

Hydrolysis of Whey Lactose by Immobilized β -Galactosidase

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ABSTRACT

*Hydrolysis of whey lactose to glucose and galactose by immobilized galactosidase comes as an alternative to enlarge the possibilities of commercial use of this feedstock. To be applied at industrial scale, the process should be performed continuously. This work aimed to study the hydrolysis of whey lactose by an immobilized enzyme reactor. β -Galactosidase from *Aspergillus oryzae* was immobilized on silica and activity and stability were evaluated. The best immobilization results were attained by using glutaraldehyde as support's activator and enzyme stabilizer. The optimized enzyme proportion for immobilization was 15-20 mg g⁻¹ of support. Treatments of whey were performed (microfiltration, thermal treatment and ultrafiltration), seeking the elimination of sludge, and the effects on operating the fixed bed reactor were evaluated. Ultrafiltration was the best treatment towards a proper substrate solution for feeding the reactor.*

Key words: β -galactosidase, immobilization, lactase, whey, lactose, hydrolysis

INTRODUCTION

It has been estimated that annually as much as 37 million kg of whey are produced in the USA and Canada (Ghaly and Kamal, 2004) and it is a huge waste disposal problem (Revillion et al., 2003). Lactose use in nutritious products is limited because of low solubility, low sweetener power and laxative effect, if consumed in high concentration (Ferreira et al., 2003). The hydrolysis of lactose in glucose and galactose by β -galactosidase is one important process in the food industry, due to the potentially beneficial effects on assimilating the foods containing lactose, as well as the technological and environmental advantages of industrial applications (Jurado et al., 2002):

- 1- Elaboration of milk with sweetened flavor, of good acceptability, that can be consumed by people with intolerance to this sugar (Jelen and Tos-savainen, 2003);
- 2- Formation of galacto-oligosaccharides during lactose hydrolysis to favor the growth of intestinal bacterial microflora (Playne et al., 2003);
- 3- Improvement in the technological and sensorial characteristics of foods containing hydrolyzed lactose from milk or whey, such as increased solubility (avoidance of lactose crystallization and the grainy aspect of ice-cream and condensed or powdered products) and higher sweetening power (and thus lower caloric content of the products). Many products of low lactose milk and whey find diversified applications, such as food and animal feed

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components, yogurts and ethanol production (Synowiecki and Maciunski, 1999) and

4- In the production of the cheese, the industrial byproducts include whey and permeate that can cause reduced environmental impacts when the lactose is removed (Mukhopadhyay et al., 2003).

Hydrolysis of lactose can be considered as a new unit operation, which can be integrated transforming several feedstocks containing lactose (Pessela et al., 2003). Industries require enzymes having high productivity and stability for repeated use over an extended period. Enzyme immobilization provides many important advantages over the use of soluble enzyme: reusability, continuous operation, controlled product formation, and simplified and efficient processing (Turecek et al., 1990). Continuous reactors provide high productivities and minimize downstream time, enzyme cost and capital investment (Abraham et al., 2004), and offer an attractive design for enzyme reactors, especially when dealing with unclarified substrate fluids (Roy and Gupta, 2003).

The present work had as objective to study immobilization and stabilization of β -galactosidase and application on continuous reactor processing whey as substrate, following research on enzyme isolation and purification (Monti et al., 2000).

MATERIALS AND METHODS

For immobilization of the enzyme, followings were used as support: aminopropyl silica, granulometry 150-300, 106-150 and 63-106 μm and 500 \AA average pore diameter. The β -galactosidase (E.C.3.2.1.23) from *Aspergillus oryzae*, product G5160 and glutaraldehyde (25%) was supplied by Sigma Chemical Co.; lactose was from Synth. Whey was supplied by S3 Nata Ind3ustria e Com3rcio de Produtos Aliment3cios Ltda, Votuporanga, SP. For filtering the whey, Pellicon system (from Millipore) was used with ultra or microfiltration cassettes (1 ft² filtration surface). Analysis of glucose was carried out by spectrophotometry using the enzymatic method (commercial kit, Laborlab). A jacketed chromatographic column was used as reactor (Sigma C5419) with 10cm length and 1cm internal diameter.

Silica Activation and β -Galactosidase Immobilization

Aminoalkyl silica activation was performed by glutaraldehyde (Weetall, 1976). To 1.00 g of silica, glutaraldehyde solution was added (400 μL of 25% glutaraldehyde and 3.60 mL of acetate buffer 0.05M, pH 4.5) and it was left to react for 2 h at room temperature. The supernatant was removed and the enzyme was added, dissolved in 3 mL of the same buffer (15mg/mL). It was left in contact for immobilization during 48 h. Supernatant was removed and reserved for determination of enzymatic activity. Immobilized enzyme was washed and stored at 4°C in the same acetate buffer. Assays were accomplished to optimize the enzyme addition.

