

Problems with reliability and safety of hot dip galvanized steel structures

(Problemas com a confiabilidade e segurança de estruturas de aço galvanizadas a quente)

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Abstract

Hot dip galvanizing is very effective means of protection against corrosion. Some recommendation concerning the steel quality are generally known and accepted. The process consists of cleaning (pickling or sand blasting) and dipping the structures or pieces into liquid zinc bath. The case study of hot dip galvanized steels is presented. Some recent failures of hot dip galvanized welded structures and hot dip galvanized high strength steel screws are presented. Structures were made of S355 grade steel and MIG/MAG process was applied for welding. Large cracks were observed in the vicinity of welds after hot dip galvanizing process. The presence of both hydrogen and liquid metal embrittlement was identified and associated mainly with higher hardness of HAZ or the quenched and tempered steels. Possible cracking mechanisms are discussed. The influence of chemical composition and production process (welding, heat treatment) was analyzed according to data published in literature. The solutions and recommendations for avoiding the failure in hot dip galvanized structures are proposed.

Keywords: steel, welding, cracking, hot dip galvanizing

Resumo: Galvanização a quente é um meio muito efetivo de proteção contra a corrosão. Recomendações relativas a qualidade do aço são geralmente conhecidas e aceitas. O processo consiste de limpar (decapagem ou jateamento) e mergulhar as estruturas ou partes destas em um banho de zinco líquido. O presente trabalho apresenta casos de falhas recentes em estruturas soldadas e em parafusos de aços de alta resistência galvanizados a quente. As estruturas foram fabricadas com aço do grau S355 e o processo MIG/MAG foi usado para a soldagem. Os parafusos foram fabricados com aço de alto carbono. Grandes trincas foram observadas nas proximidades das soldas após o processo de galvanização a quente. A ocorrência de fragilização tanto por hidrogênio como por metal líquido foi identificada e associada com a dureza elevada tanto da ZTA como dos aços temperados e revenidos. Os possíveis mecanismos de fissuração são discutidos. A influência da composição química e processo de fabricação (soldagem e tratamento térmico) é analisada de acordo com dados publicados na literatura. Soluções e recomendações para evitar a falha em estruturas galvanizadas a quente são propostas.

Palavras-Chave: Aço, Soldagem, fissuração, galvanização a quente

1. Introduction

Hot dip galvanizing is very important production process used for steel protection against corrosion. This process seems to be even more economical and effective compared to protection by painting. Hot dip galvanizing is used for protection of steel products for more than 100 years. During application of this process the recommendations have been developed for steel specification. These recommendations should assure the technological reliability and also the quality of zinc layer. Recommendations for constructions to be galvanized (such as

ventilation etc.) are the basic requirement for the continuous galvanized process. The quality of hot dip galvanizing also influence not only the technological conditions (bath chemical composition, cleaning process etc.) but also chemical composition of the steels.

The quality requirements and the procedures for quality assessment of hot dip galvanized process are specified in EN ISO 1461 [1] and the construction requirements are given in EN ISO 14713 [2].

The influence of Si and P in the steel to be galvanized on the thickness and quality of zinc layer is generally known. This requirement for the content of both Si a P is specified in the standards of structural steels EN 10025 - 1 to 6 [3] for the normalized and also for the quenched and tempered but also for TMCP steels.

The intermetallic phases containing different Fe and Zn content are formed on steel surface during hot dip galvanizing.

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The stresses and also the effect of strain level on the quality of steel products were studied. It is generally known that there is no remarkable problem with galvanizing mild steels.

Cracking of steels has been observed in steels with the yield strength over 1000 MPa [4]. Specific problem was observed with high strength steel fasteners [4]. Fracture was not observed for the fasteners without zinc layer. Some failures in welded structures of hot dip galvanized steel are reported in [6-9]. The main reasons for this failure are indicated the following parameters such as high residual stresses, local plastic strain and local hardening.

The subject of this contribution is to attain an overview on possible fracture mechanisms which were observed during hot dip galvanizing of steel constructions.

2. Failures of hot dip galvanized steel products

There were three basic embrittlement mechanisms observed according to [4] when the hot dip galvanized steel structure failed. It is:

- liquid metal embrittlement
- hydrogen embrittlement and
- strain ageing embrittlement.

