

AMMONIA LOSSES ESTIMATED BY AN OPEN COLLECTOR FROM UREA APPLIED TO SUGARCANE STRAW⁽¹⁾

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SUMMARY

The quantification of ammonia (NH₃) losses from sugarcane straw fertilized with urea can be performed with collectors that recover the NH₃ in acid-treated absorbers. Thus, the use of an open NH₃ collector with a polytetrafluoroethylene (PTFE)-wrapped absorber is an interesting option since its cost is low, handling easy and microclimatic conditions irrelevant. The aim of this study was to evaluate the efficiency of an open collector for quantifying NH₃-N volatilized from urea applied over the sugarcane straw. The experiment was carried out in a sugarcane field located near Piracicaba, São Paulo, Brazil. The NH₃-N losses were estimated using a semi-open static collector calibrated with ¹⁵N (reference method) and an open collector with an absorber wrapped in PTFE film. Urea was applied to the soil surface in treatments corresponding to rates of 50, 100, 150 and 200 kg ha⁻¹ N. Applying urea-N fertilizer on sugarcane straw resulted in losses NH₃-N up to 24 % of the applied rate. The amount of volatile NH₃-N measured in the open and the semi-open static collector did not differ. The effectiveness of the collection system varied non-linearly, with an average value of 58.4 % for the range of 100 to 200 kg ha⁻¹ of urea-N. The open collector showed significant potential for use; however, further research is needed to verify the suitability of the proposed method.

Index terms: *Saccharum* spp., nitrogen, N fertilization, volatilization, collection system.

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RESUMO: *PERDAS DE AMÔNIA ESTIMADAS POR UM COLETOR ABERTO A PARTIR DE UREIA APLICADA SOBRE PALHADA DE CANA-DE-AÇÚCAR*

A quantificação das perdas de amônia (NH₃) proveniente da aplicação de ureia sobre a palhada de cana-de-açúcar pode ser realizada com uso de coletores, que recuperam o NH₃ em absorvedores acidificados. Desse modo, o uso de coletor aberto com absorvedor envolto em película de politetrafluoroetileno (PTFE) torna-se uma opção interessante, pois apresenta baixo custo, fácil manuseio, além de não interferir nas condições microclimáticas. O objetivo deste estudo foi avaliar a eficiência de um coletor aberto na quantificação do N-NH₃ volatilizado da ureia aplicada sobre solo coberto com restos vegetais de cana-de-açúcar. O experimento foi desenvolvido em área de cana-de-açúcar localizada na região de Piracicaba, Estado de São Paulo, Brasil. As perdas de N-NH₃ foram estimadas pelo coletor semi-aberto estático calibrado com ¹⁵N (método de referência) e coletor aberto com absorvedor envolto em película de PTFE. A ureia foi aplicada sobre a superfície do solo, em tratamentos que corresponderam às doses de 50, 100, 150 e 200 kg ha⁻¹ de N. A aplicação de ureia fertilizante sobre a palhada da cana-de-açúcar resultou em perdas de N-NH₃ de até 24 % da dose aplicada. O coletor aberto não alterou a quantidade de N-NH₃ volatilizada em relação ao coletor semi-aberto estático. A eficiência do sistema coletor variou de forma não linear, com valor médio de 58,4 % na faixa de 100 a 200 kg ha⁻¹ de N-ureia. O coletor aberto apresentou expressivo potencial de uso, porém novas pesquisas são necessárias visando à adequação do método proposto.

Termos de indexação: Saccharum spp., nitrogênio, adubação nitrogenada, volatilização, sistema coletor.

INTRODUCTION

The emission of nitrogen as ammonia (NH₃) into the atmosphere is estimated at about 54 Tg year⁻¹, of which 75 % is caused by human activity (Laegreid et al., 1999). Mineral fertilizers and organic residues applied to the soil represent close to 23 % of global NH₃-N emissions (Bouwman et al., 2002). These N losses do not only reduce the economic efficiency of nitrogen fertilizers, but also harm the environment. Ammonia gas emitted into the atmosphere is generally deposited near the source of emission, promoting the acidification and eutrophication of natural ecosystems (Sommer et al., 2004).