Effect of Washings

After activation with glutaraldehyde, the effect of washing the excess of glutaraldehyde with acetate buffer (0.05M, pH 4.5) was studied. Stabilization of the immobilized enzyme by glutaraldehyde after the immobilization (crosslinking) was also evaluated. Tests were accomplished as follows: (1) immobilization and no crosslinking; (2) immobilization, following washing and crosslinking, and (3) immobilization without washing or crosslinking.

Determination of the Activity

Lactose solution (25mL of 4.5%) in acetate buffer (0.05 M pH 4.5) was added in the batch reactor and stabilized at 45°C before addition of the enzyme, solution or immobilized. Samples (0.5mL) were withdrawn at 3 min intervals and inactivated in the boiling water bath for 5min and then 25 μL was added to 1mL of the glucose enzymatic reagent. Absorbances were read after 1 h at 505nm in UV-Visible spectrophotometer (Shimadzu model UV-1203). Glucose concentration was calculated by comparing with standard glucose concentrations.

Operation of fixed bed reactor with lactose solution and whey as substrate

The immobilized enzyme (1.00 g) was packed in the jacketed fixed bed reactor (1cm diameter and 10cm length). The system was fed with 4.5% lactose solution, using a peristaltic pump (Maxterflex,

Cole-Parmer, 1-100rpm). Operational stability was studied at three temperatures (40, 45 and 55°C), working on two flow rates at each temperature (0.5mL/min and 5.0mL/min). In another assay, the system was fed with processed whey at the same flow rates at 40 and 45°C.

Treatments of the whey to be used as substrate

- 1- Whey microfiltrated using 0.22 μ m membrane by the Pellicon system (Millipore);
- 2- Whey heated for 5min at 90-95°C and microfiltrated through 0.22 μ m membrane, and,

3- Whey microfiltrated through 0.45 μ m membrane and ultrafiltrated through 10 kDa MWCO by the Pellicon system.

RESULTS AND DISCUSSION

Immobilization and stabilization of β -Galactosidase

In relation to the concentration of enzyme added for immobilization, it was observed that between 20 and 40 mg/mL, the immobilized activity remained practically constant (Fig. 1).

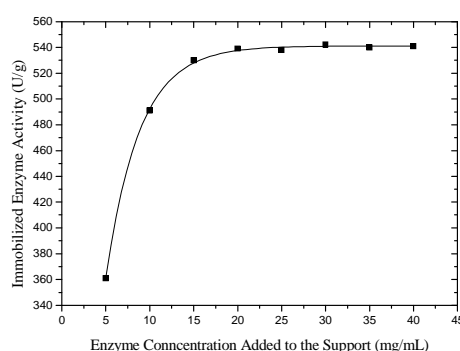


Figure 1 - Effect of enzyme concentration on the immobilization. 2.0 mL of enzyme solution added to 200 mg of silica.

Hence a concentration of β -galactosidase between 15 and 20mg/mL was selected, since above this value, there was only increase of enzyme consumption.

Considering that glutaraldehyde addition to the immobilized β -galactosidase resulted in no significant decay of activity (from 775U/g to 750U/g), stabilization was evaluated by adding the same solution as for activation of the support: 2.5% in sodium acetate buffer (0.05M, pH 4.5). Stability assays were performed at 65°C, taking two samples, after 15 and 30 minutes. Additionally, activation of the support followed by removing the supernatant and addition of enzyme solution, without washing, was considered.

It was observed that the immobilization after washing the glutaraldehyde and without crosslink

ing resulted in higher activity and less stability; using crosslinking the activity decreased, but the stability was better. The immobilization without washing the glutaraldehyde before enzyme addition and followed by the treatment with glutaraldehyde had the highest activity, initially and after incubation at 65°C (Fig. 2). This immobilized product was the best obtained.

Fixed bed reactor with lactose solution as substrate

In a first stage, immobilized enzyme was evaluated using lactose solution as substrate. At 55°C, 5,0mL/min, initial activity was 450U/g in the reactor, with relative high decrease over 11 days (66%), and the half-life was approximately seven days.