McDonald [5] supposed also the influence of Fe - Zn intermetallic layer on the fracture. The role of all mechanisms is complex and one of them may miss in the actual case.

The liquid metal embrittlement caused by zinc and cadmium was studied by Arrata at el [6] in liquid Cd, Zn and Cd + 5 % Zn, Cd + 50 % Zn alloys. The authors identified that liquid metal influences the plasticity of steels. The ductility decreases with increasing Zn content (Figure 1). The change of plasticity was associated with the actual fracture mechanism from transcrystalline to intercrystalline. They also observed that the fracture associated with the liquid metal embrittlement takes place when the minimum stress level and or deformation is applied (Figure 2).

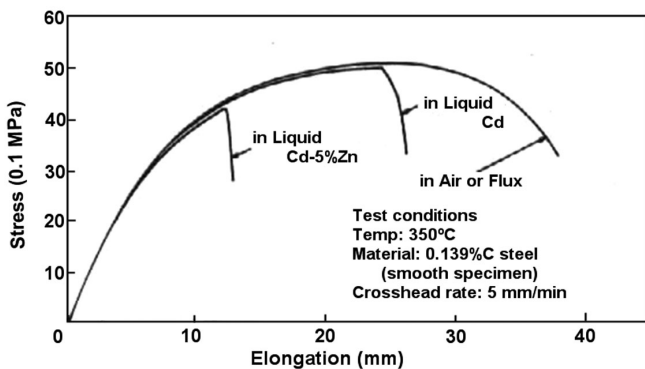


Figure 1. Influence of Cd and Zn on the steel elongation [6].

Pavlidis et al. [7] found out, that the pure zinc does not change the ductility of steels within the temperature range from 450 to 500°C. The liquid metal embrittlement is remarkably influenced by the presence of Pb in the liquid metal.

Friehe a Hankel [8] also identified that iron is more susceptible to intercrystalline liquid metal embrittlement at

475°C than the killed and also not killed steel.

Steel welded structures are hot dip galvanized very often. Mainly mild steel are used for these structures. Liquid metal embrittlement caused by zinc has also been observed in welded steel structures galvanized after welding. This problem, related with steel bridge structures was studied by Abe H. et al. [9].

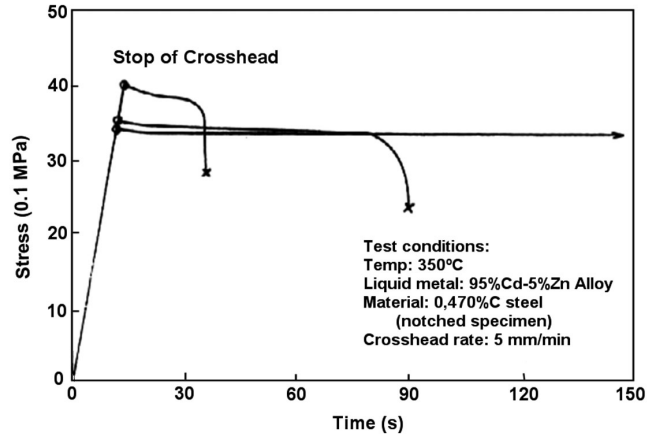


Figure 2. Influence of load on liquid metal embrittlement [6].

The influence of chemical composition was identified for hot dip galvanized steel failure. The following equation was developed by the regression analysis [10]:

$$S_{LM-400} = 201 - 370C - 22Si - 51Mn - 35P + 33S - 28Cu - 22Ni - 87Cr - 123Mo - 275V - 182Nb - 82Ti - 24Al - 1700N - 155000B \quad (1)$$

$$S_{LM-400} = 227 - 320C - 10Si - 76Mn - 50Cu - 30Ni - 92Cr - 88Mo - 220V - 200Nb + 200Ti \quad (2)$$

where S_{LM-400} is the parameter which identifies the influence of liquid metal at 470°C on the notched strength - $R_{m \text{ in zinc}} / R_{m \text{ without zinc}} \times 100\%$, and also zinc carbon equivalent (CEZ):

$$CEZ = C + Si/17 + Mn/7.5 + Cu/13 + Ni/17 + Cr/4.5 + Mo/3 + V/1.5 + Nb/2 + Ti/4.5 + 420B \quad (3)$$

Japanese authors also supposed that liquid zinc embrittlement is caused by diffusion of zinc along the primary austenitic grain boundaries due to thermal and residual stresses. They also concluded that embrittlement can be reduced by the presence of ferrite on the HAZ grain boundaries.

