

## ASSESSMENT OF HYDROPHOBICITY AND ROUGHNESS OF STAINLESS STEEL ADHERED BY AN ISOLATE OF *BACILLUS CEREUS* FROM A DAIRY PLANT

Patrícia Campos Bernardes<sup>1</sup>, Nélio José de Andrade<sup>1\*</sup>, Sukarno Olavo Ferreira<sup>2</sup>, João Paulo Natalino de Sá<sup>1</sup>, Emiliane Andrade Araújo<sup>1</sup>, Deyse Maria Zanom Delatorre<sup>1</sup>, Livia Maria Pinheiro Luiz<sup>1</sup>

<sup>1</sup>Departamento de Tecnologia de Alimentos, Universidade Federal de Viçosa, Viçosa, MG, Brasil; <sup>2</sup>Departamento de Física, Universidade Federal de Viçosa, Viçosa, MG, Brasil.

Submitted: May 15, 2009; Returned to authors for corrections: April 04, 2010; Approved: April 26, 2010.

### ABSTRACT

The interaction between the surface of stainless steel and *Bacillus cereus* was studied in terms of the characteristics of interfacial interaction determined from the measurement of the contact angle of the surface of *B. cereus* and stainless steel in the presence or absence of *B. cereus* adherence. The microtopographies and the roughness of the surface of stainless steel and stainless steel adhered by *B. cereus* were evaluated with the help of atomic force microscopy and perfilometry. The strain of *B. cereus* studied was considered hydrophilic, whereas the stainless steel was considered hydrophobic. The adhesion was not thermodynamically favorable ( $\Delta G_{\text{adhesion}} > 0$ ) between the stainless steel and the strain of *B. cereus* studied. Thus, the interaction between them was not favored by the thermodynamic aspect of adhesion. There was no difference ( $p > 0.05$ ) in the roughness of the surfaces of stainless steel adhered by *B. cereus* when analyzed by atomic force microscope and perfilometry.

**Key words:** hydrophobicity, roughness, *Bacillus cereus*, dairy plant, adhesion.

### INTRODUCTION

*Bacillus cereus*, a spore forming bacterium, is an inevitable low-grade contaminant of a wide variety of foods, including cereals, food additives and processed milk products (11, 14). Due to high incidence in these products, some outbreaks have been reported in which milk or milk-related products containing *B. cereus* were suggested to be the cause of disease (2, 14). Also, *B. cereus* isolates has been observed in a high number of samples collected from food processing contact surfaces in the dairy industry (2). The occurrence of these microorganisms in pasteurized milk can be explained by the presence of their heat-resistance spores in raw milk or by

milk recontamination, due to inadequately cleaned and sanitized surfaces (3, 9, 13). The main consequence of *B. cereus* contamination in milk is the decreasing of shelf life and the occurrence of an off-flavor (9). *B. cereus* produces thermoresistant extracellular proteases and phospholipases that cause sweet coagulation and bitterness defects in milk.

Bacterial adhesion of *B. cereus* and other microorganisms to food processing contact surfaces is affected by the interaction of physicochemical characteristics of the microorganism surface and the contact surfaces. Specific linkages between microorganisms and food processing surfaces depend on the chemical composition of the both surfaces (17). Surfaces characteristics including cell surface hydrophobicity

\*Corresponding Author. Mailing address: Department of Food Technology, Federal University of Viçosa (UFV), Viçosa, MG, Brazil.; E-mail: [nandrade@ufv.br](mailto:nandrade@ufv.br)

relative surface charge have been reported to affect the adhesion of bacteria. The prediction of bacterial adhesion based on physicochemical factors was initially studied using the DLVO theory, first proposed by Derjaguin and Landau in 1941 and complemented by Verwey and Overbeek in 1948 (15). DLVO theory was proposed for liophobic colloidal particles, considering only long forces such as van der Waals forces. In 1994, van Oss and co-workers (19) proposed the extended DLVO theory that includes the influence of short forces such as Born repulsion forces, hydration forces, hydrophobic interactions and polymer bridges. However, it is important to mention that cellular structures such as flagella, fimbriae, and pili, and that the production of extracellular polysaccharides plays an important role in the adhesion process. In addition, environmental conditions such as pH, ionic forces, temperature, exposure time and cellular concentration of microorganisms strongly influence the adhesion process (6). Better understanding of the role of physicochemical properties in the adhesion process and biofilm formation on food processing contact surfaces can aid in the control of bacterial growth in environmental milk processing at dairy plants.

In the present study, the characteristics of interfacial tension of an isolated *B. cereus* strain from a dairy plant, a stainless steel surface and a stainless steel surface adhered by *B. cereus* were examined by measurements of contact angles between the surfaces and applying the data to the extended-DLVO theory. In addition, the roughness and the microtopography of stainless steel surfaces were evaluated in the presence or absence of *B. cereus* adherence.

