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Stocks and Distribution of Soil Carbon, Nitrogen, Phosphorus and Sulfur in an Integrated Crop-Livestock System Treated with Phosphates

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HIGHLIGHTS

- Conservation agriculture practices under Typic Dystrudept increase stocks of C, N, P and S over time of five years.
- The annual applications of soluble phosphates to soil surface changes concentrations of phosphate and sulfate in the soil under no-tillage, in an integrated crop-livestock system.
- In situation of phosphorus-fixing soils, anticipated fertilization of sources and doses phosphates applied in the soil surface supply this nutrient for forage and grain crop production systems over

Abstract: Conservation agriculture practices can contribute to changes in soil nutrient dynamics over time. This experiment evaluated the changes in total stocks and distribution of carbon, nitrogen, phosphorus and sulfur concentrations in soil, during 60 months, in an integrated crop-livestock system (ICLS) due to anticipated fertilization of sources and doses phosphates applied in soil surface. The experiment was conducted over a period of five years, under Typic Dystrudept, using a randomized block design, in an incomplete factorial scheme (3×3+1), with four replications. Treatments consisted of three sources of P [triple superphosphate (TSP), rock phosphate – Arad (RP) and magnesium thermophosphate (MTP)], along with four doses of P (0, 60, 120 and 180 kg ha⁻¹ P₂O₅ total). Samples of soil were collected in 0-5, 5-10, 10-15, 15-20 and 20-30 cm layers at 24, 36, 48 and 60 months after beginning of experiment where the following chemical attributes were evaluated: (i) total organic carbon (TOC); (ii) total nitrogen Kjeldahl (TNK); (iii) available P by ion exchange resin method (P-IER); and (iv) available S-SO₄²⁻. The ICLS conditions provided increased total stocks and concentrations of TOC, TNK, P-IER and S-SO₄²⁻ over time. The applications of different phosphates had no influence on soil TOC concentrations during the five years of experimentation.

The concentrations of TNK, P-IER and S-SO₄²⁻ showed an increase in different layers of soil, with the application of sources and doses of P. The P fertilization practice that was anticipated can consist of an efficient management of soil fertility, using properly managed conservation systems.

Keywords: anticipated fertilization; soluble and insoluble phosphates in water; sustainable intensification; nutrient cycling; conservation agriculture practices.

INTRODUCTION

Conservation agriculture practices can effectively contribute to enrichment of soil fertility, as well as ensuring the sustainability of the agroecosystem. Thus, an integrated crop-livestock system (ICLS) is a practice of soil conservationist management that, owing to its benefits (e.g., deposition of residues in the soil (animals + plants) [1]; high nutrient cycling rate [2]; and the increase in the soil organic matter [3]), promotes biotic and abiotic modifications in the soil-plant-atmosphere system, which alters the biogeochemical process of soil nutrients [1,4].

The ICLS was widely evaluated in relation to carbon (C) gains in the soil [3-8] over several decades. On the other hand, little is studied considering levels of nitrogen (N) and phosphorus (P) [7], and rare are those that consider sulfur (S) [8], in ICLS. In these systems, cattle grazing alters direction, magnitude and composition of nutrient fluxes, affecting residues decomposition and nutrient release rates. Thus, deposition of organic residues in the soil, accumulation of carbon in the soil and maximization of above- and below-ground plant development will promote higher nitrogen, phosphorus and sulphur levels in the soil. Moreover, increase in the soil organic matter resulting from this production system, helps to reduce adsorption of anions on the surfaces of colloids, such as phosphates and sulfates, over time, and can maximize efficient use of phosphorus [9].

Among the most commonly used phosphate sources in Brazil are phosphates water-soluble, citrate soluble and with low solubility in both water and citrate. The P sources that provide higher solubility, such as triple superphosphate (TSP) and magnesium thermophosphate (MTP), present better performance in relation to sources of lower solubility, such as rock phosphate (RP), since the latter phosphate guarantees release of P [10]. In addition, the definition of phosphate fertilizer dose to be added to soil depends on the fertilization practice adopted. Corrective fertilization followed by maintenance and the gradual construction of soil fertility is basically the practices in use in Brazil. However, according to soil content, the dose should be adjusted to recompose export by crop and achieve or maintain optimal soil content [11].

Anticipated fertilization is a practice that provides for total or partial application of recommended doses of fertilizer to the summer crop at the time of sowing of the previous crop, in an incorporated form or at soil surface [12]. Thus the application of soil surface P to winter crops is an interesting practice in production systems such as ICLS, because it would improve P recycling, especially in the medium and/or long-term and ensure a better time and operational efficiency in sowing summer crop [13]. However, scientific information on practice of anticipated fertilization with different sources and doses of phosphates is still scarce, especially in the medium and/or long-term in ICLS.

The aim was to evaluate the changes in total stocks and distribution of carbon, nitrogen, phosphorus and sulfur concentrations in the soil, during 60 months, in an integrated crop-livestock system (ICLS) due to anticipated fertilization of sources and doses phosphates applied in soil surface.

MATERIAL AND METHODS

Experimental site, design and management

The experiment was conducted over a period of five years (April 2009 to April 2014), in the municipality of Castro - PR (24°51'49"S, 49°56'61"W, 1020 m asl), with the predominant climate being of Cfb type, according to the Köppen classification, with a temperature of 16°C, and an average annual precipitation of 1087 mm. The area was cultivated eight years ago under the no-tillage system. The soil is Typic Dystrudept [14] (605, 225 and 170 g kg⁻¹ of clay, silt and sand, respectively). Quartz, kaolinite and gibbsite are the predominant minerals in a clay fraction. The 0-20 cm soil layer at the beginning of the experiment had the following attributes: pH (CaCl₂) 4.8; total acidity (H + Al) 9.2 cmol_c dm⁻³; base saturation 38%; Al³⁺, Ca²⁺, Mg²⁺ and K⁺ of 0.04; 3.1; 2.3 and 0.35 cmol_c dm⁻³, respectively; P available [ion exchange resin (IER)] (P-IER) and

available S-SO₄²⁻ of 18 and 12.8 mg dm⁻³, respectively; total organic carbon (TOC) (Walkley-Black) and total nitrogen Kjeldahl (TNK) [semi-micro-Kjeldahl]; of 29.6 and 2.0 g dm⁻³, respectively.