There are no serious problems when the steels of yield strength in the range from 235 to 460 MPa are galvanized according to [12]. It is also recommended that the killed steels resistant to strain ageing or fine grained steel grades should be used for steel structures containing bent parts. For the steels with yield strength higher than 460 MPa it is recommended to use shorter pickling time in order to avoid the hydrogen embrittlement. The influence of elements such as Sn, Pb and also Bi on liquid metal embrittlement is reported. The content of

both Sn and Pb should not be higher than 1.3 % and the content of Bi should be 1.0 % max.

The cracking phenomenon is expected according to [13] when hardness would exceed the level of 34 HRC or 340 HV due to hydrogen. Unfortunately, these criteria are not mentioned in the standard EN 14713.

Hydrogen embrittlement during hot dip galvanizing process is discussed also in [14], [15]. The guideline [14] considers the hydrogen embrittlement as one of the fracture mechanism of galvanized structures. The steel production process, welding and pickling are possible sources of diffusible hydrogen. No criteria such as maximum hardness can be found in this document. The EN 1011-2 standard is recommended to follow, in order to avoid the hydrogen induced cracking. Usually inhibitors are added to liquid metal for avoiding this problem. Anyway, liquid metal embrittlement is expected when steels of the yield strength higher than 355 MPa are used for welded structures.

According to the literature related to cracking of welded structures after hot dip galvanizing, it is supposed that the liquid metal embrittlement is associated with local strain or stress seems to be the main mechanism for the failure. The influence of hydrogen is expected but there are no criteria given in order to reduce the hydrogen induced cracking.

3. Failed steel structures after hot dip galvanizing

Several failed welded structures of zinc coated steels and steel products have been analyzed in the last few years. The welded structures were fabricated from structural steel grade 355 MPa and welded by using MIG/MAG process.

First example of failed steel structure is a weldment in 6 m length fabricated from L profiles in dimension 6 x 80 x 80 mm and 5 x 50 x 50 mm. The cracks sometimes attaining almost 70 mm in length were observed after hot dip galvanizing (Figure 3). These cracks initiated close to the fillet welds and propagated perpendicularly to the profile surface (Figure 4). No cracks were identified by visual testing just after welding.

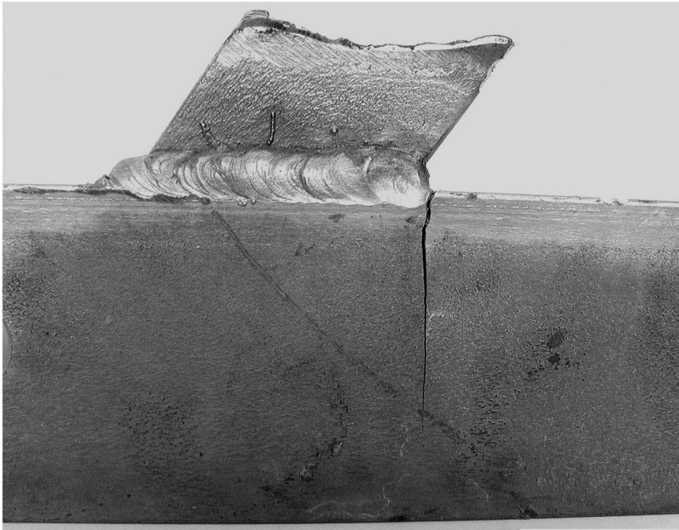


Figure 3. Fracture in a welded steel structure close to L profile weld.

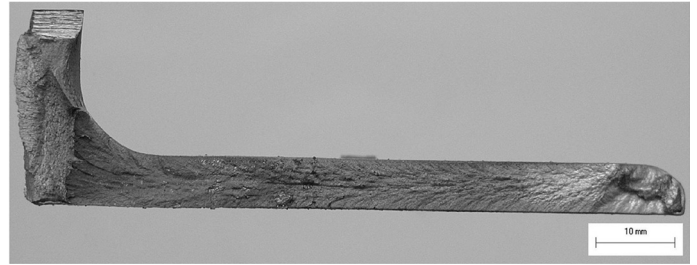


Figure 4. Fracture surface in L – profile.