The NH₃-N losses by volatilization from urea, the most widely used source of mineral nitrogen worldwide (Sommer et al., 2004), generally vary from 20 to 40 % of applied N in the case of soil surfaces covered with crop residues of sugarcane (*Saccharum* spp.) (Prammane et al., 1989; Freney et al., 1994; Cantarella et al., 2008). Sugarcane straw favors NH₃ losses by volatilization because the activity of the enzyme urease is higher in plants and plant residues than in soil. Rochette et al. (2009) found that ureolytic activity and the resulting NH₃-N volatilization after the surface application of urea was higher in soil with crop residue on the surface in comparison to conventionally tilled soil. The incorporation of urea into the soil under sugarcane straw can reduce the volatilization, but

the operational efficiency of this method is low and its cost high.

The methods for quantifying NH₃-N volatilized from soil can be classified as direct and indirect. Indirect estimates can be obtained by isotope balancing with ¹⁵N (Lara Cabezas & Trivelin, 1990). However, the high cost of ¹⁵N hinders its use under field conditions. ¹⁵N is also used to calibrate direct methods, which are based on capturing NH₃ volatilized in acid medium (Lara Cabezas et al., 1999; Araújo et al., 2009). These methods can be classified as closed static collector (Sommer & Olesen, 1991), closed dynamic collector (Kissel et al., 1977) and semi-open static collector (Nömmik, 1973). Currently, micrometeorological methods are also used; these are considered to be more accurate as they do not change the environmental conditions close to the soil surface (Sommer et al., 2004). However, these methods require extensive experimental areas (larger than a hectare) that usually need to be flat, as well as sophisticated equipment (such as micro anemometers and gas detectors), making them costly and unviable for field studies with multiple treatments.

The semi-open static collector (SOC), though often used in field studies, has limitations in quantifying volatilization under natural conditions due to changes in the atmospheric environment above the soil surface (such as temperature, relative humidity, rate of water evaporation and NH₃ vapor

pressure) caused by the presence of the collection chamber (Lara Cabezas & Trivelin, 1990; Lara Cabezas et al., 1999). An alternative to this method is the use of collection systems that favor airflow in the area that captures volatilized NH_3 . Therefore, an open collector (OC) with an absorber wrapped in polytetrafluoroethylene (PTFE) film has been proposed; this option seems interesting because of its low cost, easy handling and great potential for use under field conditions (Campana, 2008; Alves et al., 2011). OC consists of a rigid polyvinyl chloride (PVC) plate and foam saturated in an acid solution that absorbs NH_3 , and both are wrapped in PTFE film. The PTFE film is permeable to NH_3 gas (Sørensen & Jensen, 1991). In this context, the use of a SOC calibrated using ^{15}N as performed by Lara Cabezas et al. (1999) enables the evaluation of alternative systems to capture NH_3 volatilized from soil under field conditions.

Therefore, the purpose of this study was to evaluate the effectiveness of NH_3 capture by an open collector with an absorber wrapped in PTFE film using a semi-open static collector to quantify $\text{NH}_3\text{-N}$ volatilized from urea applied to sugarcane residue in an area of commercial production.

MATERIALS AND METHODS

The experiment was performed from September 28 to October 28, 2009, on a commercial sugarcane plantation of the Iracema Mill (22° 35' S; 47° 37' W; 619 m asl) near Piracicaba, São Paulo, Brazil. The local climate is tropical and, according to the Köppen climate classification, mesothermal, with a dry winter (Cwa) transitioning to a humid winter (Cfa). The mean annual rainfall is 1,275 mm, and the average temperature between 24.6 °C (in February) and 17.4 °C (in July). The soil is classified as Typic Hapludox (Soil Survey Staff, 2010). The following characteristics were determined in the 0–20 cm soil layer: pH 4.6 in 0.01 mol L⁻¹ CaCl_2 (1: 2.5 soil: solution - m/v); 95.3 mmol_c dm⁻³ cation exchange capacity (CEC) (Raij et al., 2001); 23.8 g kg⁻¹ total organic C and 1.4 g kg⁻¹ total N, both determined by a mass spectrometer with an automated C and N analyzer (ANCA-GSL model, 20–20, Sercon, Crewe, UK) (Barrrie & Prosser, 1996); and 630 g kg⁻¹ clay, determined by the densimeter method (Gee & Bauder, 1986).