## MATERIALS AND METHODS

### Microorganism and Surfaces

Studies were conducted using suspensions a *B. cereus* strain isolated from the surface of a stainless steel pasteurized milk packaging machine that was identified as Ribo 1 222 173 S4 . Surfaces of coupons (10 mm × 10 mm × 0.5 mm) of stainless steel AISI 304, #4, in the presence or absence of *B. cereus* adherence, were used for bacterial adherence studies.

One milliliter of the culture was stored at -80°C in nutrient broth (Merck, Sao Paulo, Brazil) containing glycerol (80:20). A working culture of the strain containing approximately 10<sup>7</sup> cfu/mL was prepared by inoculation of 100 µL of frozen culture into 10 mL of Brain Hearth Infusion (BHI, Merck, São Paulo, Brazil), followed by incubation at 32°C for 24 h. The culture was sub-cultured two times before use. The number of microorganisms in each suspension was obtained by total count using Plate Count Agar (PCA, Merck, São Paulo, Brazil) at 32°C for 24 h.

### Attachment of Cells

Coupons were first cleaned by washing them with neutral liquid detergent and water, followed by rinsing with distilled water and immersion in 70% ethyl alcohol for 1 h to remove fat. Subsequently, coupons were rinsed with distilled water, air-dried, and sterilized at 121°C for 15 min (12). The cleaned and sanitized coupons were added to 250 mL flasks containing 100 mL of nutrient broth, which were previously inoculated with suspensions of *B. cereus*. The initial number of cells was approximately 10<sup>3</sup> cfu/mL, and the flasks were incubated statically for different times and at different temperatures according to the full factorial at two levels (Table 1).

At the proper times, coupons were removed and rinsed for 1 min in tubes containing 10 mL of sterilized 0.1 % peptone water to remove the planktonic cells. Afterward, each coupon was placed into Petri plates and analyzed for contact angle and roughness.

**Table 1.** Temperature and time of adherence of *Bacillus cereus* on the surface of stainless steel.

Experiments	Time (days)	Temperature (°C)
E1	1	4°C
E2	1	10°C
E3	1	25°C
E4	1	35°C
E5	5.5	7°C
E6	5.5	15°C
E7	5.5	20°C
E8	5.5	30°C
E9	10	4°C
E10	10	10°C
E11	10	25°C
E12	10	35°C

### Contact Angle Measurement

For stainless steel AISI 304, #4 and stainless steel AISI 304 # 4 adhered by *B. cereus*, the contact angles between the surfaces and water, formamide, and  $\alpha$ -bromonaphthalene were determined using a goniometer (Kruss, Germany). Measurements of the contact angle of one 2.0  $\mu$ L drop were taken each second for 30 s for all liquids and surfaces. Experiments were conducted in triplicate. The contact angle for the microorganism surface was determined on a layer of vegetative cells using the drop method (5). First, the strain of *B. cereus* was activated twice in BHI broth (Merck, Sao Paulo, Brazil), creating suspensions with approximately  $10^7$  CFU/mL. The suspensions were centrifuged at 12.000 g for 10 min and washed three times in 0.1 M phosphate buffered saline (PBS). The cell pellet was suspended in the same buffer, and the suspension was filtered using an acetate cellulose membrane (0.45  $\mu$ m pore size, 27 mm in diameter) using negative pressure. During the filtration, 30 mL of pure water (Milli-Q) were added. The membranes were transferred to Petri plates containing 1 % (v/v) agar and 10 % (v/v) glycerol. The membranes were cut into three parts for contact angle measurements with water, formamide, and  $\alpha$ -bromonaphthalene.

**Table 2.** Interfacial tension components of liquids at 25 °C.

Liquid	Interfacial tension (mJ/m <sup>2</sup> )			
	$\gamma_1^{Tot}$	$\gamma_1^{LW}$	$\gamma_1^+$	$\gamma_1^-$
$\alpha$ -Bromonaphthalene	44.4	44.4	0.0	0.0
Water	72.8	21.8	25.5	25.5
Formamide	58.0	39.0	2.28	39,6

Source: (18)

The interfacial tension is equal to the sum of the two components ( $\gamma_s^{LW}$  and  $\gamma_s^{AB}$ ):

$$a) \gamma_s^{LW} = 11.1(1 + \cos \theta_B)^2 \text{ Equation (2)}$$

$$b) \gamma_s^{AB} = 2\sqrt{\gamma_s^+ \gamma_s^-} \text{ Equation (3)}$$

$$c) \gamma_s^{tot} = \gamma_s^{LW} + \gamma_s^{AB} \text{ Equation (4)}$$

### Determination of the Total Interfacial Tension ( $\gamma^{tot}$ )

The equation of Young-Good-Girifalco-Fowkes (Equation 1) relates the contact angle formed by the liquid above a solid surface with the components of interfacial tension of liquids ( $\gamma_1^{LW}$ ,  $\gamma_1^+$ ,  $\gamma_1^-$ ) and of the surface ( $\gamma_s^{LW}$ ,  $\gamma_s^+$ ,  $\gamma_s^-$ ) as follows:

