Amino acids and antioxidant activities of extracts from wild edible mushrooms from a community forest in the Nasrinual District, Maha Sarakham, Thailand

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Abstract

Twenty two species of wild edible mushrooms were collected from Nasrinual community forest. The levels of 15 amino acids, antioxidant activity, and γ -aminobutyric acid (GABA) were determined. The 8 essential amino acids were arginine, histidine, isoleucine, leucine, lysine, phenylalanine, threonine, and valine, that they were found in all the mushrooms at concentration ranges of 0.28-2.14, 0.11-0.8, 0.19-1.33, 0.33-2.22, 0.22-1.73, 0.18-1.09, 0.31-1.83, and 0.23-1.66 g/100 g dw, respectively. The greatest and smallest amount of total amino acids, 28.41 and 5.76 g/100 g dw, were found in Hed-Peung-No5 and Hed-Klum-Ma, respectively. The antioxidants from the mushroom extracts obtained by water and 60% methanol and their activities were evaluated. It was found that flavonoids content and DPPH in methanol extracts were higher than in water extracts. In contrast, the phenolic content and FRAP activity in water extracts were higher than those in methanol extracts. Total phenolic, FRAP, and flavonoids content was higher in the extracts obtained by hot water than it was in the extracts from Hed-Peung-No3. The GABA content was higher in the extract of Hed-Sai-duan (23.304 ± 0.027 mg/g dw).

Keywords: DPPH; total phenolic compounds; flavonoids; FRAP; GABA.

Practical Application: Wild edible mushrooms were collected from Nasrinual community forest, and identified by morphology. Amino acids were analyzed from 22 mushroom species by HPLC. Type of extraction solvent (water and methanol) had an influence on the level and type of antioxidant activities including GABA content.

1 Introduction

In addition to sources of various aromatic flavors, including umami, mushrooms are also healthy foods. They are rich in protein, vitamins, and minerals and low in calories and fat (Akindahunsi & Oyetayo, 2006). Some kinds of mushrooms have good flavor, such as shitake and truffles. The typical mushroom aroma comes from C₂ volatile compounds, namely, oct-1-en-3-ol, oct-1-en-3-on, octan-3-ol, and octan-3-on (Moliszewska, 2014). Mushrooms have been used as foods and food flavorings for a long time. Furthermore, some mushrooms produce secondary metabolites, such as lovastatin, γ-aminobutyric acid (GABA), and ergothioneine (Chen et al., 2012). Lovastatin, mostly produced by Monascus species, inhibits the enzyme necessary for cholesterol production, thereby reducing LDL cholesterol levels and the risk for developing coronary heart disease (Aaron et al., 2007). GABA is often used to treat blood pressure, stress and anxiety and to improve sleep, as well as to stimulate the body's natural growth hormone. Ergothioneine is an amino acid (the derivative of histidine) that has excellent antioxidant properties and is good for health (Cheah & Halliwell, 2012). Furthermore, mushrooms are also sources of beneficial bioactive substances, such as β -glucan, flavonoids, glycoproteins, sesquiterpenes, carotene, and phenolic compounds (Reis et al., 2011). These secondary metabolites are effective antioxidants that enhance human health.

The amino acid level is an indicator of the nutritional value and flavor of food. The types of amino acids that influence the taste of mushrooms include alanine, glycine, serine, and threonine, which are sweet amino acids, and arginine, leucine, isoleucine, histidine, methionine, phenylalanine, and valine, which are bitter amino acids. The tasteless amino acids are lysine and tyrosine (Kalač, 2016; Pomeranz, 2012). Mushrooms constitute a source of nine essential amino acids (histidine, isoleucine, leucine, lysine, methionine, phenylalanine, threonine, tryptophan, and valine) that humans cannot synthesize. These nine amino acids are also found in the oyster mushroom species *Pleurotus ostreatus* and *P. sajor-caju*, which have amino acids totaling 21.10 and 20.12 mg/g fresh weight, respectively (Chirinang & Intarapichet, 2009). Leucine, threonine and valine are the three most abundant essential amino acids.

Due to mushrooms are rich of protein, antioxidant and also good taste and flavor profile. They can be used in meat product to replacement animal protein such as smoke sausages with a desirable nutritional properties and acceptable sensory (Nagy et al., 2017). However wild edible mushrooms that are abundant only in rainy season. In order to preserve for long time storage, gamma irradiation was applied to dried and sliced sample of *Boletus edulis*. It was found that nutritional, chemical compositions and antioxidants of the slide mushroom slightly decreased for long storage to 12 months (Fernandes et al., 2017). One reported

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(Falandysz et al., 2019) investigation on contamination with mercury of a raw and stir-fried mushrooms (Boletaceae Family). All mushrooms that grown on polymetallic soil (accumulate Hg in elevated concentration) were collected from the region of west-central Yunnan, China. The results showed that if taken fried mushroom daily over a week period (100g x 7) could provide Hg in dose from 1.9 to 9.6 μ g/kg bm (body mass). The consumption of fried mushrooms could provide Hg in the safe dose. The accumulation radiocesium of two species mushroom from Chernobyl fallout contaminated area by Chiaravalle et al. (2018) was observed. They found that *Boletus edulis* was more contaminated with cesium-137 than *Amanita caesarea*. However the consumption of both mushrooms proved to be non-dangerous for humans.