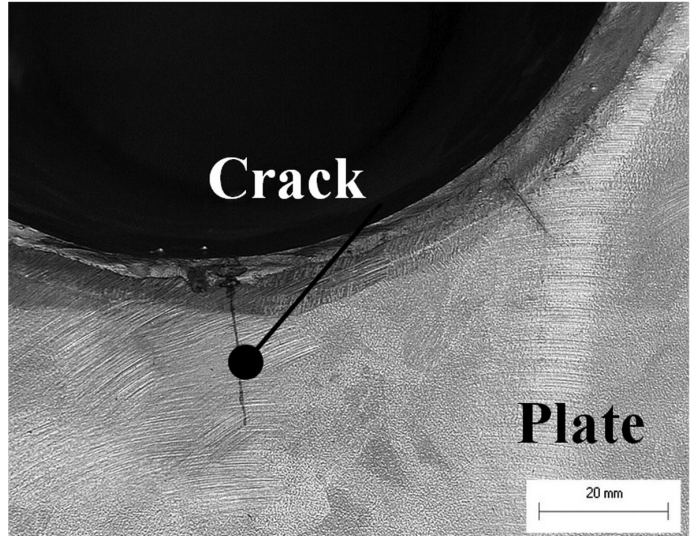


Figure 5. Fracture of circumferential weld between the tube and plate.



Figure 6. Crack morphology in a circumferential weld between the plate and tube.

Second failed steel structure consisted of a tube welded to 10 mm steel plate. The plate 10 mm in thickness was made of S355 steel grade. Several cracks were identified perpendicularly to the circumferential weld (Figure 5). A number of cracks has been identified by using dye penetrant test when the zinc layer was removed by grinding. The crack surface was covered by zinc layer and it propagated both, in plate and tube as well (Figure 6).

Third case of hot dip galvanized steel structure was the fracture of a sizable steel structure containing multi-pass welds. The structure failed in the location of stress concentration (Figure 7).



Figure 7. Crack in zinc coated welded steel structure.

Table 1. Chemical composition of steels used for steel structures.

Parent material	Composition (weight. %)							
	C	Mn	Si	P	S	Al	N	CE*
tube	0.080	-	0.01	0.014	0.006	0.034	0.005	-
plate	0.207	1.41	0.41	0.010	0.020	-	-	0.440
profile 80 x 6 mm	0.181	1.30	0.37	0.019	0.011	0.045	-	0.397
S355J2G3	Max. 0.27	Max. 1.70	Max.0.60	Max.0.055	Max. 0.055	-	-	max.0.55

$$* CE = C + Mn/6 + (Cr+Mo)/5 + (Ni+Cu)/6$$

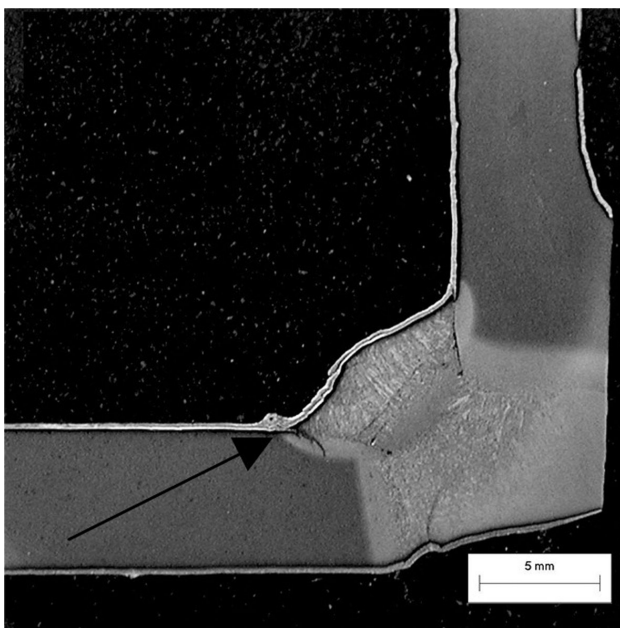


Figure 8. Location of crack initiation (arrow).