The experimental design was a randomized complete block design in an incomplete factorial scheme (3×3+1), with four replications. Treatments consisted of different sources [triple superphosphate (TSP), rock phosphate – Arad (RP) and magnesium thermophosphate (MTP)] and doses [absolute control (0 kg ha⁻¹ of total P₂O₅), 60, 120 and 180 kg ha⁻¹ of total P₂O₅] of P. Quantities utilized from each source were calculated based on total P₂O₅ content of fertilizers and applied annually at soil surface at the time of winter forage sowing (anticipated fertilization). Chemical composition of different phosphate sources used is shown in Table 1.

Crops rotation that was followed during the experimental period of 60 months in ICLS were as follows: (i) black oat (*Avena strigosa* Schreb.) (2009, 2011 and 2013); (ii) maize (*Zea mays* L.) (2009/10, 2011/12 and 2013/14); (iii) ryegrass (*Lolium multiflorum* L.) (2010 and 2012) and (iv) soybean (*Glycine max* L.) (2010/11 and 2012/13). Grain crops growth in plots measuring 273 m². Rotational grazing system was adopted, separated into each block with 5525 m² each. Cattle used during the grazing period corresponded to 21 heifers of Dutch dairy cattle, with an average weight of 250 kg, equivalent to 5.2 animal units (U.A.). These remained in each paddock for 4-7 days depending on forage species, also taking into account heights of entrances (20 cm) and exits (10 cm) for cattle.

Table 1. Chemical composition of phosphate sources used during 60 months of evaluation experimental

Sources of phosphates	Chemical composition					
	Total P ₂ O ₅	Water-soluble P ₂ O ₅	Citric acid-soluble P ₂ O ₅	CaO	MgO	SiO ₂
	-----g kg ⁻¹ -----					
Triple superphosphate	460	380	-	130	-	-
Rock phosphate - Arad	330	-	101	370	-	-
Magnesium thermophosphate	180	-	165	180	70	100

The soil acidity control was carried out in 2008. After the beginning of the experiment (April 2009), no liming was applied. However, except for P, all other nutrients were being applied at recommended doses, according to results of soil chemical analysis and crop requirements, during the experimental period. Potassium chloride (KCl) and urea [CO(NH₂)₂] fertilizers were used for the plants as sources of K and N, respectively, and distributed at soil surface. Soybean seeds were inoculated with strains selected from *Bradyrhizobium japonicum* and treated with cobalt and molybdenum at doses commonly used in that region [15]. When necessary, other agronomic practices (e.g., seed treatment, weed control, pests and diseases) were carried out in order to facilitate adequate growth and development of crops.

Soil sampling and chemical determination

In April, at 24, 36, 48 and 60 months after the beginning of the experiment, soil samples were collected at the following depths: 0-5, 5-10, 10-15, 15-20 and 20-30 cm. Samples for analysis were composed of these 12 simple randomized samples to form a composite sample. After sampling, the soil was taken to the laboratory in plastic bags, air dried at 40 °C, ground and sieved in a 2.0 mm mesh sieve for later analysis [16]. Samples were analyzed to determine: (i) total organic carbon (TOC) by the Walkley-Black method and factor-corrected for no-till managed soils [17]; (ii) total nitrogen by the semi-micro-Kjeldahl method (TNK); (iii) available P by the IER method (Amberlite IRA-410 and Amberlite IRA-120) (P-IER) and; (iv) available S-SO₄²⁻ extract by ammonium acetate (0.5 mol L⁻¹) in acetic acid (0.25 mol L⁻¹) solution, and measure by turbidimetric method. Total stocks were estimated to be 0-30 cm based on weighted average of soil layers analyzed.

Statistical analysis

The results of stock and distribution of TOC, TNK, P-IER and S-SO₄²⁻ in the soil were submitted to univariate statistical analysis according to the experimental model in a randomized complete block design. In cases where significant F (p<0.05), Tukey test (α = 0.05) was applied. Equation (1) shows the statistical model used for analysis of variance (ANOVA) for each soil layer evaluated (for stocks, 0-30 cm and, for

distribution in the soil, 0-5, 5-10, 10-15, 15-20 e 20-30 cm). All statistical analyses were performed using SAS Version 9.2 program (SAS Institute Inc. 9.1.2).

$$Y_{ijkl} = \mu + B_i + S_j + D_k + P_l + (SD)_{jk} + (SP)_{jl} + (DP)_{kl} + (SDP)_{jkl} + \varepsilon_{ijkl} \quad (1)$$

Where: μ = overall mean of experiment; B = blocks ($i = 1, 2, 3, 4$); S = phosphate sources ($j = 1, 2, 3$); D = total P_2O_5 doses ($k = 1, 2, 3, 4$); P = experimentation periods ($l = 1, 2, 3, 4$) and; ε = experimental error.

RESULTS

Stocks of soil carbon, nitrogen, phosphorus and sulfur in integrated crop-livestock system

The total stocks of TOC, TNK, P-IER and S-SO₄²⁻ were strongly influenced by the experimentation period (Table 2). The greater deposition of residues (vegetable + animal) in ICLS may have led to an increase in stocks over time. There was a progressive annual addition in TOC stock (on average, 4.3 Mg ha⁻¹ year⁻¹) in the 0-30 cm layer (Figure 1A). For TNK, a small annual increase (approximately 0.81 Mg ha⁻¹ year⁻¹) was observed (Figure 1B). However, the sources and doses of total P₂O₅, did not change the total stocks for these elements (Table 2).

The P-IER and S-SO₄²⁻ stocks increased at 36 and 48 months (Figure 1C and 1D), where doses of total P₂O₅ were applied. The P-IER stock in the soil was superior when applying 120 kg ha⁻¹ of total P₂O₅. Higher stocks of S-SO₄²⁻ were observed in the absence of phosphate application (0 kg ha⁻¹ of total P₂O₅). In this case, plants showed a reduction in growth as a result of the absence of P, and consequently, they uptake smaller amounts of nutrients, particularly S-SO₄²⁻.