Sugarcane (cultivar SP89-1115) was planted on March 8, 2008, in rows spaced 1.5 m apart. The cane was mechanically harvested without burning cane straw, on June 15, 2009, and the crop residues (tips, leaves and pieces of stalk) were left on the soil surface. The dry weight of straw left on the field ($17.2 \pm 3.1 \text{ Mg ha}^{-1}$) was determined by sampling 1 m² areas (20 replications). The experiment was set

up during the first ratoon cycle and began 105 days after harvesting.

The experiment had a randomized block design and the treatments consisted of four N rates (50; 100; 150 and 200 kg ha⁻¹ N) with four replications. Urea was manually applied over the sugarcane straw (granules containing 46 % N) in a line 20 cm away from the sugarcane row on only one side of each row. Both collector types, SOC and OCs, were installed in each experimental plot (one unit of each collector per replication) to measure volatilized $\text{NH}_3\text{-N}$. In each block, one control was included to evaluate NH_3 losses from the mineralization of straw-covered soil. The NH_3 losses from these plots were subtracted from those treated with urea.

The SOC (Figure 1a) was proposed by Nömmik (1973) and adapted by Lara Cabezas & Trivelin (1990). Cylindrical chambers were fixed on top of PVC bases that were partially inserted into the soil, maintaining the sugarcane straw within. Each SOC has three PVC bases, and the collectors were moved between these bases every time the foam disc was replaced and/or when rain fell during the sampling interval, according to the method described by Cantarella et al. (2003) and Costa et al. (2003). This procedure reduces the microclimate effect within the collector and exposes the fertilizer within the bases to rain and wind (Cantarella et al., 2008). Therefore, the urea was exposed to the same climatic conditions as in the remaining experimental plots.

The OC, modified according to Alves et al. (2011), consisted of a 16 x 16 x 0.2 cm rigid PVC plate (Figure 1b). An NH_3 absorber consisting of polyethylene foam, identical to that used in

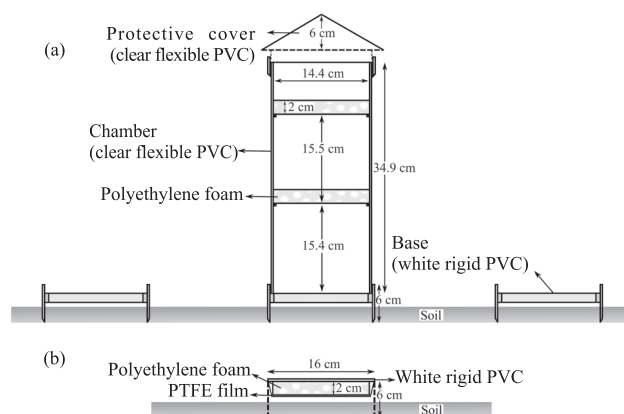


Figure 1. A semi-open static collector and its respective PVC bases (a) and an open collector with an absorber wrapped in PTFE film (b) used to quantify $\text{NH}_3\text{-N}$ volatilized from urea applied over sugarcane straw. Adapted from Lara Cabezas & Trivelin (1990) and Alves et al. (2011).

the SOC (diameter 14.4 cm, thickness 2 cm and density 0.02 g cm^{-3}), was fixed to the underside of the plate. The absorbing foam and PVC plate were wrapped in PTFE film. The PTFE kept the PVC plate and absorbing foam in place, and is highly NH_3 permeable as well (Sørensen & Jensen, 1991). The function of the PVC plate on top of the OC is to prevent NH_3 absorption from soil and atmosphere. The collectors were supported by four metal rods horizontally at 1 cm from the soil surface. Unlike the SOC, which rotated periodically between the bases, the OCs remained in the same place throughout the experiment.

