

Characterization and Metal Detoxification Potential of Moderately Thermophilic *Bacillus cereus* from Geothermal Springs of Himalaya

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ABSTRACT

Two thermophilic *Bacillus cereus* strains (*B. cereus*-TA2 and *B. cereus*-TA4) used in the present study were isolated from the geothermal spring of Hunza valley, Gilgit, Pakistan. They showed the ability to withstand and grow at high temperature (85°C). Both these strains could resist multiple metals (copper, cadmium, mercury, manganese, zinc, arsenic, chromium and selenium). Strain *B. cereus*-TA4 reduced Cr (VI) at pH 5.0 to 9.0 but maximum reduction (83%) was observed at pH 7.0 after 48 h when initially supplied with 200 µg mL⁻¹ of K₂CrO₄. Lower initial concentrations such as 100 µg mL⁻¹ supported higher reduction (90 to 95%) than that of high concentration such as 500 µg mL⁻¹ (20 to 30%). Both the strains reduced nearly 70% of Se (IV) after 48 h of growth at pH 7.0 when initially supplied with 200 µg mL⁻¹ of Na₂SeO₃. The optimum temperature for maximum Se (IV) reduction was 45°C for both the strains.

Key words: chromium, bioremediation, selenium, heavy metals

INTRODUCTION

Vast scientific and technological development in the modern world has also brought numerous challenges in the field of environmental protection and its management (Bennett et al. 2003). Metals such as copper, chromium, mercury, lead, zinc, nickel, arsenic and selenium are considered as the major environmental contaminants (Pekey et al. 2010). They can cause serious problems to the organisms when present above certain level. Chromium is considered as non-essential and toxic metal for the microorganisms and plants while it regulates glucose utilization in the animals (Shelnutt et al. 2007). The wide-spread industrial use of Cr has caused a serious environmental concern. Chromium exists in nature as Cr (III) and Cr (VI); both differ in terms of mobility,

bioavailability and toxicity. Cr (III) is an essential micronutrient for normal glucose utilization in the animals (Krikorian et al. 2010). Hexavalent chromium is highly soluble, more mobile and is the most toxic form of chromium known. It is necessary to convert this carcinogenic form of chromium (Cr VI) to less mobile and less toxic trivalent chromium. Chemical reduction of hexavalent form into trivalent form is a known method employed for chromate decontamination (Bewley and Clarke 2010; Cismasiu 2011). Nowadays, microorganisms, especially bacteria are considered as good option for the conversion of hexavalent form into trivalent form (Liao et al. 2014).

Another metal, which has become the element of interest to many investigators because of its toxicological and physiological importance is

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selenium. Selenium exists in the soil as trace element and is very important component of human diet, while its high concentration is toxic (Rayman 2000). Three forms of selenium (selenate, selenite and elemental selenium) are more prominent in nature. Selenate and selenite are toxic while elemental selenium being the insoluble in water is less mobile and usually remains in the soils posing a smaller risk to the environment (Fesharaki et al. 2010). Biological methods for the remediation of toxic forms of chromium and selenium are environment friendly (Hunter 2007; Hunter and Manter 2008; Verma et al. 2009). Bioremediation is inexpensive and established technology, which is commonly used in the environment without posing any damaging effects to the ecosystem (Khan et al. 2009).

The purpose of this study was to isolate novel microbes, especially from some extreme environment and to utilize them for the bioremediation of chromium and selenium polluted soils and wastewater.

MATERIALS AND METHODS

Bacterial isolation and characterization

Bacterial strains used in this study were isolated from the northern hilly (8200 ft) geothermal springs of Hunza Gilgit, Pakistan. Water and soil samples were collected and transported to the laboratory in controlled conditions. Some physico-chemical parameters such as water temperature (65°C), pH (6.5) and air temperature (2°C) were recorded on the spot. Water and soil samples were diluted and spread on nutrient agar plates and incubated at different temperatures. After 24 h, different colonies of strains were picked and purified. The strains were basically characterized morphologically and biochemically following Gerhardt et al. (1994). The effect of temperature and pH on bacterial growth was studied by growing the strains in nutrient broth at 45, 55, 65, 75 and 85°C and pH 5.0, 7.0 and 9.0 for 24 h. Metal resistance profile of the isolated strains was also determined against copper, cadmium, mercury, manganese, zinc, arsenic, chromium and selenium.