$$(1 + \cos \theta_B) \gamma_l^{TOT} = 2(\sqrt{\gamma_s^{LW} \gamma_l^{LW}} + \sqrt{\gamma_s^+ \gamma_l^-} + \sqrt{\gamma_s^- \gamma_l^+}) \text{ Equation (1)}$$

where  $\gamma^{tot}$  is the total interfacial tension of the surface,  $\gamma^{LW}$  is the interfacial tension of the interactions of the Lifshitz-van der Waals forces,  $\gamma^{AB}$  is the polar component of the Lewis acid-base interaction,  $\gamma^+$  is the interfacial tension of the electron acceptor component of the acid-base component,  $\gamma^-$  is the interfacial tension of electron donor component of the acid-base component,  $\theta_B$  is the contact angle obtained with  $\alpha$ -bromonaphthalene and *s* and *l* indicate surface and liquid, respectively (19).

The three components of the interfacial tension of the surfaces were determined from the contact angles obtained from three liquids with different polarities, whose interfacial tensions are known as shown in Table 2.

### Total Free Energy of Interaction ( $\Delta G_{sws}^{TOT}$ )

The total free energy of interaction among molecules of the surface(s) immersed in water (w) is determined by the sum of the apolar and polar free energies of interaction,  $\Delta G_{sws}^{LW}$  and  $\Delta G_{sws}^{AB}$ , respectively.

$$\Delta G_{sws}^{TOT} = \Delta G_{sws}^{LW} + \Delta G_{sws}^{AB} \text{ Equation (5)}$$

$$\Delta G_{sws}^{LW} = -2x \sqrt{\gamma_s^{LW} - \gamma_w^{LW}} \text{ Equation (6)}$$

$$\Delta G_{sws}^{AB} = -4 \left( \sqrt{\gamma_s^+ \gamma_s^-} + \sqrt{\gamma_w^+ \gamma_w^-} - \sqrt{\gamma_s^+ \gamma_w^-} - \sqrt{\gamma_w^+ \gamma_s^-} \right)$$

Equation (7)

When  $\Delta G_{sws}^{TOT} > 0$ , the surface is considered hydrophilic. Conversely, if  $\Delta G_{sws}^{TOT} < 0$ , the surface is considered hydrophobic.

Determination of the Total Free Energy of Adhesion ( $\Delta G_{adhesion}$ )

From the values of the components of the interfacial tensions, it is possible to determine the  $\Delta G_{adhesion}$  between two surfaces (microbial cells (b) and food processing surfaces (s)):

$$\gamma_{bs} = \gamma_{bs}^{LW} + \gamma_{bs}^{AB} \quad \text{Equation (8)}$$

$$\gamma_{bs}^{LW} = \gamma_b^{LW} + \gamma_s^{LW} - 2\sqrt{\gamma_b^{LW} \gamma_s^{LW}} \quad \text{Equation (9)}$$

$$\gamma_{bs}^{AB} = 2 \left( \sqrt{\gamma_b^+ \gamma_b^-} + \sqrt{\gamma_s^+ \gamma_s^-} - \sqrt{\gamma_b^+ \gamma_s^-} - \sqrt{\gamma_b^- \gamma_s^+} \right)$$

Equation (10)

When free energy is related to the interfacial tension, then  $\Delta G_{adhesion}$  can be represented by the following:

$$\Delta G_{adhesion} = \Delta G_{bls}^{LW} + \Delta G_{bls}^{AB} \quad \text{Equation (11)}$$

$$\Delta G_{bls}^{LW} = \gamma_{bs}^{LW} - \gamma_{bl}^{LW} - \gamma_{sl}^{LW} \quad \text{Equation (12)}$$

$$\Delta G_{bls}^{AB} = \gamma_{bs}^{AB} - \gamma_{bl}^{AB} - \gamma_{sl}^{AB} \quad \text{Equation (13)}$$

where  $\gamma_{bs}$  is the interfacial tension between the bacterial surfaces and the adhesion surface,  $\gamma_{bl}$  is the interfacial tension between the bacterial surfaces and the liquid, and  $\gamma_{sl}$  is the interfacial tension between the adhesion surfaces and the liquid. The  $\Delta G_{adhesion}$  values allow for evaluation of the thermodynamics of the adhesion process: if  $\Delta G_{adhesion} < 0$ , the process is favorable, but if  $\Delta G_{adhesion} > 0$ , the process is unfavorable.