The urea contained within the SOC bases and under the OC bases, taking each treatment into account, was weighed and applied to the surface over the sugarcane straw. The foam absorbers used for both collector systems were previously soaked in approximately 30 mL of orthophosphoric acid (1.5 mol L^{-1} solution with 5 % glycerol - v/v), packed in plastic bags and stored in a freezer ($-15 \text{ }^\circ\text{C}$) until they were used in the experiment to avoid contamination with NH_3 from air.

The foam absorbers were collected and replaced 2, 4, 6, 8, 11, 14, 18, 22, 26 and 30 days after treatment application. To determine the NH_3 retained in the absorbers, NH_3 was extracted by washing the foam with 450 mL of deionized water. The NH_4^+ -N from the samples was determined by a flow injection analysis system (FIA) through the formation of NH_3 in alkaline medium in a permeable hydrophobic membrane (consisting of PTFE) from where the flow was directed to a conductivity cell (Reis et al., 1997). The results of volatilized N from the SOC (NVdSOC, kg ha^{-1}) were corrected using the calibration equations proposed by Lara Cabezas et al. (1999), whereas the OC (NVaOC, kg ha^{-1}) results represented the NH_3 -N absorbed and extrapolated to one hectare, considering the $6,667 \text{ m ha}^{-1}$ application range (0.1 m strip width).

The efficiency (E) of an OC to quantify NH_3 -N volatilization from soil in relation to N volatilization from a SOC was defined by the following equation (Equation 1):

$$E(\%) = \frac{(NVaCA)}{(NVdCSA)} \cdot 100 \quad (1)$$

where $E(\%)$ = efficiency of an OC in the quantification of NH_3 -N volatilization; $NVaCA$ = amount of volatilized NH_3 -N absorbed by the OC (kg ha^{-1}); $NVdSOC$ = amount of volatilized NH_3 -N determined by the SOC (kg ha^{-1}).

The soil moisture content was measured simultaneously with the collection and replacement of the foam absorbers by collecting four undisturbed

topsoil (depth 0–20 cm) samples from the control plot. The water content was determined after oven-drying the samples at $105 \text{ }^\circ\text{C}$ for 48 h. The air temperature was monitored 2 m above the soil surface by an automated weather station installed close to the experimental area, and the data were stored in a data logger (CR200 model, Campbell Scientific®, Logan, USA).

The NH_3 -N losses estimated using the collection systems were subjected to variance analysis and the means compared by Tukey's test at 0.05. The best model was adjusted to correlate the volatilized NH_3 -N absorbed by the collection systems.

RESULTS AND DISCUSSION

During the experimental period, the average air temperature varied between 16.2 and $26.8 \text{ }^\circ\text{C}$, whereas the soil moisture content in the 0–20 cm layer had a small range from 143 to $193 \text{ g kg}^{-1} \text{ H}_2\text{O}$ (Figure 2). Rainy weather, which was atypical for the time of year, prevailed in the region of Piracicaba, and heavy rain occurred at the experimental site, especially after the first 15 days. Rainfall, with a total of 142 mm in the 30 days of the evaluation, is very important for NH_3 -N volatilization because it dissolves and hydrolyzes the soil-applied urea (Prasertsak et al., 2002) and maintains a higher moisture content in the residues and soil surface, thus increasing the NH_3 -N losses (Freney et al., 1992). Other environmental factors that directly influence volatilization are soil temperature and pH (Sommer & Olesen, 1991), relative humidity (Cabrera et al., 2010), wind speed (Sommer et al., 1991), and the concentration of N fertilizer in the soil (Cabrera et al., 1991).

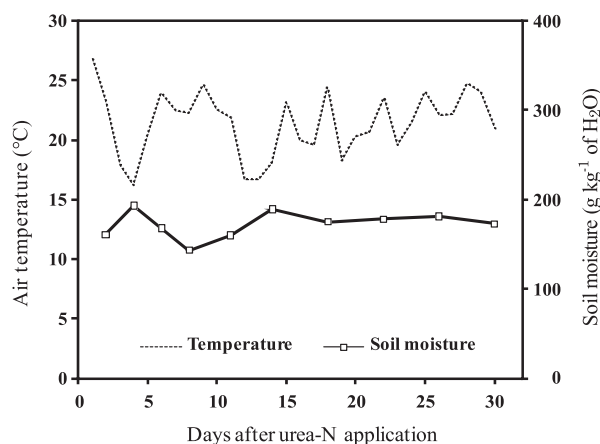


Figure 2. Mean air temperature and soil moisture content in the surface layer (0–20 cm) during the experiment performed in Piracicaba, São Paulo, Brazil.