3.1 Investigation of the failed welded steel structures

Fracture mechanism of steel structures was investigated using optical microscope METAVAL. The specimens for investigation were cut from the broken welds. These specimens were prepared for microscopical analysis by grinding, polishing and etching. Also Vickers hardness measurements of base metal and HAZ were performed at the load of 49.05 N (HV 5). Chemical composition of steels used for failed steel structures is given in Table 1.

Macrostructure of the L – profile fillet weld is shown in Figure 8. The cracks initiated in the notch at the interface between the weld and base metals, e. g. in the HAZ. The crack length is shown in Figures 9 and 10. The cracks propagated perpendicularly to the L-profile length from the HAZ into base metal. The cracks were trans- and intergranular in the HAZ and intergranular in the base metal. The crack surface is covered by zinc layer (Figure 11).

Macrostructure of the tube to plate circumferential weld is shown in Figure 12. The cracks were oriented perpendicularly to the circumferential weld (Figure 13) and propagated into plate and also into the tube. In some areas the cracks oriented parallel to weld seam were observed. Fine trans- and intergranular microcracks were observed in the HAZ close to the main circumferential crack (Figure 14). The intergranular cracks were observed in the base metal.

The presence of cracks was identified in a large weldment in the vicinity of multi-pass welded joint located in the stress concentration area. Cracks initiating from the weld surface and weld root were identified in one cross-section (Figure 15). The crack initiation area could not be identified. The cracks of branched type were observed with intergranular propagation in the base metal and filled also by zinc layer (Figure 16).

Hardness measurements have shown that the HAZ hardness in the tube to plate weldment attained 274 HV5 and the HAZ hardness in L – profile of structure attained 360 HV5.



Figure 9. Crack initiation and propagation.

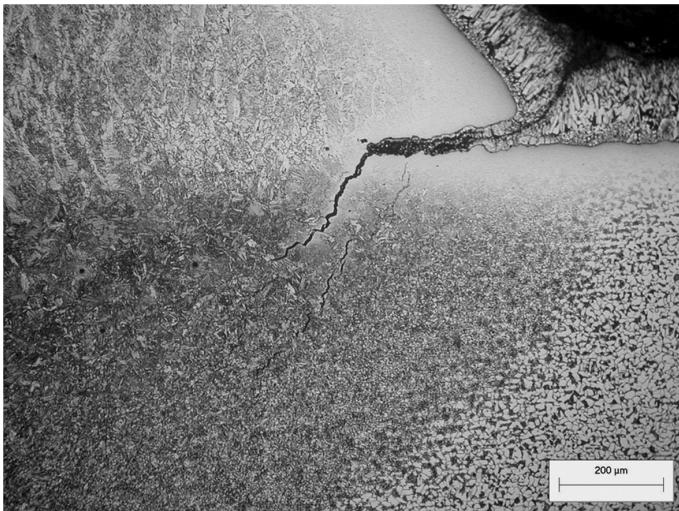


Figure 10. Crack initiation in the HAZ.

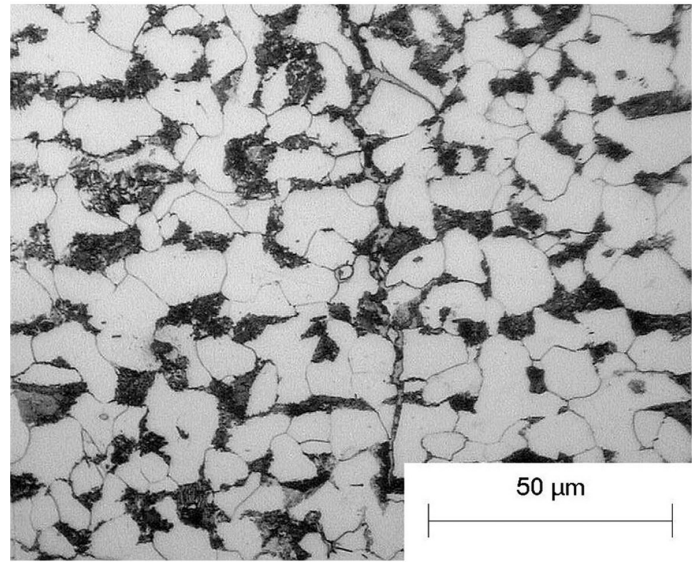


Figure 11. Crack propagation in the base metal.

