} 41.80), CH₂-15 (δ_{C} 27.90), CH-18 (δ_{C} 52.85), CH-19 (δ_{C} 39.10), CH-20 (δ_{C} 38.87), CH₂-21 (δ_{C} 30.65) and CH₂-24 (δ_{C} 65.50) (Table 1). The quaternary carbons C-8 (δ_{C} 39.30) and C-14 (δ_{C} 41.80) of **1** were distinguished from the earlier assignment of **3**⁸ (Fig. 1) since the chemical shifts of these carbon atoms are almost invariant for **1** and **3**. Thus, in the HMBC spectrum of **3**, only the carbon C-8 (δ_{C} 39.35) showed connectivities with the hydrogen atoms 2H-11 (δ_{H} 1.90-1.85), while only the



1a R₄=Me R₆=Me R₇=H
2a R₄=H R₆=Me R₇=Me

Figure 2. Stereochemical view of the structures of **1** and **2**.

carbon C-14 (δ_C 41.69) showed a correlation with the hydrogen atom H-12 (δ_H 5.26). On the other hand, the H-18 doublet (δ_H 2.24, $J = 11.4$ Hz) of **1** allowed the localization of H-19 (δ_H 1.32) in the ¹H x ¹H - COSY spectrum which in turn determined the chemical shift of the CH-19 (δ_H 39.35) through ¹H - ¹³C - HMQC Heteronuclear Multiple Quantum Coherence - spin-spin interaction of hydrogen and carbon-13 via one (¹J_{CH}) bond spectrum. Thus, the resonance at δ_C 38.87 correlates to CH-20. Two methine (CH-2 and CH-12), four methylene (CH₂-6, CH₂-11, CH₂-16 and CH₂-22) and one quaternary (C-17) carbons showed no cross peak with methyl hydrogens in the HMBC experiment as expected. Nevertheless, all were unambiguously identified. CH-2 (δ_C 69.28) is readily assigned as an oxygenated carbon and CH-12 (δ_C 125.00) is a sp² carbon. The methylene carbons were identified by use of a COSY

technique. For example, the signal of H-9 (δ_H 1.57) allowed the localization of the 2H-11 (δ_H 1.92) hydrogens which in turn, lead to assignment of CH₂-11 (δ_C 23.30) from the HMQC spectrum. In this way, were also assigned the chemical shifts of the CH₂-16 (δ_C 24.20) and CH₂-22 (δ_C 36.60) carbons. The overcrowded region of the COSY map did not allow the characterization of the 2H-6 hydrogens with respect to 2H-7. By exclusion, the last signal corresponding to one methylene carbon was attributed to CH₂-6 (δ_C 19.20). The remaining quaternary carbon C-17 (δ_C 48.00) was identified by comparative analysis of ¹³C-NMR - PND and ¹³C-NMR - DEPT spectra.

Finally, all methyl ¹³C signals were assigned from their connectivities with assigned methyl ¹H signals. These methyl ¹H signals were identified and distinguished on the basis of the observed connectivities in the HMBC map with

Table 2. NMR data for methyl 2 α ,3 β ,24-tri-O-acetylolean-12-en-28-oate (**2**)*.

C	¹ H x ¹³ C-HMQC (¹ J _{CH})		¹ H x ¹³ C-HMBC		¹ H x ¹ H-NOESY
	δ_C	δ_H	² J _{CH}	³ J _{CH}	
4	43.40	---	H-3, 3-H23, 2H-24		---
8	39.70	---	3H-25		---
10	38.50	---	H-1 β		---
13	143.90	---		H-15 β	---
14	42.20	---	H-15 β		---
17	47.30	---			---
20	30.80	---	3H-29, 3H30		---
28	178.40	---		MeO-28	---
AcO-2	170.80	---	<u>H</u> ₃ CCO ₂ -2	H-2	---
AcO-3	171.00	---	<u>H</u> ₃ CCO ₂ -3	H-3	---
AcO-24	171.10	---	<u>H</u> ₃ CCO ₂ -24	2H-24	---
CH					
2	69.26	5.15 (dt, $J = 12.0, 4.8$ Hz, H-2 β)	H-1 β , H-3		H-1 β , 2H-24, 3H-25
3	79.87	4.82 (d, $J = 12.0$ Hz, H-3 α)		H-1 β , 2H-24	H-1 α , H-5 α , 3H-23
5	55.52	1.08 (H-5 α)		H-1 β , 2H-24, 3H-25	H-3 α
9	47.74	1.58 (H-9 α)		3H-25, 3H-26	3H-27
12	121.84	5.27 (t, $J = 4.0$ Hz)			2H-11, H-18 β
18	41.26	2.86 (dd, $J = 12.0, 3.4$ Hz, H-18 β)			H-12, 3H-30
CH₂					
1	44.10	2.08 (H-1 β) 1.07 (H-1 α)		3H-25	H-2 β , 3H-25 H-3 α
6	19.32	1.64 (H-6 α) 1.47 (H-6 β)			H-5 α 3H-26

Table 2. (cont.)