The rainfall pattern appeared to be the determining component of NH_3 volatilization. A few hours after urea application and installation of the collection systems, there was a 23 mm rainfall (Figure 3), which was insufficient to incorporate the fertilizer into the soil. Due to the presence of plant residue on the surface, more water was required to reduce the urea-N losses by volatilization (Haysom et al., 1990), because the water naturally flows in the channels formed in the crop residues, thus reducing the capacity for dissolution and incorporation of urea into the soil (Freney et al., 1994; Cabrera et al., 2005; Rochette et al., 2009). Calcino & Burgess (1995) found that a rainfall of 23 mm was not enough to dissolve all urea and incorporate it into soil covered with sugarcane residue.

Therefore, the rainfall that occurred on the first day after N fertilization resulted in the dissolution and hydrolysis of urea by urease, an enzyme produced by microorganisms and found intracellularly and extracellularly in soil (Klose & Tabatabai, 2000). Urease is more active in soil covered with crop residues than in tilled soil (Rochette et al., 2009). The hydrolysis of the fertilizer generates NH_4^+ -N and NH_3 -N ions, which increases the pH around the fertilizer granules (Sherlock et al., 1987) and consequently increases the losses of urea-N applied on sugarcane straw (Freney et al., 1992; Prasertsak et al., 2002). The 15 mm rainfall on the 15th day after fertilization caused a small increase in NH_3 volatilization, although the potential for losses is reduced during

this stage (Lara Cabezas & Trivelin, 1990). The NH_3 -N losses stabilized after a 30 mm rainfall event, 18 days after the application of N fertilizer, evidently by incorporating the remaining urea and its hydrolysis products into the soil (Prasertsak et al., 2002), which considerably decreased NH_3 -N emissions after day 19 of the treatment.

The NH_3 -N losses by volatilization were highest in the first week after N fertilization (Table 1), as has been reported by other authors (Lara Cabezas & Trivelin, 1990; Costa et al., 2003). The daily NH_3 -N loss rates during the first six days were 3.6, 2.7 and 1.7 times higher by the SOC than by the OC method, respectively, for N rates of 100, 150 and 200 kg ha^{-1} . These results could be associated with the maintenance of the OC apparatus in the same position during the measurements, while the SOC chambers were alternated between the PVC bases during the experiment. Changing the SOC chambers position over the time allowed the contact of N fertilizer granules with rainfall and wind, decreasing the microclimatic effect within the chamber. By the OC method, the contact between N fertilizer granules and rainfall was probably delayed, decreasing urea hydrolysis, when compared with the SOC method, resulting in lower NH_3 -N losses in the first 6 days after fertilization. Campana (2008) also observed a lower NH_3 -N absorption by the OC than by the SOC method in the first 6 days after the application of 100 kg ha^{-1} N as urea to a pasture soil. Campana (2008) explained the lower NH_3 -N recovery by the OC method as being a result of the saturation of the foam surface due to the intense reaction between the acid and NH_3 -N volatilized during the first 6 days.

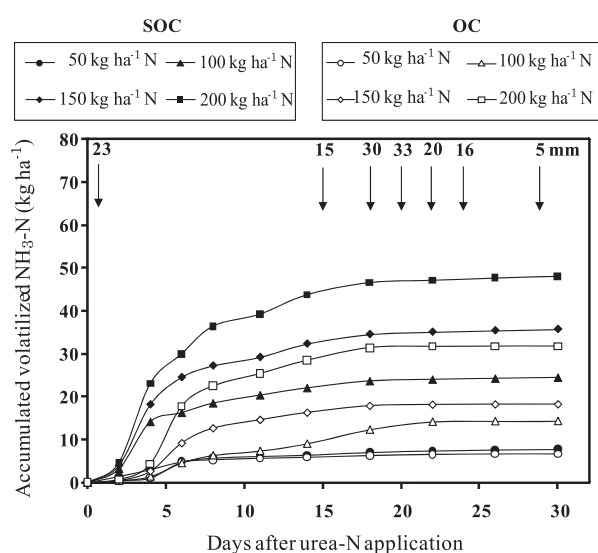


Figure 3. Cumulative volatilized NH_3 -N in 30 days after fertilization, as estimated by two collection methods: a semi-open static collector (SOC) and an open collector with an absorber wrapped in PTFE film (OC). The arrows indicate rainfall events, and the numbers above them indicate the rainfall quantity (mm).