Table 2. F values for stocks and distribution of total organic carbon (TOC), total nitrogen Kjeldahl (TNK), available P (P-RTI) and S-SO₄²⁻ in the different soil layers.

Sources of variation	DF	Layers (cm)					
		0-30 [†]	0-5	5-10	10-15	15-20	20-30
TOC							
Sources ¹	2	1.32 ^{ns}	0.26 ^{ns}	1.11 ^{ns}	1.84 ^{ns}	0.42 ^{ns}	0.66 ^{ns}
Doses ²	3	1.31 ^{ns}	1.62 ^{ns}	0.35 ^{ns}	1.12 ^{ns}	0.70 ^{ns}	2.11 ^{ns}
Sources x Doses	6	1.73 ^{ns}	0.79 ^{ns}	2.23 ^{ns}	1.53 ^{ns}	1.17 ^{ns}	1.18 ^{ns}
Periods ³	3	27.97 ^{**}	67.10 ^{**}	19.56 ^{**}	31.72 ^{**}	46.63 ^{**}	5.63 ^{**}
Sources x Periods	6	0.02 ^{ns}	1.37 ^{ns}	0.49 ^{ns}	0.14 ^{ns}	0.10 ^{ns}	0.27 ^{ns}
Doses x Periods	9	0.24 ^{ns}	0.69 ^{ns}	2.37 [*]	0.77 ^{ns}	1.17 ^{ns}	0.47 ^{ns}
Sources x Doses x Periods	17	0.24 ^{ns}	0.72 ^{ns}	0.50 ^{ns}	0.23 ^{ns}	0.33 ^{ns}	0.61 ^{ns}
TNK							
Sources	2	1.36 ^{ns}	0.23 ^{ns}	0.42 ^{ns}	3.55 [*]	0.32 ^{ns}	0.48 ^{ns}
Doses	3	1.71 ^{ns}	3.48 [*]	5.39 ^{**}	6.03 ^{**}	1.23 ^{ns}	3.16 [*]
Sources x Doses	6	0.66 ^{ns}	0.32 ^{ns}	0.27 ^{ns}	1.33 ^{ns}	0.39 ^{ns}	2.21 ^{ns}
Periods	3	43.03 ^{**}	3.59 [*]	34.58 ^{**}	53.72 ^{**}	9.92 ^{**}	20.91 ^{**}
Sources x Periods	6	0.53 ^{ns}	0.20 ^{ns}	0.37 ^{ns}	3.60 [*]	0.01 ^{ns}	0.26 ^{ns}
Doses x Periods	9	1.20 ^{ns}	3.49 ^{**}	2.05 [*]	7.31 ^{**}	0.62 ^{ns}	2.44 [*]
Sources x Doses x Periods	17	0.35 ^{ns}	0.18 ^{ns}	0.31 ^{ns}	0.74 ^{ns}	0.14 ^{ns}	0.99 ^{ns}
P-IER							
Sources	2	2.67 ^{ns}	0.37 ^{ns}	3.46 [*]	2.94 [*]	5.57 [*]	0.06 ^{ns}
Doses	3	42.09 ^{**}	50.20 ^{**}	14.66 ^{**}	5.08 ^{**}	4.89 [*]	0.72 ^{ns}
Sources x Doses	6	1.45 ^{ns}	0.58 ^{ns}	1.67 ^{ns}	3.63 ^{ns}	4.10 ^{ns}	1.16 ^{ns}
Periods	3	4.20 ^{**}	4.68 ^{**}	13.20 ^{**}	31.52 ^{**}	12.28 ^{**}	23.16 ^{**}
Sources x Periods	6	1.41 ^{ns}	0.76 ^{ns}	4.10 ^{**}	3.28 ^{**}	6.70 ^{**}	1.38 ^{ns}
Doses x Periods	9	1.24 ^{ns}	2.57 [*]	4.10 ^{**}	1.27 ^{ns}	3.46 ^{**}	4.04 ^{**}
Sources x Doses x Periods	17	0.99 ^{ns}	0.73 ^{ns}	1.75 [*]	2.18 ^{**}	1.83 [*]	0.89 ^{ns}
S-SO₄²⁻							
Sources	2	2.24 ^{ns}	0.25 ^{ns}	4.70 [*]	4.25 [*]	0.66 ^{ns}	1.80 ^{ns}
Doses	3	4.26 ^{**}	0.42 ^{ns}	0.18 ^{ns}	4.49 ^{**}	9.88 ^{**}	2.92 [*]
Sources x Doses	6	0.97 ^{ns}	1.40 ^{ns}	1.36 ^{ns}	2.57 ^{ns}	0.48 ^{ns}	0.28 ^{ns}
Periods	3	5.98 ^{**}	12.61 ^{**}	5.37 ^{**}	15.32 ^{**}	3.02 [*]	15.73 ^{**}
Sources x Periods	6	1.10 ^{ns}	1.59 ^{ns}	2.36 [*]	1.42 ^{ns}	0.50 ^{ns}	3.06 ^{**}
Doses x Periods	9	0.62 ^{ns}	1.23 ^{ns}	2.29 [*]	4.87 ^{**}	2.87 ^{**}	3.05 ^{**}
Sources x Doses x Periods	17	1.15 ^{ns}	1.59 ^{ns}	0.71 ^{ns}	2.32 ^{**}	1.51 ^{ns}	1.72 ^{ns}

[†]Estimated layer, based on the weighted average of the analyzed soil layers, to evaluate total stocks. ¹Source: three phosphate sources (triple superphosphate, rock phosphate and magnesium thermophosphate); ²Dose: four doses of P (0, 60, 120 and 180 kg ha⁻¹ total P₂O₅); ³Periods of soil attribute evaluation: 24, 36, 48 and 60 months after the beginning of the experiment. *P < 0.05. **P < 0.01. ns: not significant.