### Surface Roughness

The microtopography of the stainless surface in the

presence or absence of *B. cereus* adherence was evaluated using atomic force microscopy (AFM) that analyzed areas of 100  $\mu\text{m}^2$  and 10  $\mu\text{m}^2$  (Universal SPM System Ntegra Prima/NT-MDT) and using a Perthometer (Ambios Technology, XP1) that analyzed one line of 1 mm in each coupon. The experiment was conducted with three repetitions. The roughness of the surfaces was compared before and after bacterial adhesion by different numbers of *B. cereus* by regression analysis at 5% of probability. The means of the roughness were submitted to Tukey's test ( $\alpha = 0.05$ ) by using the Statistical Analysis System (SAS), version 9.1.

## RESULTS AND DISCUSSION

### Adhesion of *B. cereus*

The numbers (log cfu/cm<sup>2</sup>) of *Bacillus cereus* cells adhered to stainless steel AISI 304 #4 in different experiments are shown in Table 3.

**Table 3.** Number (log cfu/cm<sup>2</sup>) of *Bacillus cereus* cells adhered to stainless steel AISI 304 #4 in different experiments.

Experiments	Log cfu/cm <sup>2</sup> *
E1 (4 °C/1 d)	0.91 ± 0.66
E2 (10 °C/1 d)	1.40 ± 1.00
E3 (25 °C/1 d)	3.21 ± 0.67
E4 (35 °C/1 d)	4.01 ± 0.80
E5 (7 °C/5.5 d)	0.50 ± 0.42
E6 (15 °C/5.5 d)	3.32 ± 0.43
E7 (20 °C/5.5 d)	3.15 ± 0.21
E8 (30 °C/5.5 d)	3.66 ± 0.61
E9 (4 °C/10 d)	0.31 ± 0.61
E10 (10 °C/10 d)	1.16 ± 0.95
E11 (25 °C/10 d)	3.43 ± 0.81
E12 (35 °C/10 d)	4.43 ± 0.77

\*Mean of three repetitions.

### Analyses of the Contact Angles

The contact angle with water ( $\theta_w$ ) measured for *B. cereus* was lower than 50 °, (Table 4) indicating that it is a hydrophilic surface according to the classification system proposed by Azeredo (1). According to this author, surfaces with a  $\theta_w$  less than 50 ° are classified as hydrophilic, whereas surfaces with a

$\theta_w$  greater than  $50^\circ$  are classified as hydrophobic. The stainless steel surfaces adhered by *B. cereus* at  $20^\circ\text{C}/5.5$  d and  $10^\circ\text{C}/10$  d with  $\theta_w$  values lower than  $50^\circ$  also were considered hydrophilic. The surfaces were evaluated in others conditions and were classified as hydrophobic. The stainless steel surface without cells adhered was hydrophobic. The contact angle with water is a qualitative criterion used to classify the hydrophobicity of food processing or microorganism surfaces. Faille *et al.* (7) found contact angles ( $\theta_w$ ) for stainless steel of  $75^\circ$ , confirming the hydrophobic characteristics of this material.

### Thermodynamic Parameters of the Surfaces

The contact angle measurements for the three substances were used to calculate the components of interfacial tension and levels of hydrophobicity (Table 5). It is possible to estimate the hydrophilic or hydrophobic character of the surfaces by components of interfacial tension. With increasing  $\gamma^{LW}$  values, the apolarity of a surface increases, which results in lower affinity of that surface for polar liquids. A high  $\gamma^{AB}$  component value means more water of hydration on the surface and increased hydrophilicity. According to these criteria, *B. cereus* surfaces and stainless steel surfaces adhered by *B. cereus* at  $10^\circ\text{C}/10$  d are considered to hydrophilic because their  $\gamma^{AB}$  values are higher than those of the other surfaces. This result does not agree with those obtained by contact angle measurement with water ( $\theta_w$ ) that also classified as hydrophilic

the surfaces of stainless steel adhered by *B. cereus* at  $20^\circ\text{C}/5.5$  d.

The component  $\gamma^-$  can also be a semi-quantitative measure of hydrophobicity  $\gamma^-$  values  $\leq 25.5$  mJ/m<sup>2</sup> indicate a hydrophobic surface regardless of the value of the apolar component. The  $\gamma^-$  values between 25 mJ/m<sup>2</sup> and 35 mJ/m<sup>2</sup> suggest that the hydrophobicity is dependent upon the apolar component. In these cases, the surfaces are hydrophilic when  $\gamma^{LW} \leq 45$  mJ/m<sup>2</sup> and hydrophobic when  $\gamma^{LW} \geq 46$  mJ/m<sup>2</sup> (1). According to the results, *B. cereus* is an electron donor because the  $\gamma^-$  values are higher than the  $\gamma^+$  values. Strevett and Chen (15) demonstrated that  $\gamma^-$  is always higher than  $\gamma^+$  for *E. coli*, *P. fluorescens*, *B. subtilis* and *P. aeruginosa*, confirming the electron donor characteristics of these bacterial cells. All biosurfaces are predominantly electron donors because of the presence of oxygen in the atmosphere and the hydration of microbial cells (18). Stainless steel surfaces not adhered by *B. cereus* also act as electron donors. Similar to our findings, Chaves (6) observed that the surface of stainless steel AISI 316 predominately acts as an electron, similar to most solid surfaces. Furthermore, *B. cereus* surfaces and stainless steel surfaces adhered by cells of *B. cereus* at  $20^\circ\text{C}/5.5$  d are considered hydrophilic because they did not present  $\gamma^- \leq 25.5$  mJ/m<sup>2</sup>. The other surfaces, including stainless steel surfaces adhered by *B. cereus* at  $10^\circ\text{C}/10$  d, are considered hydrophobic.