Table 1. Daily rates of NH_3 -N volatilization from urea applied onto sugarcane straw as estimated by two collection systems in different subperiods during the experiment

Collection system	Daily rate of NH_3 -N loss ($\text{kg ha}^{-1} \text{ day}^{-1}$)			
	50	100	150	200
	kg ha^{-1} of urea-N			
	0 to 6 days			
SOC ⁽¹⁾	0.81	2.70 a*	4.09 a	5.00 a
OC ⁽²⁾	0.78	0.75 b	1.53 b	2.96 b
	7 to 14 days			
SOC	0.19	0.72	0.98	1.73
OC	0.15	0.56	0.89	1.34
	15 to 30 days			
SOC	0.09	0.33	0.21	0.27
OC	0.05	0.15	0.12	0.21

⁽¹⁾ SOC: semi-open static collector. ⁽²⁾ OC: open collector with an absorber wrapped in PTFE film. * Means followed by different letters in each column, for each subperiod, differ significantly by Tukey's test ($p < 0.05$).

From day 7 onwards, the $\text{NH}_3\text{-N}$ volatilization rate for all treatments sank to below that of the previous subperiod. The hypotheses for the decline in loss rates beginning between 5 and 10 days after urea application were based on a reduction in the concentration of ammoniacal nitrogen in the soil due to the initial volatilization, the intense fertilizer immobilization/nitrification reactions (Martikainen, 1985; Recous et al., 1990) and the frequent rainfall between 15 and 30 days after nitrogen fertilization, which incorporated the remaining urea-N from the surface into the soil.

The cumulative $\text{NH}_3\text{-N}$ losses measured by the SOC represented 63, 66, 69 and 62 % of the total emissions in the first six days of the study for the urea-N rates of 50, 100, 150 and 200 kg ha^{-1} , respectively (Table 2). In this same period, Lara Cabezas & Trivelin (1990) observed a mean accumulation of 95 % of total volatilized $\text{NH}_3\text{-N}$ using a SOC for 37 days of evaluation in a controlled environment, confirming that the $\text{NH}_3\text{-N}$ losses occur primarily on the first days after fertilizer application. In the case of the OC, the cumulative $\text{NH}_3\text{-N}$ volatilization for N rates of 50, 100, 150 and 200 kg ha^{-1} represented 70, 31, 50 and 56 % of the total $\text{NH}_3\text{-N}$ lost to the atmosphere, respectively. As previously discussed, the maintenance of the OC apparatus in fixed positions on the soil surface probably reduced the recovery of volatilized $\text{NH}_3\text{-N}$, compared with the SOC method, underestimating the cumulative $\text{NH}_3\text{-N}$ losses for N rates of 100, 150 and 200 kg ha^{-1} .

Table 2. Cumulative $\text{NH}_3\text{-N}$ volatilization from urea applied onto sugarcane straw as estimated by two collection systems in different subperiods during the experiment

Collection system	Daily rate of Cumulative $\text{NH}_3\text{-N}$ loss (kg ha^{-1})			
	50	100	150	200
	————— kg ha^{-1} of urea-N —————			
	0 to 6 days			
SOC ⁽¹⁾	4.89	16.22 a*	24.51 a	30.00 a
OC ⁽²⁾	4.68	4.53 b	9.20 b	17.75 b
	7 to 14 days			
SOC	1.54	5.77	7.81	13.8
OC	1.20	4.51	7.15	10.7
	15 to 30 days			
SOC	1.36	2.43	3.41	4.29
OC	0.79	5.29	1.97	3.31

⁽¹⁾ OC: semi-open static collector. ⁽²⁾ OC: open collector with an absorber wrapped in PTFE film. * Means followed by different letters in each column, for each subperiod, differ significantly by Tukey's test ($p < 0.05$).