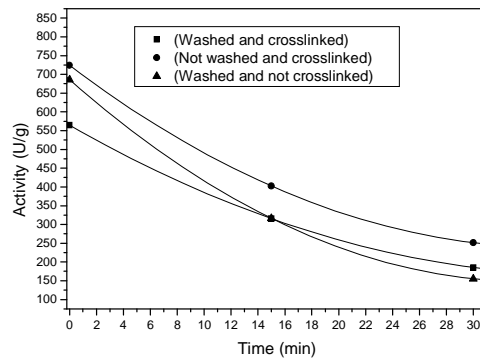


Figure 2 - Effect of washing and crosslinking with glutaraldehyde on β -galactosidase deactivation at 65°C.

At 45°C, initial activity was 410U/g and the reactor operated for 100 days, reduced 53% of the activity. The half-life was about 100 days. At 40°C, the initial activity was slightly lower

(400U/g) and the stability was better; the reactor operated for 130 days, showed reduced activity (32%). The half-life was about 12 months, as shown by the slope of the plotted data (Fig. 3).

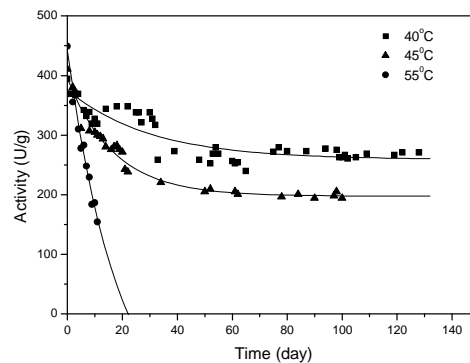


Figure 3 - Activity and inactivation of β -galactosidase in fixed bed reactors using lactose solution as substrate at 5.0 mL/min flow rate.

During operation at 0.5 mL/min, with high conversion, initial activity at 55°C was 112U/g, with decay of 40% after 11 days, and the half-life was above 11 days. At 45°C, the initial activity was 106U/g and the reactor operated for 100 days, decaying 29% of the activity, with half-life above 100 days. At 40°C, the initial activity was 100U/g,

but the stability better; the reactor operated for 130 days, activity was reduced 24%, with half-life of about 12 months (Fig. 4).

Considering results with lactose solution as substrate, it was opted to run the fixed bed reactor with treated whey at 40°C and 45°C, temperatures at which the enzyme was more stable.

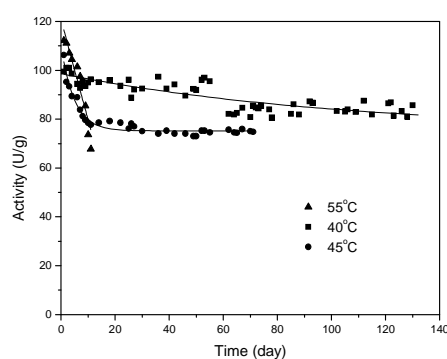


Figure 4 - Activity and inactivation of the enzyme in reactors of fixed bed using lactose solution as substrate in the 0.5mL/min flow

Whey treatments and operational stability in fixed bed reactor

The first treatment, to which the crude whey was submitted, consisted of microfiltration through 0.22 μ m membrane. The purpose was to obtain the clarified whey, free from the fat and insoluble protein that could cause blockage of the reactor. The microfiltrated whey was limpid and apparently suitable as substrate. However, during operation at 45°C, precipitation occurred inside the reactor. This precipitated material consisted of whey proteins subjected to denaturation at the reactor temperature. In an attempt to decrease or to eliminate the precipitation, the temperature was reduced to 40°C, but without success. The reactor was operated at 45°C for 40 days and the final activity of the immobilized enzyme was 162U/g (30% of the initial). Analysis of the decrease of the glucose concentration at the exit of the reactor indicated

final activity of 40%, at 5mL/min, with half-life of about 32 days.

In order to increase the half-life of the enzyme, and reduce the precipitation, crude whey was submitted to thermal treatment at 90-95°C, for 5min with subsequent filtration through 0.22 μ m membrane. The objective was to eliminate the proteins and fats, coagulated by the high temperature, facilitating the microfiltration step. This treatment induced the precipitation of albumins and globulins, allowing the removing colloidal material and fat.

Operation of the reactor at 45°C with thermally treated whey still resulted in precipitate, but to much less extent. The reactor was run for 44 days and the final activity of the immobilized enzyme was 132U/g (20% of the initial). Analysis of the decay of glucose concentration indicated final activity of 44%, at 5mL/min, with half-life around 40 days (Fig. 6).

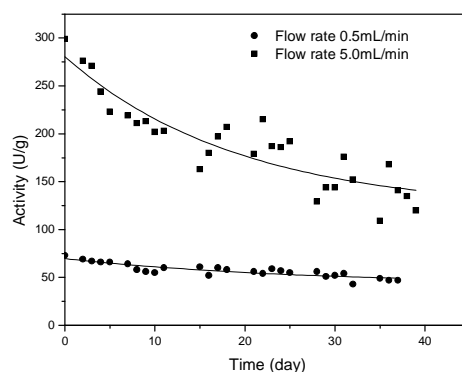


Figure 5 - Fixed bed reactor with microfiltrated whey as substrate, at 45°C.