**Table 4.** Measurements of the contact angles ( $\theta$ ) of the cells of *Bacillus cereus* and stainless steel in the presence or absence of *Bacillus cereus* adherence with water ( $\theta_A$ ), formamide ( $\theta_F$ ) and  $\alpha$ -bromonaphthalene ( $\theta_B$ ).

	Contact angles ( $^\circ$ )*		
	$\theta_w$	$\theta_F$	$\theta_B$
Stainless steel	70.77 $\pm$ 7.9	53.36 $\pm$ 9.4	28.1 $\pm$ 3.1
<i>B. cereus</i>	24.52 $\pm$ 2.8	15.37 $\pm$ 1.3	45.17 $\pm$ 0.5
E1 (4 $^\circ\text{C}/1$ d)	72.1 $\pm$ 2.6	56.1 $\pm$ 8.6	34.6 $\pm$ 6.7
E2 (10 $^\circ\text{C}/1$ d)	73.2 $\pm$ 0.5	57.0 $\pm$ 7.2	30.5 $\pm$ 5.9
E3 (25 $^\circ\text{C}/1$ d)	57.3 $\pm$ 11.9	57.5 $\pm$ 6.7	25.3 $\pm$ 6.7
E4 (35 $^\circ\text{C}/1$ d)	60.1 $\pm$ 10.5	40.5 $\pm$ 1.5	32.0 $\pm$ 3.5
E5 (7 $^\circ\text{C}/5.5$ d)	69.6 $\pm$ 1.4	52.2 $\pm$ 4.2	35.6 $\pm$ 4.7
E6 (15 $^\circ\text{C}/5.5$ d)	56.7 $\pm$ 10.7	56.7 $\pm$ 7.0	32.2 $\pm$ 4.3
E7 (20 $^\circ\text{C}/5.5$ d)	49.3 $\pm$ 6.1	44.2 $\pm$ 2.6	39.0 $\pm$ 7.3
E8 (30 $^\circ\text{C}/5.5$ d)	73.7 $\pm$ 5.0	53.5 $\pm$ 8.3	34.5 $\pm$ 0.2
E9 (4 $^\circ\text{C}/10$ d)	77.6 $\pm$ 8.9	57.6 $\pm$ 6.0	33.6 $\pm$ 6.7
E10 (10 $^\circ\text{C}/10$ d)	48.9 $\pm$ 14.8	19.2 $\pm$ 3.2	34.2 $\pm$ 3.3
E11 (25 $^\circ\text{C}/10$ d)	62.8 $\pm$ 8.2	43.2 $\pm$ 6.0	31.5 $\pm$ 1.6
E12 (35 $^\circ\text{C}/10$ d)	58.9 $\pm$ 12.7	39.4 $\pm$ 11.9	32.8 $\pm$ 6.3

\* Mean of three repetitions.

**Table 5.** Values of the interfacial tension components ( $\gamma^{LW}$ ,  $\gamma^+$ ,  $\gamma^-$ ,  $\gamma^{AB}$ ,  $\gamma^{TOT}$ ) of the cells of the *Bacillus cereus*, stainless steel and stainless steel adhered by cells of *B. cereus*.

Surfaces	Interfacial tension (mJ/m <sup>2</sup> )				
	$\gamma^{LW}$	$\gamma^+$	$\gamma^-$	$\gamma^{AB}$	$\gamma^{TOT}$
Stainless steel	39.3206	0.0899	6.3333	1.5091	40.8297
<i>B. cereus</i>	32.2682	3.2454	45.2929	24.2482	56.5164
E1 (4 °C/1 d)	36.8944	0.0956	12.2929	2.1681	39.0625
E2 (10 °C/1 d)	38.4688	0.0207	11.6946	0.9840	39.4528
E3 (25 °C/1 d)	40.2433	0.3842	0.4922	0.8697	41.1130
E4 (35 °C/1 d)	37.9096	1.0378	16.8009	8.3510	46.2606
E5 (7 °C/5.5 d)	36.4894	0.3478	12.4987	4.1699	40.6593
E6(15 °C/5.5 d)	37.8335	0.1378	7.6388	2.0519	39.8854
E7(20 °C/5.5 d)	35.0565	0.4573	32.1712	7.6712	42.7277
E8(30 °C/5.5 d)	36.9345	0.3508	9.1703	3.5871	40.5216
E9 (4 °C/10 d)	37.2915	0.1260	7.6212	1.9598	39.2513
E10(10 °C/10 d)	37.0542	3.3996	20.0995	16.5324	53.5866
E11(25 °C/10 d)	38.0982	0.8236	15.1795	7.0715	45.1697
E12(35 °C/10 d)	37.6033	1.1625	17.5686	9.0384	46.6417