In the subperiod from 7 to 14 days, based on the entire N rates, the cumulative volatilization for SOC and OC represented, respectively, 24 and 30 % of the total $\text{NH}_3\text{-N}$ released from soil. In the second half of the experiment, the $\text{NH}_3\text{-N}$ losses were lowest (11 % for SOC and 17 % for OC). Lower accumulation of volatilized $\text{NH}_3\text{-N}$ usually occurs 10 days after urea application, depending on the environment and soil conditions (Sommer et al., 2004). Sommer & Ersbøll (1996) used a sigmoid model to relate the cumulative $\text{NH}_3\text{-N}$ volatilization from urea to the time after application, showing that in clay soil, half of the total losses can occur between 2 and 7 days after N fertilizer application.

Despite the underestimation of volatilization for the initial stages (0–6 days), the amount of volatilized $\text{NH}_3\text{-N}$ in the OC system did differ from that in the SOC (Figure 4a); SOCs are considered to provide reliable estimates of the process due to the calibration of the system with ^{15}N . In a greenhouse experiment, Alves et al. (2011) observed no differences between a SOC and an OC in the quantification of $\text{NH}_3\text{-N}$ volatilized from soil. The estimates of mean $\text{NH}_3\text{-N}$ volatilization in relation to urea-N considering all rates (Figure 4b) were 22.0 ± 4.7 and 13.9 ± 7.2 for the SOC and OC, respectively. These N losses were within the estimated range (18–26 %) of total $\text{NH}_3\text{-N}$ emissions from the surface application of urea (Bouwman et al., 2002). The results are also similar to those of studies that evaluated $\text{NH}_3\text{-N}$ volatilization from urea applied onto sugarcane residues (Prammanee et al., 1989; Freney et al., 1994; Cantarella et al., 2008).

The best correlation between the amount of volatilized $\text{NH}_3\text{-N}$ absorbed by an OC (NVaOC) and the amount determined by a SOC (NVdSOC) was obtained with an exponential model (Figure 5). The efficiency of the OC varied non-linearly with the amount of NVdSOC. Lara Cabezas & Trivelin (1990), evaluating the efficiency of SOCs, also found that the amounts of volatilized $\text{NH}_3\text{-N}$ fit an exponential model. Araújo et al. (2009), calibrating a semi-open static collector for the quantification of volatilized N, observed that the recovery of $\text{NH}_3\text{-}^{15}\text{N}$ in the collector rose linearly with the increase in the NH_3 volatilization rate from soil.

The mean efficiency of the OC was 58.4 % of the total $\text{NH}_3\text{-N}$ volatilized from soil for the NVdSOC range obtained at rates between 100 and 200 kg ha^{-1} urea-N (Table 3). Lara Cabezas & Trivelin (1990) reported an efficiency of 16.2 % for a SOC, explained by the layout of the foam absorbers in the collector, which hinder air circulation within the chamber and thus decrease water evaporation from the soil in relation to the surrounding area. Araújo et al. (2009), studying a collector in which the interior air flow was free, observed an efficiency of 57 %.