Distribution of carbon, nitrogen, phosphorus and sulfur in the soil

The experimentation period explained (Table 2), on average, 33%, 34%, 15% and 20% of total data variability, in different soil layers, for TOC, TNK, P-IER and S-SO₄²⁻, respectively. This shows that, over time, conservation agriculture practices in an production system, such as ICLS, can influence on changes in the distribution of TOC, TNK, P and S-SO₄²⁻ in the soil. Higher concentrations of TOC in the soil were observed at 48 and 60 months of experimentation, except for the 20-30 cm soil layer that presented higher concentrations at 24 months (Figure 2).

There was an increase of 83% (1.75 to 3.22 g dm⁻³) in TNK concentrations in the soil 10-15 cm layer, with application of 180 kg ha⁻¹ of total P₂O₅ as RP at 36 months of treatment as compared to other treatments during the same period (Figure 3). For other layers of soil, TNK concentrations did not undergo any change due to P fertilizers; however, the greatest values were observed at 48 and 60 months of experimentation.

The concentrations of P-IER and S-SO₄²⁻, in the 0-5 cm soil layer, were not influenced due to P fertilizers, during five years of experimentation (Table 2). However, higher concentrations of P-IER (Figure 4) and S-SO₄²⁻ (Figure 5), in the 0-5 cm soil layer, was observed at 36 months. In the same conditions was also observed the increase of 34% of P-IER (35.1 to 46.9 mg dm⁻³) as a result from the application of 120 kg ha⁻¹

of total P_2O_5 . For the 5-10, 10-15, 15-20 and 20-30 cm soil layers, higher concentrations of P-IER at 48 months due to the application of 180 kg ha^{-1} of total P_2O_5 as TSP.

Only the experimentation period changed the concentrations of $S-SO_4^{2-}$ in the 5-10 (48 and 60 months) and 15-20 cm layer (36 months). An increase in concentrations was observed in the 10-15 cm layer of soil with annual application of 180 kg ha^{-1} of total P_2O_5 in the form of TSP at 48 and 60 months of experimentation. Also, in the 20-30 cm soil layer, the use of the same dose of total P_2O_5 increased the concentrations of $S-SO_4^{2-}$, but without any influence of different phosphate sources (Figure 5).

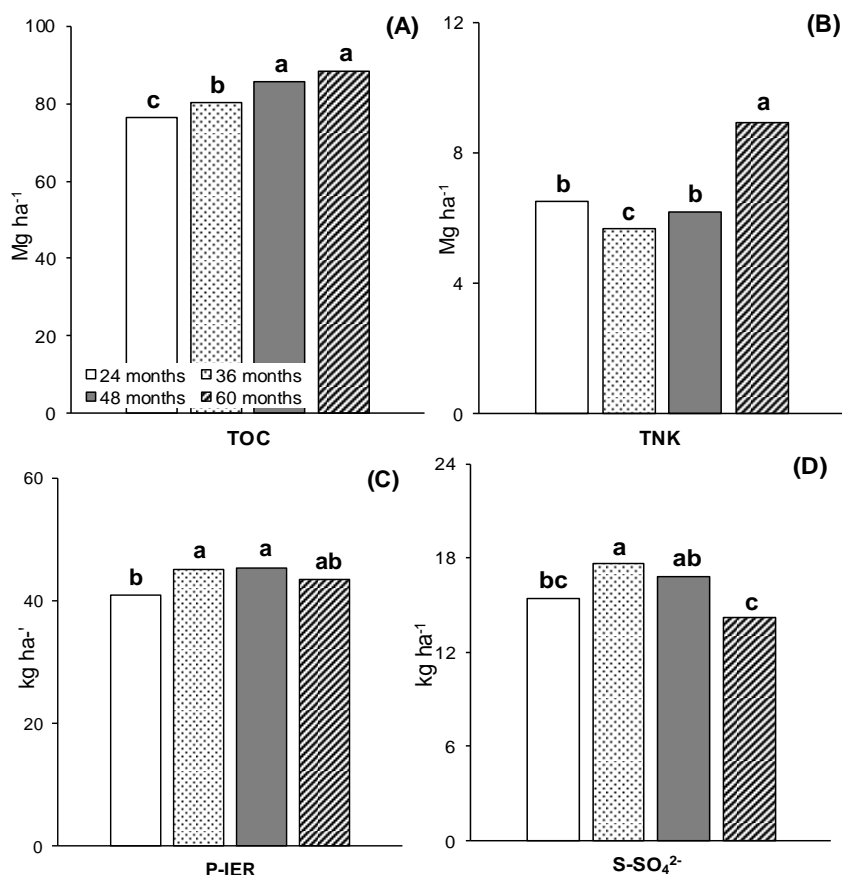


Figure 1. Stocks of (A) total organic carbon (TOC) (C.V.= 8.2%) and total (B) nitrogen Kjeldahl (TNK) (C.V.= 11.7%); and (C) available P by ion exchange resin (P-IER) (C.V.= 9.7%) and (D) available sulfate ($S-SO_4^{2-}$) (C.V.= 8.8%) in the 0-30 cm layer of soil treated with soluble and insoluble in water phosphates, applied annually on soil surface in integrated crop-livestock system. Means followed by the same letters do not differ from each other according to Tukey's test ($P < 0.05$).