CR (VI) Reduction

Cr (VI) reduction by these bacterial strains was evaluated by using the chromate reduction medium (g/L) tryptone 10, yeast extract 5, NaCl 5,

citric acid 1, Na₂HPO₄ 6.9 (Deleo and Ehrlich 1994). Hexavalent chromium reduction was monitored at 45, 55 and 65°C and pH 5.0, 7.0 and 9.0, incubation times of 48 and 96 h and initial chromate concentrations of 100, 250 and 500 µg mL⁻¹. About 250 mL of chromium reduction media was prepared and inoculated with 1.0 mL of fresh culture of bacterial strains TA2 and TA4. On regular time period, flasks along with samples were withdrawn and Cr (VI) reduction was estimated by using 1,5-diphenylcarbazide following Clesceri et al. (1998).

SE (IV) Reduction

All the selenite reduction experiments were carried out under aerobic conditions. Selenite reduction was monitored at same temperatures, pH, incubation times and initial Na₂SeO₃ concentrations as in the above experiment. Two hundred and fifty milliliter of selenite reduction medium was prepared and inoculated with 1.0 mL of fresh culture of bacterial strains TA2 and TA4. Medium was also amended with sodium acetate 2.0 g L⁻¹ as carbon source. At 48 and 96 h incubation, cultures were withdrawn taking 1.0 mL of sample and centrifuged at 14,000 xg for five min. Brown and Watkinson (1977) method was used for the estimation of selenite reduction in the supernatant. The absorbance was measured at 377 nm using UV-vis spectrophotometer (Cecil 7200, UK).

16S rRNA gene sequencing

For the exact identification of these strains (TA2 and TA4), ribotyping was carried out. The sequencing services were carried out using Big Dye terminator cycle sequencing kit (Applied Biosynthesis, USA). Sequencing products were resolved on an Applied Biosystems model 13730 XL automated DNA sequencing system. Then the sequences were analyzed by using BLAST from NCBI web site also by multiple sequence alignments through Clustal W.

Statistical analysis

Data obtained were statistical analysis by using student package statistical software v11.0.

RESULTS

Strains isolation and identification

More than 50 bacterial strains were isolated from the soil and water of geothermal springs of Chillas.

Two strains, TA2 and TA4 could grow up to a temperature of 90°C in nutrient broth and were selected for further study. On the basis of various morphological, biochemical, physiological and 16S rRNA gene sequencing study, both the strains were identified as *Bacillus cereus* (Table 1). The accession number of strains *B.cereus*-TA2 and *B.cereus*-TA4 are GU980764 and GU980765, respectively.

Strains characterization

Both the strains were Gram positive motile rods

and formed the spores under unfavorable conditions (Table 1). These were facultative anaerobes and were able to reduce nitrate. They showed resistances against copper, cadmium, mercury, manganese, zinc, arsenic, chromium and selenium (Table 2) and tolerated 700 µg mL⁻¹ of Cu, 400 µg mL⁻¹ of Cd, 1500 µg mL⁻¹ of Mn, 900 µg mL⁻¹ of As, 600 µg mL⁻¹ of Cr and 1000 µg mL⁻¹ of Se. *B. cereus*-TA2 tolerated 200 µg mL⁻¹ of Hg while this was 150 µg mL⁻¹ for other strain (Table 2). These strains showed optimum growth at 55°C and pH7.0 (Figs. 1A, B).

Table 1 - Morphological and biochemical characteristics of strains.

Characteristics	Strains		Characteristics	Strains	
	<i>Bacillus cereus</i> -TA2	<i>Bacillus cereus</i> -TA4		<i>Bacillus cereus</i> -TA2	<i>Bacillus cereus</i> -TA4
Colony shape	round	round	Lactose	-	-
Colony size (mm)	3.5	4	Inositol	-	-
Cell shape	rods	rods	D-sorbitol	-	-
Cell size (µm)	1.2-2.4	1.2-2	L-rhamnose	-	-
Gram staining	+ve	+ve	D-sucrose	-	-
Capsules staining	-	-	D-melibiose	-	-
Spore staining	+	+	Amygdalin	-	-
Motility	+	+	L-Arabinose	-	-
ornithine decarboxylase	-	-	Glucose	+	+
Urea	-	-	Oxidase	+	+
Arginine dihydrolase	-	-	Catalase	+	+
Indole production	-	-	Nitrate reduction	+	+
lysine decarboxylase	-	-	Growth on	-	-
Citrate utilization	+	+	MacConkey agar		
acetoin production	+	+	O.F test	F.A	F.A
gelatin hydrolysis	+	+	H2S production	-	-
			Starch hydrolysis	-	-

OF, Oxidation fermentation; -, Negative; +, Positive

Table 2 - Heavy metals resistance profile of chromium resistant bacterial strains.