**Total Free Energy of Interaction ( $\Delta G_{sws}^{TOT}$ )**

According to the quantitative criteria, the *B. cereus* strain and the stainless steel surfaces adhered by *B. cereus* at 20 °C/5.5 d were considered hydrophilic because the  $\Delta G_{sws}^{TOT}$  values for both surfaces were  $> 0$  (Table 6). The other surfaces were hydrophobic ( $\Delta G_{sws}^{TOT} < 0$ ). Similar results were found using the semiquantitative  $\gamma^-$  value. However, these results are different from those determined by qualitative criteria using contact angles with water ( $\theta_w$ ), which considered stainless steel surfaces adhered by *B. cereus* at 10 °C/10 d as hydrophilic. The  $\Delta G_{sws}^{LW}$  values are generally negative in the bacterial interactions, indicating the Lifshitz van der Waals forces are predominantly attractive, whereas the  $\Delta G_{sws}^{AB}$  can be positive or negative, indicating repulsion or attraction, respectively (4). In our experiment,  $\Delta G_{sws}^{AB}$  values indicated attraction for most of the surfaces evaluated (Table 6). Only the surfaces considered hydrophilic showed  $\Delta G_{sws}^{AB} > 0$ , demonstrating that hydrophobicity is predominantly determined by polar forces of attraction. The component  $\Delta G_{sws}^{AB}$  represents the hydration degree of surfaces, which means that high  $\Delta G_{sws}^{AB}$  values equal low hydrophobicity of the surfaces. Therefore, the qualitative and quantitative criteria used to evaluate hydrophobicity showed a slight divergence from the results of the hydrophobicity of the surfaces of the stainless steel adhered

by *B. cereus* at 10 °C/10 d (Tables 4 and 6). For the other evaluations, the results for both criteria were in agreement. The *B. cereus* surface was classified as hydrophilic and stainless steel surfaces were classified as hydrophobic. A possible explanation for the hydrophilicity of stainless steel surfaces adhered by microorganisms as classified by quantitative criteria is the random distribution of the drops of water that were placed on the surfaces to measure the contact angles. These drops can be placed in regions with different concentrations of adhered cells. Therefore, the contact angles measurements can be related to the hydrophobicity of the stainless steel surfaces or the cells, reflecting the characteristics of only one surface.

**Global Free Energy of Adhesion ( $\Delta G_{adhesion}$ )**

According to the thermodynamic theory of adhesion, if attractive forces are higher than repulsive forces, the interaction of short reach plays an important role in bacterial adhesion to surfaces. Such forces include polar and apolar interactions. Bacterial adhesion is favorable if the interactions lead to a decrease in the free energy of adhesion ( $\Delta G_{adhesion} < 0$ ) (6). In our experiment, the free energy of adhesion between stainless steel and *B. cereus* was positive ( $\Delta G_{adhesion} > 0$ ) (Table 7), being thermodynamically unfavorable. This finding was

similar those observed by Teixeira *et al.* (16), in which strains of *Pseudomonas aeruginosa* and *Staphylococcus sciuri* isolated from a milk machine were hydrophilic, the stainless steel surface was hydrophobic and the free energy of adhesion was positive. According to van Oss (20), it is well known that

bacterial adhesion in an aqueous solution is favorable between hydrophobic surfaces, which can remove the water among them. However, it should be emphasized that adhesion between hydrophobic and hydrophilic surfaces or two hydrophilic surfaces can occur.

**Table 6.** Values of the apolar ( $\Delta G_{\text{sws}}^{\text{LW}}$ ) and polar ( $\Delta G_{\text{sws}}^{\text{AB}}$ ) components of the total free energy of interaction ( $\Delta G_{\text{sws}}^{\text{TOT}}$ ) of *B. cereus* surfaces and stainless steel surfaces in the presence or absence of *B. cereus* adherence.