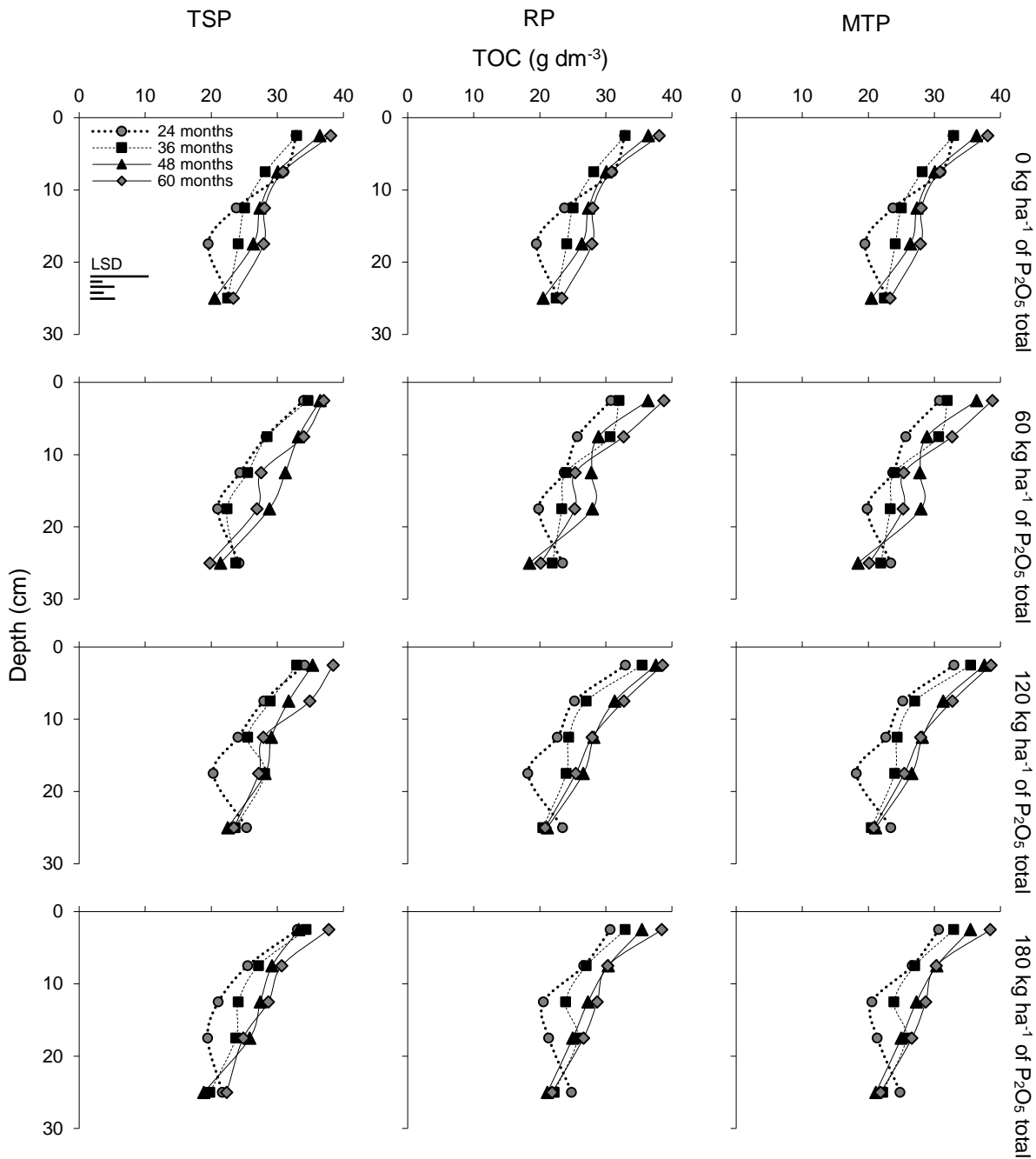


Figure 2. Total organic carbon (TOC) in treated soil, with sources (triple superphosphate – TSP, rock phosphate - RP and magnesium thermophosphate – MTP), and doses (0, 60, 120 and 180 kg ha⁻¹) annually of phosphates applied at soil surface in a total area under an integrated crop-livestock system. Horizontal bars indicate the least significant difference (LSD) for each layer (0-5, 5-10, 10-15, 15-20 and 20-30 cm) of soil. Points are on the basis of the average of four replicates.