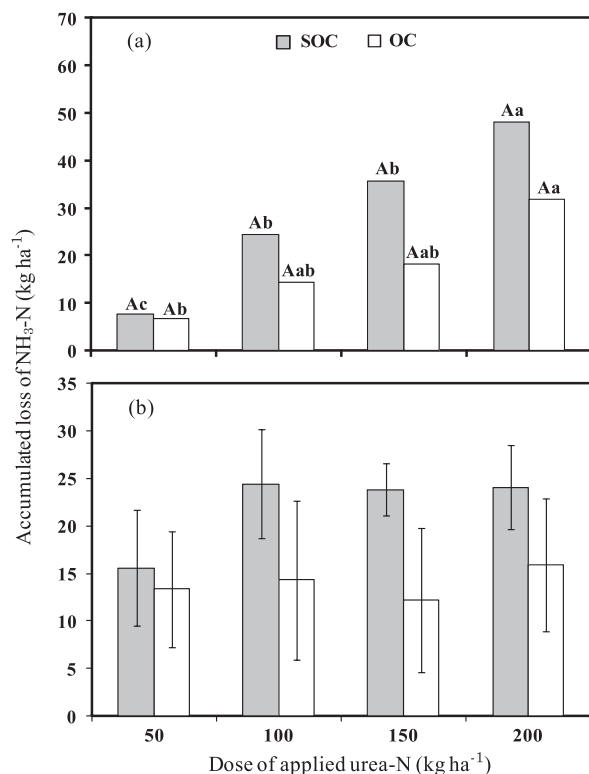


Figure 4. (a) Final NH₃-N loss by surface application of urea on sugarcane straw estimated by a semi-open static collector (SOC) and an open collector with an absorber wrapped in PTFE film (OC). The different upper and lower case letters above the bars indicate significant differences according to Tukey's test ($p < 0.05$) between collector systems and N fertilizer rates, respectively. (b) Percentage of NH₃-N loss. The vertical bars indicate standard deviations from the mean for $n = 4$.

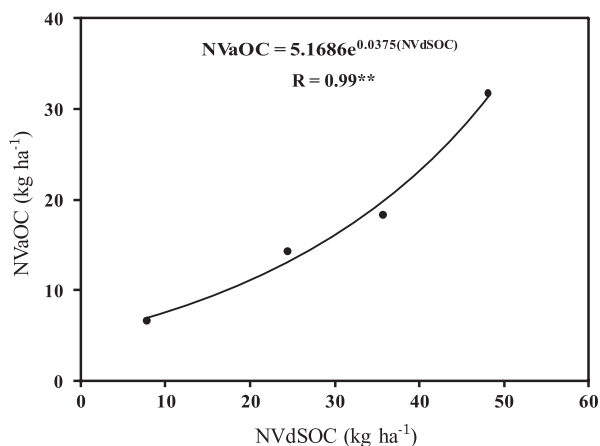


Figure 5. Regression curve between volatilized NH₃-N absorbed by an open collector with an absorber wrapped in PTFE film (NVaOC) and volatilized NH₃-N as determined by a semi-open static collector (NVdSOC). **: Significant at 1%.

Table 3. An open collector with an absorber wrapped in PTFE film efficiency related to urea dose and volatilized NH₃-N as determined by a semi-open static collector (NVdSOC) according to the selected best fit model

N rate	NVdSOC	Efficiency
— kg ha ⁻¹ N —		%
50	7.79	96.5a*
100	24.43	53.2b
150	35.73	55.5b
200	48.09	66.6b

* Means followed by different letters differ significantly by Tukey's test ($p < 0.05$).

Therefore, it is evident that collection systems that favor airflow in the area of NH₃-N capture are more efficient to quantify volatilization.

Since the efficiency of the OC method was related to the use of fixed chamber positions in the field, the rainfall did not reach the N fertilizer applied under the collector apparatus, preventing urea hydrolysis and decreasing volatilized NH₃-N, as confirmed in the first 6-d period. On the other hand, overestimations of NH₃-N volatilization could also occur when heavy rainfall does not reach the N fertilizer applied under the OC, increasing the NH₃-N losses in comparison with places in the surroundings of the collector, where rainfall can incorporate the N fertilizer into the soil. Thus, for reliable estimations of NH₃-N losses using the OC method it is highly recommendable that the collector position should be shifted after rainfall events and foam replacements periods, similarly as adopted for the SOC method.

CONCLUSIONS

1. Applying urea-N fertilizer over sugarcane straw results in losses of NH₃-N of up to 24 % of the applied rate.
2. Results of NH₃-N volatilization measured by an absorber wrapped in PTFE film were not significantly different from those of a semi-open static collector.
3. The efficiency of the collection system varied non-linearly depending on the rate of NH₃-N volatilized, with a mean value of 58.4 % for the range of 100 to 200 kg ha⁻¹ of urea-N.
4. To ensure the reliability of estimations of NH₃-N losses by the open collector method it is

suggested to change the collector position after events of rainfall and absorber replacement.

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