	Total free energy of interaction (mJ/m <sup>2</sup> )		
	$\Delta G_{\text{sws}}^{\text{LW}}$	$\Delta G_{\text{sws}}^{\text{AB}}$	$\Delta G_{\text{sws}}^{\text{TOT}}$
Stainless steel	-5.1300	-23.6028	-28.7328
<i>B. cereus</i>	-2.0468	21.8664	19.8195
E1 (4 °C/1 d)	-3.9462	-29.2732	-33.2194
E2(10 °C/1 d)	-4.7024	-31.9868	-36.6892
E3(25 °C/1 d)	-5.6075	-77.0592	-82.6668
E4(35 °C/1 d)	-4.4238	-15.3336	-19.7574
E5(7 °C/5.5 d)	-3.7673	-27.0336	-30.8009
E6(15 °C/5.5 d)	-4.3900	-42.7792	-47.1692
E7(20 °C/5.5 d)	-3.1313	10.8848	7.7535
E8(30 °C/5.5 d)	-3.9668	-36.0436	-40.0104
E9(4 °C/10 d)	-4.1335	-42.9876	-47.1211
E10(10 °C/10 d)	-4.0224	-7.2648	-11.2872
E11(25 °C/10 d)	-4.5199	-19.1152	-23.6351
E12(35 °C/10 d)	-4.2813	-13.6348	-17.9161

**Table 7.** Global free energy of adhesion values ( $\Delta G_{\text{adhesion}}$ ) between *Bacillus cereus* (b) and the stainless steel surfaces of AISI 304 (s) in aqueous liquid media (l) and their apolar ( $\Delta G_{\text{bls}}^{\text{LW}}$ ) and polar ( $\Delta G_{\text{bls}}^{\text{AB}}$ ) components.

	Global free energy of adhesion (mJ/m <sup>2</sup> )		
	$\Delta G_{\text{bls}}^{\text{LW}}$	$\Delta G_{\text{bls}}^{\text{AB}}$	$\Delta G_{\text{adhesion}}$
<i>B. cereus</i>	-3.3353	5.8849	2.5496

Despite that the strain studied in this experiment is considered hydrophilic, some strains of *Bacillus* can produce highly hydrophobic spores that are able to adhere strongly to stainless steel, a surface recognized as hydrophobic. After adhesion, the process of spore germination can occur and vegetative cells may colonize the surfaces (10).

### Roughness and Microtopography of the Surfaces

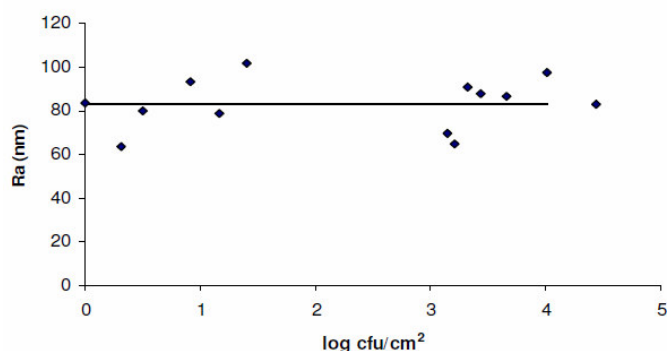
There is no difference ( $p \geq 0.05$ ) in the roughness of the surfaces analyzed by AFM and profilometry (Figures 1 and 2). The increase of the number of adhered cells to surface coupons of stainless steel did not lead to an increase in surface

roughness. These results can be explained due to the low number of cells that adhered to the stainless steel in the different experiments, which reached approximately  $10^4$  cfu/cm<sup>2</sup>. Thus, the measurements of roughness by AFM and profilometry reflected the average roughness of the surface of the stainless steel.

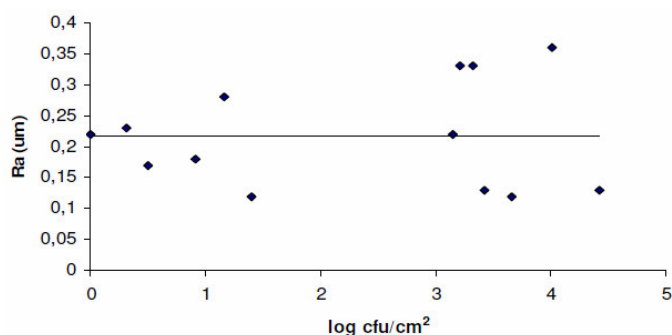
According to Flint *et al.* (8), the relation between roughness and bacterial adhesion is not clear. This divergence is probably related to the degree of roughness studied, the bacterial strains tested, physicochemical characteristics of the surfaces and the method for detecting bacteria. Flint *et al.* (8) did not find a relationship between thermoresistant

streptococcus adhesion to stainless steel and the roughness of surfaces. However, it was observed that maximum adhesion occurred with mean roughness values close to mean length of the bacteria (0.9  $\mu\text{m}$ ). The authors suggested that some surface irregularities might provide protection to entrap cells in the cracks and crevices of the surface.

In 1 of 14 surfaces studied, a difference was observed between the qualitative criterion (contact angle measurement with water) and quantitative criterion ( $\Delta G_{\text{sws}}^{\text{TOT}}$  - total free energy of interaction) to evaluate the hydrophobicity of the surfaces. The free energy of adhesion between stainless steel AISI 304 #4 and *B. cereus* was positive ( $\Delta G_{\text{adhesion}} > 0$ ), being therefore thermodynamically unfavorable. Thus, an interaction between them was not favorable according to the thermodynamic aspect of adhesion. There was no difference ( $p \geq 0.05$ ) in the roughness of the surfaces analyzed by AFM and profilometry.