Metals concentration (µg mL ⁻¹)	Strains	
	<i>Bacillus cereus</i> -TA2	<i>Bacillus cereus</i> -TA4
Cu	700	700
Cd	400	400
Hg	200	150
Mn	1500	1500
Zn	200	100
As	900	900
Cr	600	600
Se	1000	1000

CR (VI) Reduction

Both the strains were able to reduced carcinogenic Cr (VI) into less toxic Cr (III) aerobically at various pH and incubation times (Fig. 1C). *B. cereus*-TA4 reduced Cr (VI) at pH 5.0 and 9.0 but maximum reduction (83%) was observed at pH7.0

after 48 h of incubation when initially supplied with 200 µg mL⁻¹ K₂CrO₄. Almost same trend was observed after 96 h incubation period. As both the strains grew at higher temperature, the reduction potential was also observed at various temperatures (45-65°C). Increasing temperature led to more reduction of hexavalent chromium. The most favorite temperature for optimum chromate reduction was 55°C for both the strains (Fig. 1D). But above 55°C, a slight decrease in the reduction potential in both the strains was recorded. The rate of chromate removal was faster in the early 48 h after which there was no significant impact on the Cr (VI) reduction potential of these strains. Figure 2 shows the effects of various initial Cr (VI) concentrations on the reduction ability of both the strains. Lower initial concentration such as 100 µg mL⁻¹ supported higher reduction (90 to 95%) than high concentration such as 500 µg mL⁻¹ (20 to 30%).