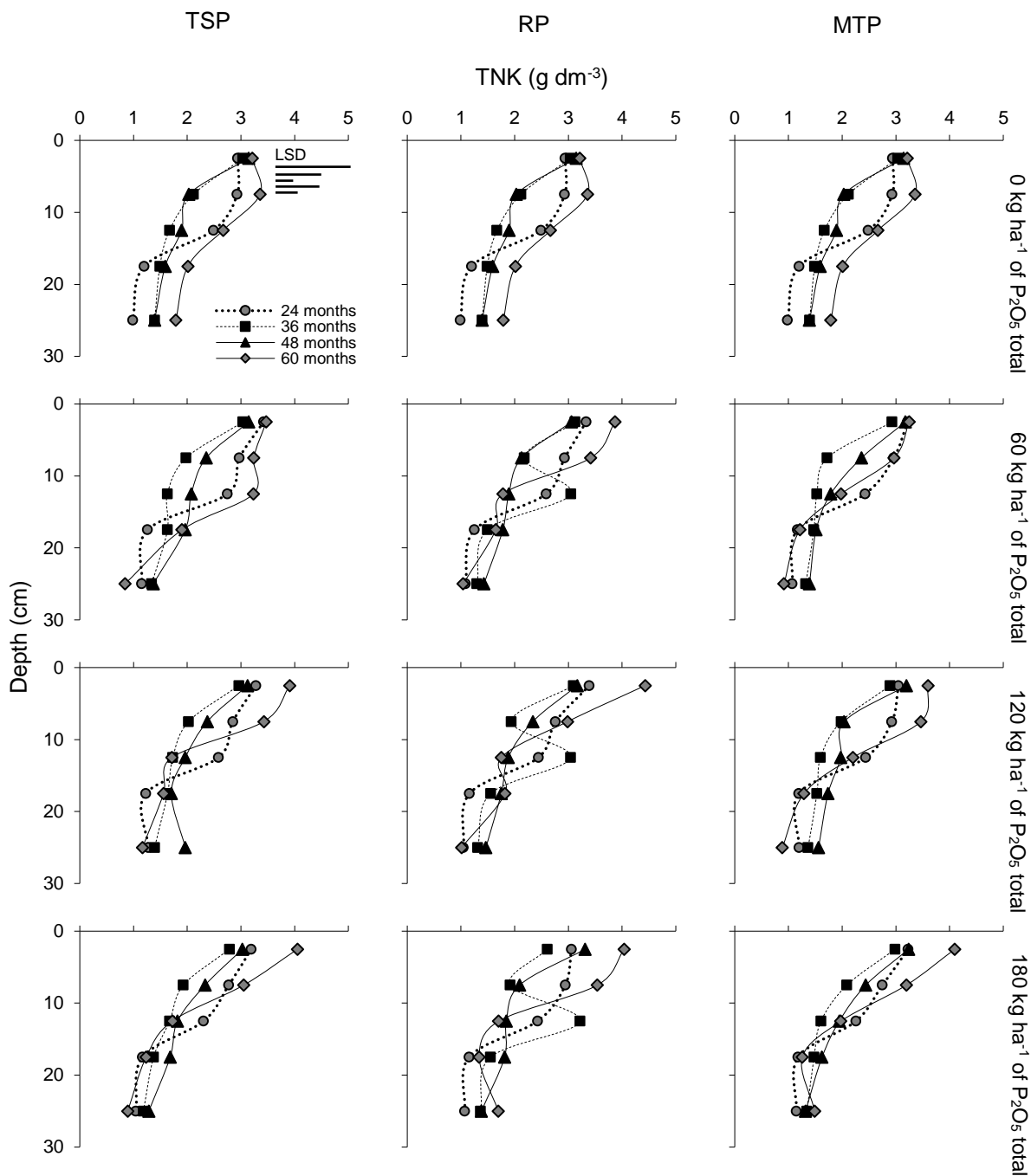


Figure 3. Total nitrogen Kjeldahl (TNK) in treated soil, with sources (triple superphosphate – TSP, rock phosphate - RP and magnesium thermophosphate – MTP), and doses (0, 60, 120 and 180 kg ha⁻¹) annually of phosphates applied at soil surface in a total area under an integrated crop-livestock system. Horizontal bars indicate the least significant difference (LSD) for each layer (0-5, 5-10, 10-15, 15-20 and 20-30 cm) of soil. Points are on the basis of the average of four replicates.

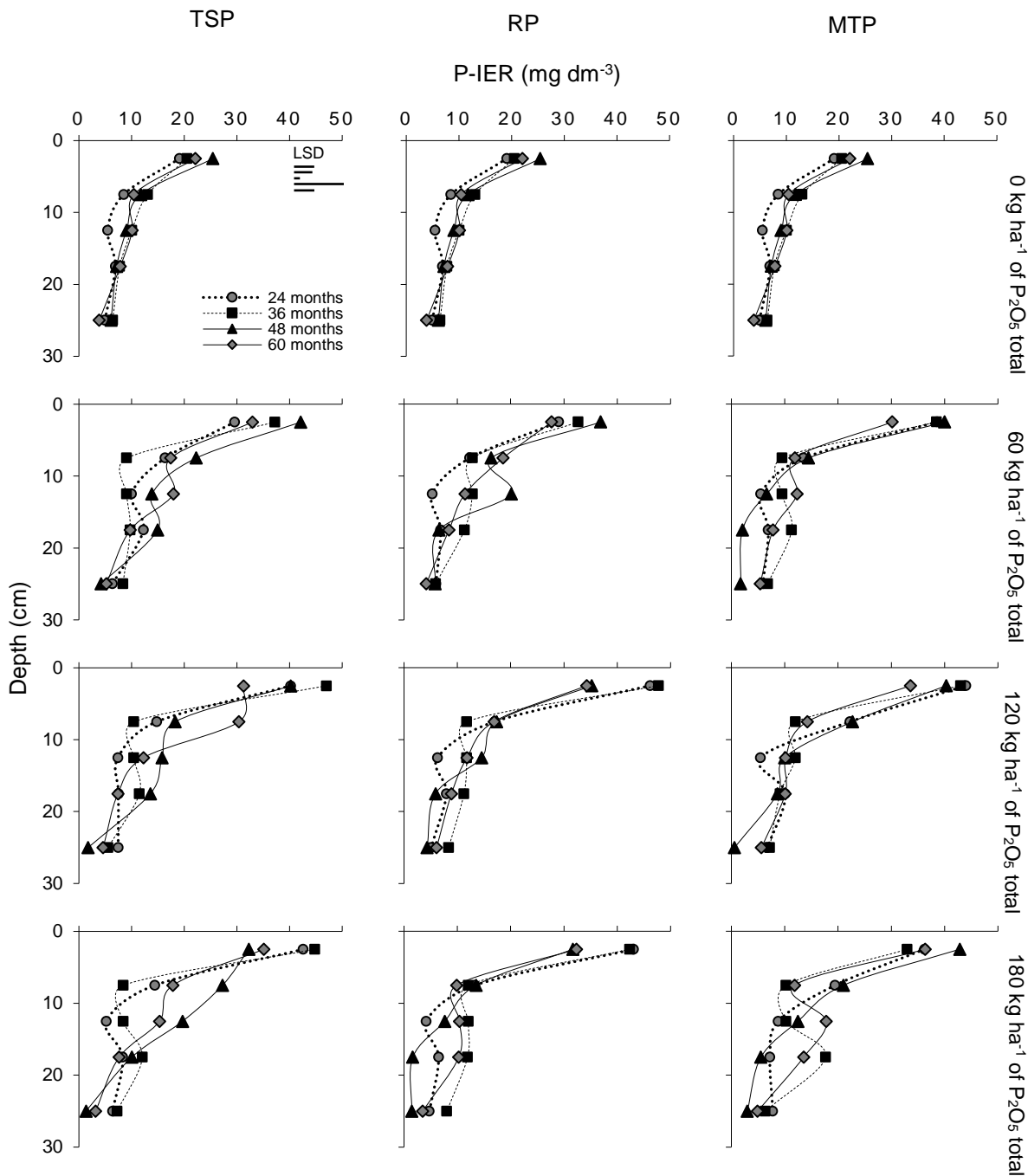


Figure 4. Available phosphorus by ion exchange resin method in treated soil, with sources (triple superphosphate – TSP, rock phosphate - RP and magnesium thermophosphate – MTP), and doses (0, 60, 120 and 180 kg ha⁻¹) annually of phosphates applied at soil surface in a total area under an integrated crop-livestock system. Horizontal bars indicate the least significant difference (LSD) for each layer (0-5, 5-10, 10-15, 15-20 and 20-30 cm) of soil. Points are on the basis of the average of four replicates.