**Figure 1.** Means roughness (Ra) of the stainless steel adhered with different number of cells of *Bacillus cereus* as evaluated by AFM.



**Figure 2.** Means roughness (Ra) of the stainless steel adhered with different number of cells of *Bacillus cereus* as evaluated by Perthometer.

## REFERENCES

- Azeredo, J. (1998). *Adesão de microrganismos e composição da matriz de bioagregados desenvolvimento de técnicas e estudo da influência de exopolímeros*. Braga, Portugal, 231p. (D. Sc. Thesis. Departamento de Engenharia Biológica. Uminho).
- Bartoszewicza, M.; Hansen, B.M.; Swiecicka, I. (2008). The members of the *Bacillus cereus* group are commonly present contaminants of fresh and heat-treated milk. *Food Microbiol.* 25, 588–596.
- Boor, K.J. (2001). ADSA foundation scholar award – fluid dairy product quality and safety: looking to the future. *J. Dairy Sci.* 84, 1–11.
- Bos, R.; Van Der Mei, H.C.; Busscher, H.J. (1999). Physico-chemistry of initial microbial adhesive interactions - its mechanisms and methods for study. *FEMS Microbiol. Rev.* 23, 79-230.
- Busscher, H.J.; Weerkamp, A.H.; Van Der Mei, H.C.; Van Pelt, A.W.; Jong, H.P.; Arends, J. (1984). Measurement of the surface free energy of bacterial cell surface and its relevance for adhesion. *Appl. Environ. Microbiol.* 48 (5), 980-983.
- Chaves, L.C.D. (2004). *Estudo da cinética da formação de biofilmes em superfícies em contato com água potável*. Braga, Portugal, 156 p. (M. Sc. Dissertation. Departamento de Engenharia Biológica. Uminho).
- Faillie, C.; Fontaine, F.; Bénézec, T. (2001). Potencial occurrence of adhering living *Bacillus* spores in milk product processing lines. *J. Appl. Microbiol.* 90, 892-900.
- Flint, S.H.; Brooks, J.D.; Bremer, P.J. (2000). Properties of the stainless steel substrate, influencing the adhesion of thermo-resistant streptococci. *J. Food Eng.* 43, 235-242.
- Fromm, H.I.; Boor, K.J. (2004). Characterization of pasteurized fluid milk shelf-life attributes. *J. Food Sci.* 69 (8), 207-214.
- Hüsmark, U.; Rönner, U. (1992). The influence of hydrophobic, electrostatic and morphologic properties on the adhesion of *Bacillus* spores. *Biofoul.* 5, 335-344.
- Larsen, H.D.; Jørgensen, K. (1997). The occurrence of *Bacillus cereus* in Danish pasteurized milk. *Int. J. Food Microbiol.* 34, 179–186.
- Parizzi, S.Q.F.; Andrade, N.J.; Soares, N.F.F.; Silva, C.A.S.; Monteiro, E.A.M. (2004). Bacterial adherence to different inert surfaces evaluated by epifluorescence microscopy and plate count method. *Braz. Arch. Biol. Technol.* 47 (1), 77-83.
- Peng, J.S.; Tsai, W.C.; Chou, C.C. (2002). Inactivation and removal of *Bacillus cereus* by sanitizer and detergent. *Int. J. Food Microbiol.* 77, 11–18.
- Reyes, J.E.; Bastías, J.M.; Gutiérrez, M.R.; Rodríguez, M.O. (2007). Prevalence of *Bacillus cereus* in dried milk products used by Chilean School Feeding Program. *Food Microbiol.* 24, 1-6.
- Strevett, K.A.; Chen, G. (2003). Microbial surface thermodynamics and applications. *Res. Microbiol.* 154, 329-335.
- Teixeira, P.; Lopes, Z.; Azeredo, J.; Oliveira, R.; Vieira, M.J. (2005). Physico-chemical surface characterization of a bacterial population isolated from milking machine. *Food Microbiol.* 22, 247-251.



17. Valcarce, M.B.; Busalmen, S.R.; Sánchez, S.R. (2002). The influence of the surface condition on the adhesion of *Pseudomonas fluorescens* (ATCC 17552) to copper and aluminium brass. *Int. Biodeter. Biodeg.* 50, 61-66.
18. Van Der Mei, H.C.; Bos, R.; Busscher, H.J. (1998). A reference guide to microbial cell surface hydrophobicity based on contact angles. *Coll. Surfac.* 11, 213-221.
19. Van Oss, C.J. (1994). *Interfacial Forces in Aqueous Media*. Marcel Dekker Inc, New York, N.Y.
20. Van Oss, C.J. (1997). Hydrophobicity and hydrophilicity of biosurfactants. *Curr. Opin. Coll. Int. Sci.* 2, 503-512.