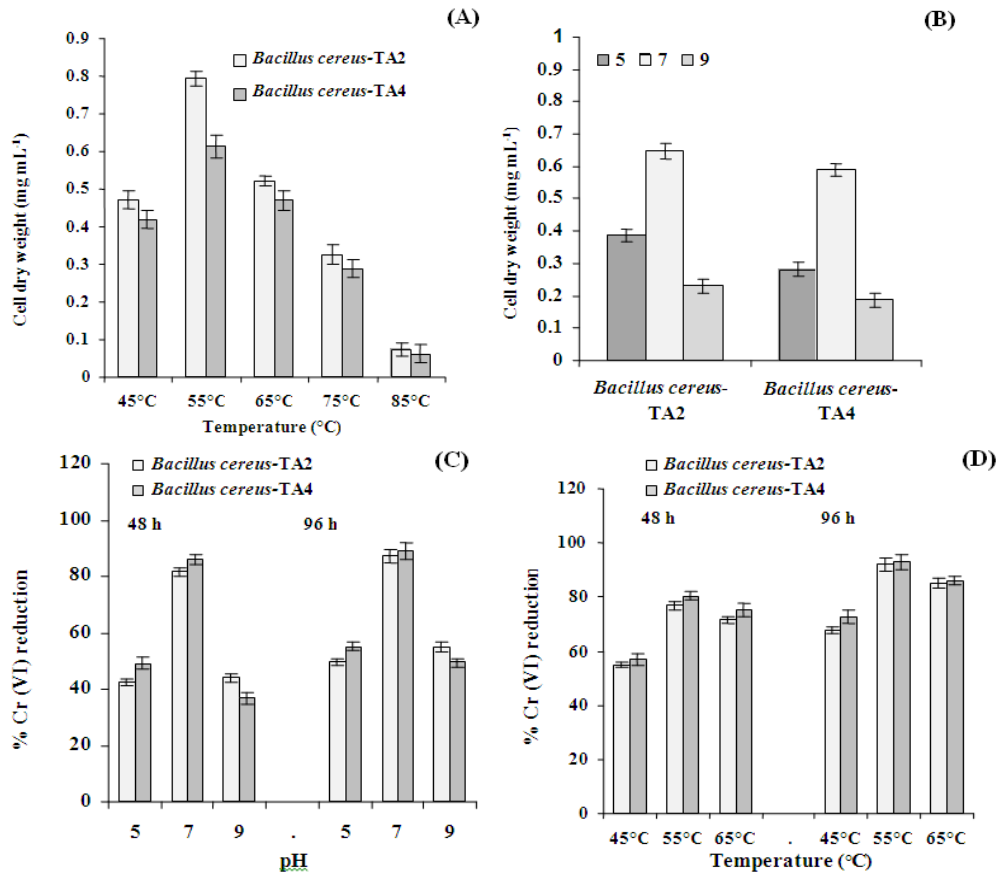


Figure 1 - Growth of strains TA2 and TA4 (A) at 45, 55, 65, 75 and 85°C, (B) at pH 5.0, 7.0 and 9.0; Reduction of K₂CrO₄ (C) at pH 5.0, 7.0 and 9.0 and (D) at 45, 55 and 65°C. Reduction was monitored after 48 and 96 h of growth incubation. Initial K₂CrO₄ concentration used for reduction experiment was 200 µg mL⁻¹.

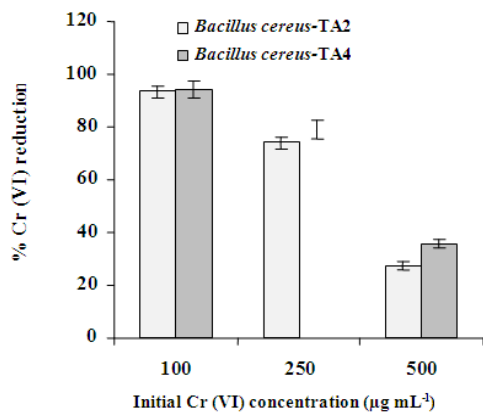


Figure 2 - Reduction of K₂CrO₄ at three initial K₂CrO₄ concentrations (100, 250 and 500 µg mL⁻¹). Reduction was monitored after 48 h of growth at pH 7.0 and 55°C.

SE (IV) REDUCTION

Results showed that both the strains had reduced selenite into elemental selenium under aerobic conditions (Fig. 3). Reduction of selenite was more at pH7.0 as compared to the other pH values. Both the strain reduced nearly 70% of total Se (IV) after 48 h of growth at pH7.0 (Fig. 3). Interestingly, Se (IV) reduction potential of the strain TA4 was more at pH5.0 and 9.0 than at pH7.0 after 48 and 96 h as compared to the strain TA2. The optimum temperature for maximum Se (IV) reduction was 45°C for both the strains. As seen from Figure 4, Se (IV) reduction occurred at various initial selenite concentrations (100 to 500 µg mL⁻¹). The percentage selenite reduction was highest at lowest initial selenite concentrations but the amount of Se (IV) reduced was maximum at highest initial selenite concentration (500 µg mL⁻¹).

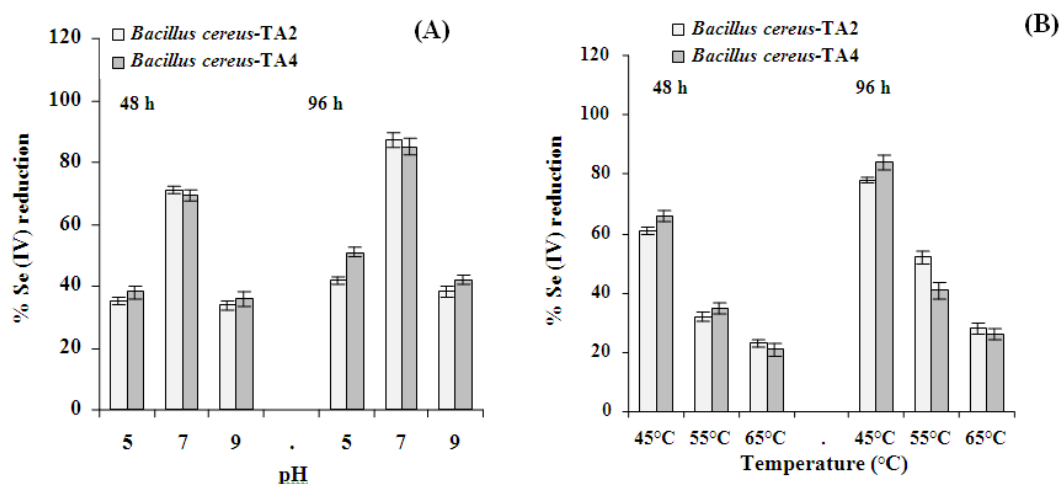


Figure 3 - Reduction of Na_2SeO_3 at pH 5.0, 7.0 and 9.0 and 45, 55, and 65°C. Reduction was monitored after 48 and 96 h of growth. Initial Na_2SeO_3 concentration used for reduction experiment was $200 \mu\text{g mL}^{-1}$.