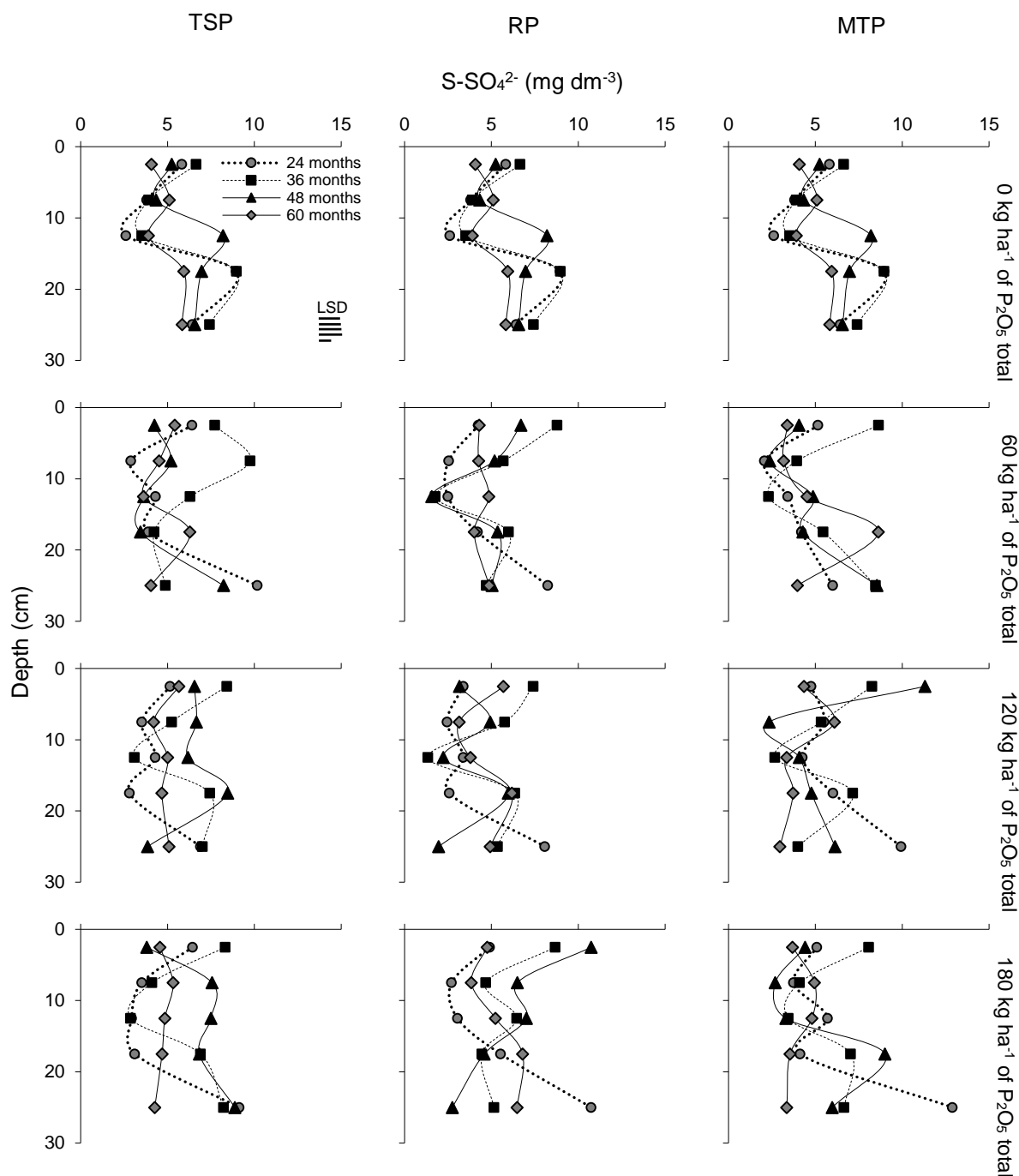


Figure 5. Available sulfur ($S-SO_4^{2-}$) in treated soil with sources (triple superphosphate – TSP, rock phosphate - RP and magnesium thermophosphate – MTP), and doses (0, 60, 120 and 180 $kg\ ha^{-1}$) annually of phosphates applied at soil surface in a total area under an integrated crop-livestock system. Horizontal bars indicate the least significant difference (LSD) for each layer (0-5, 5-10, 10-15, 15-20 and 20-30 cm) of soil. Points are on the basis of the average of four replicates.

DISCUSSION

Impact of soil management system on carbon, nitrogen, phosphorus and sulfur stocks over time

Conservation agriculture (CA) practices, such as ICLS, results in higher deposition and quality of organic residues, and an increase in the C, N, P and S stocks [1,3,18,19]. It has been observed when using global meta-analysis, that CA practices can be increased to 3.3 $Mg\ ha^{-1}$ of organic C in 0-30 cm layer of soil [8]. Increases in N and P stocks were also related in different regions of Brazil under ICLS. For these systems, the 0-30 cm soil layer presented, on average, 3.94 $Mg\ ha^{-1}$ of N and 49.5 $kg\ ha^{-1}$ of P [7]. In our work, N

stocks were 6.81 Mg ha^{-1} of N, on average, during the five years of the experiment, and P stocks (43.7 kg ha^{-1} of P) had similar results, compared to data found in other regions [7].

The fluxes between the soil-plant-atmosphere compartments are modified by introduction of animals into production systems, through ingestion, consumption and return of nutrients to the soil (nutrient cycling) [20,21]. Thus, the cycle of plant and animal residues becomes an important source of maintaining the equilibrium of the agroecosystem, as it contributes to the supply of nutrients to the crops [22], and C sequestration in the soil management systems [23]. In addition, defoliation process imposed by animals allows greater activity in the roots and, consequently, increases release of organic compounds in the soil. In this way, it favors storage of C in the soil, and ameliorates physical, chemical and biological soil attributes, maximizing the release of nutrients through rhizodeposits [1,23].

Small C inputs from release of exudates and root renewal, boost P and S cycling [24]. Soil P stocks increase considerably in ICLS. Compared to native vegetation (21.74 kg ha^{-1} of P), in ICLS, P stocks in the 0-30 cm soil layer more than doubled (49.50 kg ha^{-1} P) with the conversion to these systems [7]. This can be explained by the addition of phosphate fertilizers to cultivated areas [9] and/or deposition of organic residues in the soil [1,9].