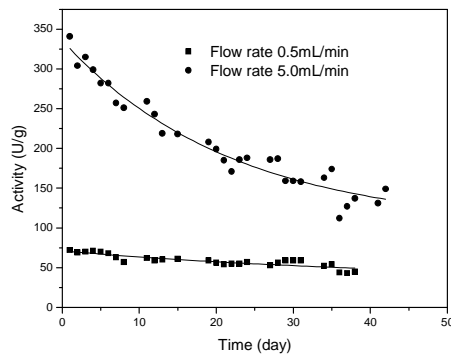


Figure 6 - Fixed bed reactor of with thermally treated and microfiltrated whey as substrate, at 45°C.

Despite each treatment to which the whey was submitted, solid accumulation in the reactor still persisted. According to Toledo (1997), a way to remove this from whey would be by ultrafiltration. During ultrafiltration, the components of low molecular weight as lactose, minerals and vitamins leave with the permeate, while proteins become concentrated.

Reactor operation with ultrafiltrated whey was performed for 51 days and the final immobilized enzyme activity was 197U/g (32% of the initial). Analysis on the decay of glucose concentration indicated the final activity of 50% at 5.0mL/min, with half-life of approximately 45 days (Fig. 7).

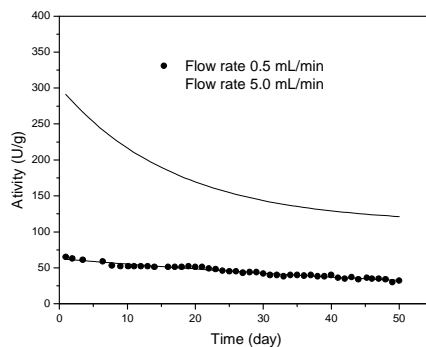


Figure 7 - Fixed bed reactor with micro- and ultrafiltrated (10kDa) whey, at 45°C.

Ultrafiltration of the whey allowed the operation of the reactor for a longer period, since there was no obstruction of the reactor as before. However, this didn't guarantee the stability of the enzyme. Comparing half-life of 45 days at 45°C, with ultrafiltrated whey, with 100 days with lactose solution, it seemed that additional treatment was necessary to optimize the process. It was suggested to remove salts and take special cares as cleaning of the bioreactor, that could help to preserve the integrity of the immobilized enzyme (Vyas and Tong, 2003).

CONCLUSIONS

In relation to the methodology of β -galactosidase immobilization, good results were achieved using glutaraldehyde (2.5% solution in acetate buffer at 0.05 mol.L⁻¹, pH 4.5) as activator and stabilizer.

It was found to be better not to wash the activated silica for the addition of the enzyme, but just to remove the supernatant. Addition of the same glutaraldehyde solution after immobilization resulted in increased stability of the enzyme.

The concentration of β -galactosidase added in the immobilization should be 15-20 mg/mL (10 mL/g of support). Larger amounts just increased the

enzyme consumption, without significant increase in the activity. The activity of immobilized β -galactosidase was around 650 U/g.

On long-term operation, with lactose solution at 40°C, initial activity of 400U/g was obtained in the reactor and the half-life of the biocatalyst was estimated to be 12 months. This result confirmed the good performance of the immobilized enzyme and allowed to establish the half-life limit for operation with whey.

Whey was submitted to three treatments, aiming to increase the operational stability. Microfiltration through 0.22 μ m membrane resulted only in the clarification of the whey. Heating followed by microfiltration reduced sensibly solid accumulation, but did not hinder the blockage. Only the ultrafiltration through 10kDa MWCO membrane eliminated the obstruction, but the half-life of the enzyme did not increase significantly (45 days at 45°C, compared with 40 days for the treatment 2 and 130 days for lactose solution). Then, although the ultrafiltration guaranteed the operational stability to the system, it was not good enough to increase appreciably the lifetime of the biocatalyst. The first treatment to be tested would be the demineralization, preferentially for electro dialysis (Rozhkova et al., 1992).

Another important factor to be considered would be the operation in continuous system where the whey could be ultrafiltered and pumped straight to the reactor, reducing the risk of microbial contamination, as was observed sometimes during the process.