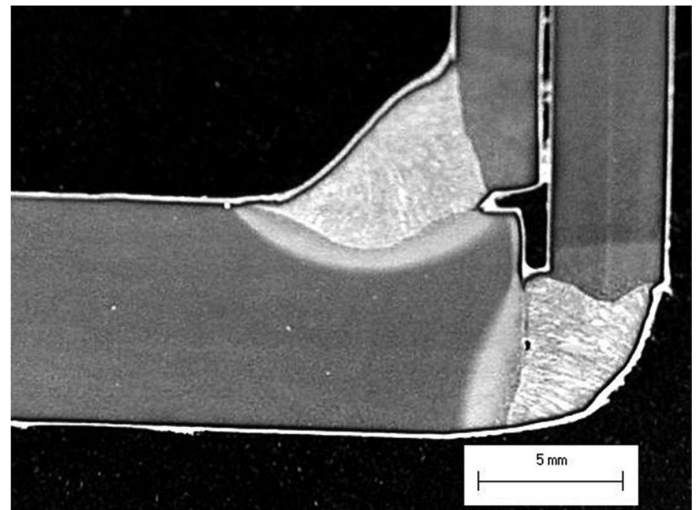


Figure 12. Cross-section of the circumferential weld.

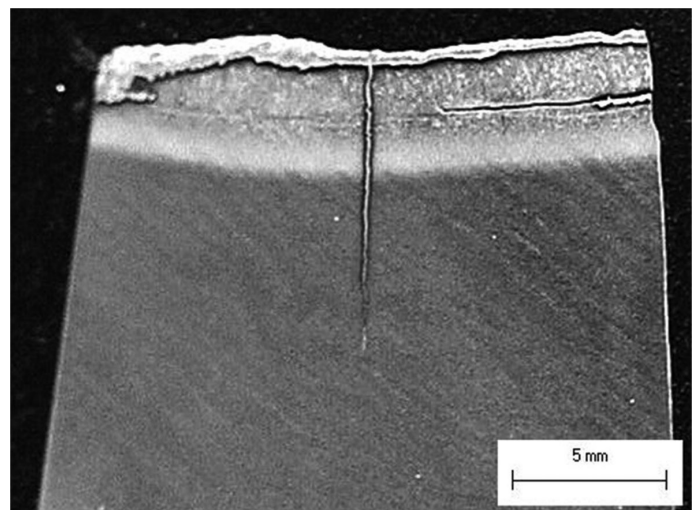


Figure 13. Crack in the circumferential weld.

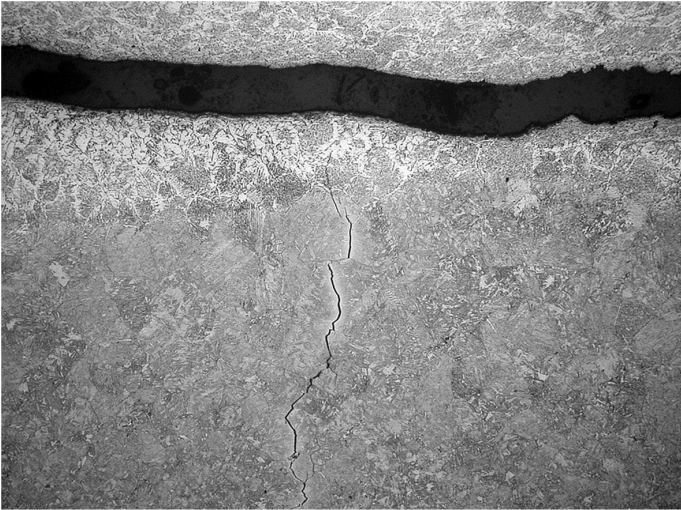


Figure 14. Crack in the circumferential weld.

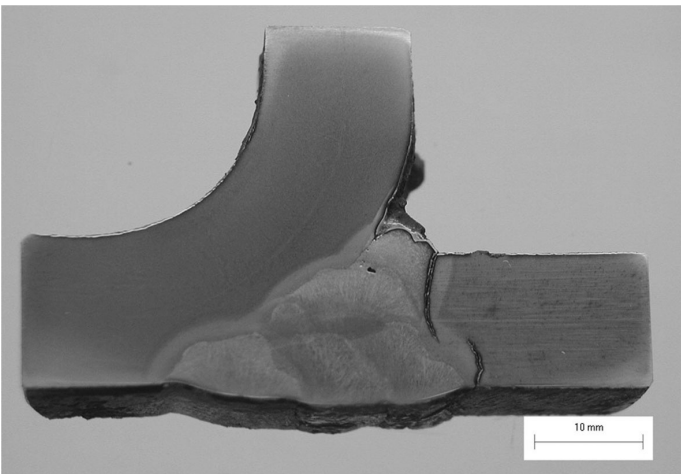


Figure 15. Crack location in the steel welded structure.