7	32.90	1.44 (H-7 α)** 1.32 (H-7 β)		3H-26	3H-27 3H-26
11	23.09	1.90 (H-11 α , H-11 β)			H-12, 3H-25, 3H-26
15	27.63	1.62 (H-15 β) 1.07 (H-15 α)		3H-27	3H-26 3H-27
16	23.48	1.97 (H-16 α) 1.62 (H-16 β)	H-15 β		H-16 β , 3H-27 H-16 α
19	45.86	1.64 (H-19 α)** 1.13 (H-19 β)		H-21 β , 3H-29, 3H-30	3H-30
21	33.87	1.33 (H-21 α)** 1.18 (H-21 β)		3H-29, 3H-30	
22	32.30	1.68 (H-22 β) 1.57 (H-22 α)			3H-30
24	65.44	4.20 (s)		H-3, 3H-23	3H-23
CH ₃					
23	23.09	1.02 (s)		H-3, 2H-24	H-3 α
25	16.18	1.05 (s)		H-5	H-1 β , H-2 β , 2H-11
26	16.63	0.72 (s)			H-6 β , H-7 β , 2H-11, H-15 β
27	25.83	1.11 (s)			H-7 α , H-9 α , H-15 α , H-16 α
29	33.14	0.89 (s)		3H-30	
30	23.65	0.92 (s)		3H-29	H-18 β , H-19 β , H-22 β
AcO-2	21.07	1.96 (s)			
AcO-3	20.85/21.07	2.04/2.05 (s)			
AcO-24	20.85/21.07	2.04/2.05 (s)			
MeO-28	51.60	3.61 (s)			

* Multiplicity of signals of carbon atoms deduced by comparative analysis of PND- and DEPT-¹³C-NMR. Chemical shifts of hydrogen atoms obtained from 1D ¹H-NMR. The 2D ¹H x ¹H - COSY and 2D ¹H x ¹³C - HMQC spectra were also used in these assignments.

** The H-7 and H-19/H-21 showed long range coupling (⁴J) with 3H-26 and 3H-30, respectively.

previously assigned carbon atoms (Table 1). In this way, only the signal of 3H-23 (δ_{H} 1.03) showed a cross peak at the ¹³C frequency with the readily assigned CH-3 (δ_{C} 79.90); 3H-25 (δ_{H} 1.07) with CH₂-1 (δ_{C} 44.32); 3H-26 (δ_{H} 0.74) with CH₂-7 (δ_{C} 33.20); 3H-27 (δ_{H} 1.06) with C-13 (δ_{C} 138.60); 3H-29 (δ_{H} 0.84) with CH-18 (δ_{C} 52.85) and 3H-30 (δ_{H} 0.94) with CH₂-21 (δ_{C} 30.65). The assignment of the carbon signals were then carried out by an HMQC experiment (Table 1).

The assignment of CH₂ hydrogens as α or β were obtained by analysis of the splitting patterns (multiplicity and coupling constant) in the 1D NMR, long range coupling⁹ (⁴J) in the ¹H x ¹H-COSY and especially ¹H x ¹H-NOESY spectra (Table 1). These spectra also indicated

that **1** was in conformation **1a** and confirmed the *cis* junction of rings D and E (see **4**).

In a report of 3 β -O-acetyl-27-norurs-13-en-28-oic acid¹⁰, the chemical shifts of the carbon atoms CH₃-29 (δ_{C} 20.02) and CH₃-30 (δ_{C} 17.66) were inconsistent with that of the same carbon atoms in compound **1**. In the case of **1** the ¹H x ¹³C - HMBC spectrum allowed the obvious assignment of CH₃-29 (δ_{C} 16.95) by connectivities of the resonance at δ_{H} 2.24 (H-18) to the resonance at δ_{C} 16.95 (CH₃-29) and of the resonance at δ_{C} 52.85 (CH-18) to the resonance at δ_{H} 0.84 (3H-29). Consequently, the signal at δ_{C} 21.15 was attributed to CH₃-30.

Using the same procedure, the HMBC, COSY, NOESY and HMQC spectra furnished the ¹H and ¹³C chemical

shifts of compound **2** (Fig. 1, Table 2); and also the molecular conformation **2a** and *cis* ring junction as in **4**.

Acknowledgments

The authors gratefully acknowledge Coordenação de Aperfeiçoamento de Pessoal de Nível Superior (CAPES), Brazilian Agency for financial support and Laboratoire de RMN et de Modélisation Moléculaire, ULP, Strasbourg, France, for facilities provided. Special gratitude is extended to R. Graff for technical assistance with the NMR measurements.

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