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RESUMO

A hidrólise de lactose de soro de leite, resultando em glicose e galactose, apresenta-se como uma alternativa para ampliar as possibilidades de uso comercial deste insumo. Para ser aplicado em escala industrial, o processo deve ser operado de modo contínuo. Reporta-se o estudo de um sistema objetivando hidrólise de lactose de soro de leite através de um reator com enzima imobilizada.

β -Galactosidase de *Aspergillus oryzae* foi imobilizada em sílica, sendo avaliadas a estabilidade e atividade. Os melhores resultados de imobilização foram obtidos usando glutaraldeído como ativante do suporte e estabilizador da enzima. A proporção otimizada entre enzima e suporte foi 15-20 mg.g⁻¹. Foram estudadas formas de tratamento do soro (microfiltração, tratamento térmico e ultrafiltração), objetivando eliminação de material suspenso, e avaliando os efeitos na operação de reator de leite fixo. A ultrafiltração foi o melhor tratamento, na busca de uma solução de substrato apropriada para o reator contínuo.

REFERENCES

- Abraham, T. E.; Joseph, J. R.; Bindhu, L. B. V. and Jayakumar, K. K. (2004), Crosslinked enzyme crystals of glucoamylase as a potent catalyst for biotransformations. *Carbohydrate Research*, 339, 1099-1104.
- Ferreira, L. S.; Souza, M. B.; Trierweiler, J. O.; Hitzmann, B. and Folly, R. O. M. (2003), Analysis of experimental biosensor/FIA lactose measurements. *Brazilian Journal of Chemical Engineering*, 20, 7-13.
- Ghaly, A. E. and Kamal, M. A. (2004), Submerged yeast fermentation of acid cheese whey for protein production and pollution potential reduction. *Water Research*, 38, 631-644.
- Jelen, P. and Tossavainen, O. (2003), Low lactose and lactose-free milk and dairy products - prospects, technologies and applications. *Australian Journal of Dairy Technology*, 58, 161-165.
- Jurado, E.; Camacho, F.; Luzon, G. and Vicaria, J. M. (2002), A new kinetic model proposed for enzymatic hydrolysis of lactose by a beta-galactosidase from *Kluyveromyces fragilis*. *Enzyme and Microbial Technology*, 31, 300-309.
- Monti, R.; Basilio, C. A.; Trevisan, H. C. and Contiero, J. (2000), Purification of papain from fresh latex of *Carica papaya*. *Brazilian Archives of Biology and Technology*, 43, 501-507.
- Mukhopadhyay, R.; Talukdar, D.; Chatterjee, B. P. and Guha, A. K. (2003), Whey processing with chitosan and isolation of lactose. *Process Biochemistry*, 39, 381-385.
- Pessela, B. C. C.; Mateo, C.; Fuentes, M.; Vian, A.; Garcia, J. L.; Carrascosa, A. V.; Guisan, J. M. and Fernandez-Lafuente, R. (2003), The immobilization of a thermophilic beta-galactosidase on Sepabeads supports decreases product inhibition - Complete hydrolysis of lactose in dairy products. *Enzyme and Microbial Technology*, 33, 199-205.

- Playne, M. J.; Bennett, L. E. and Smithers, G. W. (2003), Functional dairy foods and ingredients. *Australian Journal of Dairy Technology*, 58, 242-264.
- Revillion, J. P. D.; Brandelli, A. and Ayub, M. A. Z. (2003), Production of yeast extract from whey using *Kluyveromyces marxianus*. *Brazilian Archives of Biology and Technology*, 46, 121-127.
- Roy, I. and Gupta, M. N. (2003), Lactose hydrolysis by Lactozym (TM) immobilized on cellulose beads in batch and fluidized bed modes. *Process Biochemistry*, 39, 325-332.
- Rozhkova, M. V.; Shaposhnik, V. A.; Mizilina, A. K. and Tyagunova, V. I. (1992), Lactose transport through ion-exchange membranes in electrodialysis. *Journal of Applied Chemistry of the Ussr*, 65, 2071-2074.
- Synowiecki, J. and Maciunska, J. (1999), Lactose in human and animal nutrition. *Medycyna Weterynaryjna*, 55, 497-500.
- Toledo, R.T. (1997) *Fundamentals of Food Processing Engineering*. Chapman and Hall, New York.
- Turecek, P. L.; Pittner, F. and Birkner, F. (1990), Degradation of polysaccharides by immobilized depolymerizing enzymes. *International Journal of Food Science and Technology*, 25, 1-8.
- Vyas, H. K. and Tong, P. S. (2003), Process for calcium retention during skim milk ultrafiltration. *Journal of Dairy Science*, 86, 2761-2766.
- Weetall, H. H. (1976), Immobilized enzyme technology. *Cereal Foods World*, 21, 581-587.

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