However, the dynamics of P, after addition of a source in the soil, vary considerably depending on how it is added (organic waste and/or fertilizer) [24], and this fact may interfere in the dynamics of other anions present in the soil. In our work, phosphate sources did not interfere with changes in P and S stocks in the soil. Nevertheless, the total P_2O_5 doses created a response gradient, where the presence of P in the soil led to the displacement of S. Anions, such as phosphates, have a high affinity for oxides and hydroxides of iron and aluminum. In this case, it has been observed that a decrease of inorganic anions adsorption with equal or lesser affinity, such as sulfate, takes place over a wide pH range. The orthophosphate being present in the soil solution in greater quantity can displace sulfate, due to higher adsorption affinity on colloid surface and increased P in the presence of available S in the soil [25].

Studies on soil P and S stocks and their interactions are scarce, especially in ICLS [8]. P nutrition is highly important, with global crop production restricted by phosphorus-fixing soils. In fact, it is estimated that by the year 2050, a total of 169 to 365 million hectares of land will contain low levels of available P in world [26]. In fact, agriculture places a high demand for phosphate fertilizers. Of the 14 essential nutrients to plants, global reserves of P are the smallest [26-28]. Finally, nutrition S is of increasing importance due to reduction of atmospheric inputs and decrease of additions of S into fertilizers or coated [8]. Thus, studies aimed at elucidating storage and availability processes of P and S for plants are extremely important from the point of view of the sustainability of cropping systems.

Carbon, nitrogen, phosphorus and sulfur soil distribution over time

In general, distribution of TOC, TNK, P-IER and S-SO_4^{2-} in the soil was higher in the upper soil layer (0-15 cm of soil) compared to deep soil layers. Studies have shown that soil C, N and P content decreases with increasing soil depth [4,7]. However, CA practices, such as ICLS, concentrations of C and nutrients increase over time [18,19]. Higher root yield of plants under grazing conditions [1,29] allows a greater accumulation of C in the soil, thus becoming an important pool of C, and also promotes formation of biopores, leading to migration of nutrients to deeper layers of the soil [9]. However, information about distribution of S in the soil under ICLS, are still inconsistent. The applications of different phosphate sources had no influence on soil TOC concentrations (Figure 2), and TNK, P-IER and S-SO_4^{2-} concentrations in the soil surface layer of 0-5 cm. This is due to microbial activity, fed by high levels of C in a superficial layer, which has its mineralization potentiated, releasing organic and inorganic forms in the soil solution [30,31]. Thus, this contributes significantly to maintenance of high concentrations of elements in these layers of soil, not being changed by use of different sources of P applied to the soil. In addition, a higher concentration of P is observed at the soil surface due to low mobility, especially in no-tillage systems [32,33].

In the soil, in the 10-15 cm layer, TNK concentrations increased 83% due to RP treatment up to a maximum dose (180 kg ha^{-1} of total P_2O_5) (Figure 3), at 36 months of experimentation. Under these conditions, there was an increase in production of 1781 kg ha^{-1} of total dry matter in the soybean crop (crop grown in the referred period) compared to other treatments in the same period. Nitrogen is one of the most important ecosystemic nutrients, and its availability is directly linked to primary net production, as well as stabilization

of soil organic matter. The long-term storage of nitrogen-containing organic matter in the soil is attributed to chemical complexity of nitrogen-rich plant residues and soil microbial activity [34].

The distribution of P-IER (below 5 cm soil layer) and S-SO₄²⁻ (10-15 cm soil layer) changed due to application of TSP at a dose of 180 kg ha⁻¹ of total P₂O₅. The major water soluble fertilizers include simple superphosphate (SSP), monoammonium phosphates (MAP), diammonium phosphates (DAP) and TSP. All these compounds are readily soluble in the soil and readily release phosphates in the soil solution for plant uptake [10]. Thus, P concentration increases in the soil, favored by migration of nutrients by formation of bioporosity in ICLS.

For concentrations of S-SO₄²⁻, there are many factors that control sulfate adsorption and leaching. Some of these factors are: (i) soil pH [35]; (ii) the presence of other anions, such as phosphate [35]; and (iii) organic matter content [8]. The superficial layer of no-tillage soil presents unfavorable conditions to sulfate adsorption due to additions of phosphate fertilizers, superficial application of limestone and practices that increase organic matter content of soil. This conditions can lead to displacement of S from superficial layers, for deeper layers of soil, over the long term [36].

In the 0-20 cm layer, P (on average, 17.8 mg dm⁻³) and S (on average, 5.1 mg dm⁻³) concentrations were above critical levels, in addition to a high level of organic C (on average 29 g dm⁻³) [15,37] after 60 months of experimentation. This shows that anticipated fertilization of P is an efficient practice using properly managed conservation systems with no detriment to productivity [4762.8 and 10559.8 kg ha⁻¹ in soybean (2012/2013) and maize (2013/2014), respectively]. In addition, this practice provides greater efficiency in sowing operation, resulting in a reduction of operational and total costs, when compared to the traditional practice (seeding furrow application) [12,38].

During the 60 months of experimentation, experimental conditions favored high concentrations of carbon, nitrogen, phosphorus and sulphur due to: (i) high clay content [11]; (ii) high organic matter content [3]; (iii) high nutrient recycling [2]; (iv) no-tillage and no erosion [39]; (v) correct pasture management [1]; (vi) doses used in response intervals of regional indications [15] resulting in maintenance of soil fertility, productivity and sustainability of the production system.

CONCLUSIONS

Conservation agriculture practices, in an integrated crop-livestock system under Typic Dystrudept, resulted in an increase in total stocks of organic carbon, nitrogen, phosphorus and sulfur available, over five years, regardless of the phosphate source used. The concentrations of phosphorus and sulfur available in the soil were changed with the annual applications of water-soluble phosphate on the soil surface, under no-tillage in an integrated grain crops and forage system. The anticipated fertilization of sources and doses of phosphates applied annually to the soil surface proved to be an efficient practice, guaranteeing supply of P to crops over time and increasing soil fertility, without losses in productivity.

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