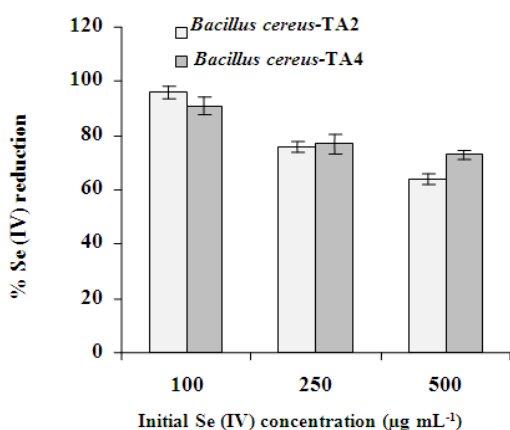


Figure 4 - Reduction of selenite at three initial Na_2SeO_3 concentrations (100, 250 and $500 \mu\text{g mL}^{-1}$). Cultures were harvested after 48 h of growth at pH 7.0 and temperature 55°C.

DISCUSSION

Present study deals with two metal resistant bacterial strains *B. cereus*-TA2 and *B. cereus*-TA4, which were isolated from the hot spring of Hunza valley, Gilgit (Himalayan range), Pakistan. Both these strains had the ability to tolerate and reduce chromium and selenium simultaneously. Interestingly, the strains were isolated from a place, which was not previously or presently contaminated with heavy metals (chromium and selenium) due to anthropogenic activity. Dib et al.

(2008) also reported bacterial strains from high altitude having extreme environment (above 4400 m), which showed arsenic resistance. Both these strains were Gram positive facultative anaerobic motile rods and had the ability to reduce the nitrate. They showed multiple metals resistances, which could be an important property to use these strains to perform better in polluted wastewater for its treatment. In another study, a metal resistant *Halomonas* was isolated, which tolerated multiple metals (chromate, cobalt, zinc, copper and cadmium) (Osman et al. 2010). Hong et al. (2010) also reported three strains of *Fusarium solani*, which not only showed resistance to copper but also resistant to zinc and this ability showed promising results for the treatment of zinc, copper and pyrene from polluted wastes. Both these strains (*B. cereus*-TA2 and *B. cereus*-TA4) showed growth over a wide range of pH and temperature but maximum growth was observed at 55°C at pH7.0.

The enzyme, which is responsible for Cr (VI) reduction in both these strains might be active at very high temperature. Eberly and Ely (2008) isolated a set of hydrogenases from thermophilic microorganisms, which showed their activity from 50 to 125°C and could be exploited for bioremediation, biosensors and for H_2S production. In this study, for both the strains, maximum Cr (VI) reduction occurred at 55°C and pH7.0. Opperman and Heerdan (2007) studied

aerobic conversion of carcinogenic hexavalent chromium in a complex organic medium by thermophilic *Thermus scotoeductus* SA-01. The temperature and pH for optimum Cr (VI) reduction was 80°C and 7.0, respectively. In both the strains, the reduction rate was not affected up to 250 µg mL⁻¹ of K₂CrO₄ but at higher initial Cr (VI) concentration up to 500 µg mL⁻¹, reduction was decreased significantly (64%). Chromate-resistant bacterial mechanisms have been reported either due to plasmids or by chromosomal encoded genes (Cervantes and Campos-Garcia 2007). The efflux of chromate from cytoplasm through membrane transporter in bacterial cells was encoded on the genes present on plasmids. The chromate reduction strategies like free-radical detoxifying activities, repairing of DNA damage, specific or unspecific Cr (VI) reduction are under the control of genes, which are present on chromosome (Pimentel et al. 2002).

The strains *B. cereus*-TA2 and *B. cereus*-TA4 also reduced selenite to elemental selenium aerobically at an initial selenite concentration of 200 µg mL⁻¹. But unlike chromate, maximum reduction was observed at 45°C in case of selenium. It showed that the selenite reductase enzymes/protein involved in selenite reduction worked optimally at 45°C. Others investigators have also studied the reduction of selenite to elemental selenium by bacteria such as *Azospira oryzae*, *Rhizobium sullae* and *Tetrathioabacter kashmirensis*.

CONCLUSION

Results showed that both these thermotolerant strains isolated from the hot spring of Hunza valley, Gilgit, Pakistan not only were resistant to a variety of metals and also gave promising results in the bioremediation of toxic chromate and selenite.

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