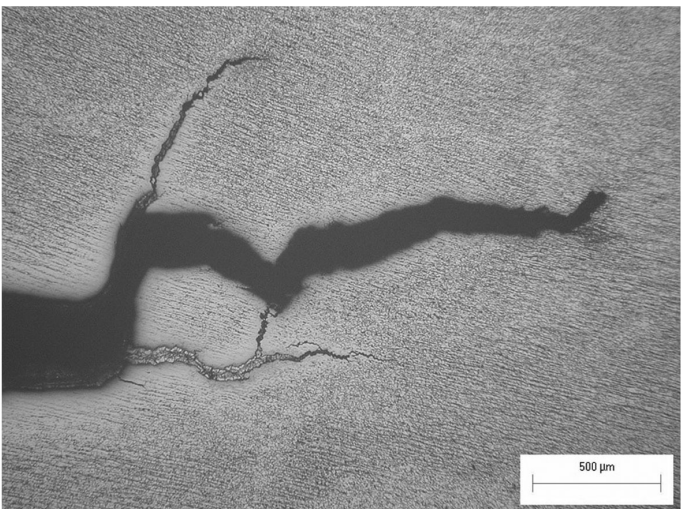


Figure 16. Crack propagation in the base metal.

4. Discussion

The analysis of failed steel structures after hot dip galvanizing has shown some common features. The fracture

initiation has always been observed in the vicinity of welds. Hardness level higher than 270 HV 5 has been measured in coarse grained HAZ. All failed structures were made of S355 steel grade. The crack surface was always covered by a zinc layer. Trans- and intergranular fracture close to the fracture initiation and intergranular fracture in the crack propagation direction were identified by microscopic examination. The presence of transgranular fracture suggested that most probably hydrogen embrittlement of the HAZ region is responsible for crack initiation and liquid metal embrittlement (intergranular fracture) and for crack propagation in the base metal. Nevertheless, the local strain during hot dip galvanizing cannot be excluded as a possible factor responsible for the fracture initiation and propagation.

Failures of welded structures have appeared in zinc coated applications. It seems that coating process unanimously affects the reliability of products. In case of welded structures, failures appeared before putting the structure into service and were associated with higher hardness, diffusible hydrogen, liquid zinc and also local residual stresses as the consequences of welding.

Based on the presented case studies, we assume that the local stresses, high hardness and diffusible hydrogen are the main factors responsible for failures of zinc coated products in general. The presence of active, diffusible, hydrogen may be affected by the steel production, welding process but we assume that in case of steel structures mainly by the pickling. This process is used for cleaning the structures prior to coating.

Anyway, according to literature survey also local strains may increase the risk of failure due to non-uniform heating during hot dip galvanizing. This local strain can increase the risk that LME could also appear as the main fracture mechanism. We assume that in presented cases the LME is active in the crack propagation range but not in the crack initiation range. The failures have shown that there is a need for criteria for hot dip galvanized steel structures welded prior to galvanizing, in order to avoid the cracking occurrence. We suppose that the HAZ hardness limit or choice of suitable steel with lower CE can be helpful to avoid the crack occurrence. Of course such recommendations are highly dependent on other factors such as structure rigidity which is responsible for local strains during hot dip galvanizing. More data are necessary in order to estimate such criteria.

5. Conclusions

Several failures of zinc coated steel products have been presented in this paper. The investigation of the failed steel structures have shown that the main factor responsible for fracture initiation seems to be the hydrogen embrittlement due to presence of high-hardness microstructure and residual or external stresses. The active diffusible hydrogen appears mainly from the coating process – pickling, which is used for cleaning prior to galvanizing. The LME is supposed as the mechanism for crack propagation in welded steel structures.

6. Acknowledgement

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