The aim of this study was to collect and identify wild edible mushrooms from the Nasrinual community forest. The amounts of amino acids from dried mushrooms were evaluated, and the antioxidant activity in the water and methanol extracts of these wild mushrooms was determined by their phenolic content, their ability to scavenge DPPH free radicals, and the amounts of FRAP, GABA, and flavonoids they contained.

2 Material and methods

2.1 Mushroom collection and extraction

Wild mushrooms were collected from the forest in the Nasrinual District, Maha Sarakham, Thailand. All the samples were kept in cooler with ice and transported to the laboratory. The morphological features of the mushrooms were studied and identified. Each type of fruiting body was individually dried at 50 °C, ground into powder and kept in plastic zip-locked bags. The stored mushroom powder was kept in a refrigerator until used in experiments. A total of 0.1 g of dried mushroom was used to create an extract in 5 mL water and an extract in 60% methanol by sonication for 1 h. The clear supernatant was generated by centrifugation at 10,000 g and kept at -20 °C until analyzed.

2.2 Antioxidant activity of the mushroom extracts

The total phenol content in the mushroom extracts obtained by water and by 60% methanol were determined by Folin-Ciocalteu reagent (Miliauskas et al., 2004). Each mushroom extract sample (250 μ L) was mixed with 2.5 mL Folin-Ciocalteu reagent diluted in distilled water (1:10). Then, 2 mL of 1 M Na₂CO₃ was added, and after it was mixed well, the solution was kept in the dark for 15 min before the absorbance was measured at 765 nm. Gallic acid at various concentrations (200-1000 μ g/mL) was prepared to generate data for the standard curve. The results are expressed in mg of gallic acid equivalent per gram of dried mushrooms (mg GAE/g dw).

The total flavonoid content of the mushroom extracts obtained by water and by 60% methanol were determined by aluminum chloride solution (Soares et al., 2017). Each mushroom extract sample (150 μ L) was mixed with 100 μ L of 5% NaNO₂ for 6 min. Then, 150 μ L of 10% AlCl₃ was added, and the mixture was maintained for 5 min. Then, 650 μ L of 1 M NaOH was added before measuring the absorbance at 510 nm. Rutin at concentrations between 200 and 1000 μ g/mL was used

to prepare the standard curve. The total flavonoid content was determined in mg of rutin equivalent per gram of mushroom dry weight (mg RE/g dw).

2.3 Scavenging activity

The scavenging activity of the mushroom extracts was determined by DPPH radical scavenging (Chen et al., 1999). Briefly, each mushroom extract sample ($200 \,\mu$ L) was mixed with 1.8 mL of DPPH solution ($3.94 \,\text{mg}$ of 2,2-diphenyl-1-picrylhydrazyl dissolved in 50 mL of 50% ethanol). The mixture was kept in the dark for 10 min before the absorbance was measured at 517 nm. Ascorbic acid ($10-50 \,\mu$ g/mL) was used as a reference compound. The absorbance was converted into the percentage of radical scavenging capacity by the following Equation 1:

$$DPPH \ scavenging \ (\%) = \frac{DPPH_{control} - (Sample_{control} - A_{sample})}{DPPH_{control}} \times 100$$
(1)

where: $\text{DPPH}_{\text{control}}$ is the absorbance of the DPPH solution; Sample_{control} is the absorbance of the mushroom extract in reaction with the DPPH solution; and A_{sample} is the absorbance of the mushroom extract.

The DPPH radical scavenging activity was reported as the mg ascorbic acid equivalent (AAE) per gram of mushroom dry weight (mg AAE/g dw).

2.4 Ferric Reducing Antioxidant Power (FRAP)

The reducing power of the mushroom extracts was determined by ferric-TPTZ reagent (Fu et al., 2010). Acetate buffer (0.25 M) was prepared by adding 30 µL of acetic acid in sodium acetate solution (41 mg), and the pH was adjusted to 3.6 before the volume was increased to 1000 mL. Ten millimolar TPTZ (2,4,6-Tris(2-pyridyl)-s-triazine) was prepared by dissolving 624.66 mg in a small volume of HCl and adjusting the volume to 200 mL with 0.25 M acetate buffer. Ferric chloride (20 mM) was generated by dissolving 1.0812 g of FeCl₂6H₂O in 0.25 M acetate buffer and adjusting the volume to 200 mL. The FRAP reagent was prepared by mixing 25 mL of acetate buffer with 2.5 mL of TPTZ solution and 2.5 mL of ferric chloride. The reagent mixture was incubated at 37 °C for 15 min before use, as it must be freshly prepared. Ascorbic acid (0-35 µg/mL) in water was used as a standard solution in this assay and was generated by mixing 150 µL of each concentration with 2.85 mL of FRAP reagent. The mixture was kept in the dark for 30 min, and its absorbance was measured at 590 nm. The FRAP value was measured based on the calibration curve of the ascorbic acid. The FRAP values were reported as a microgram of ascorbic acid equivalent per gram of mushroom dry weight (µg AAE/g dw).

2.5 GABA (y-aminobutyric acid) determination

The GABA content was determined by spectrophotometry (Karladee & Suriyong, 2012). Borate buffer was prepared by dissolving 2.54 g borax into 4.7 g boric acid and adjusting the volume to 1000 mL. Phenol (6%) was prepared by dissolving 6 mL phenol in 100 mL distilled water. The 6% sodium hypochlorite was prepared by adding 6 g NaClO into 100 mL distilled water. The reaction was as follows: each 0.5 mL of mushroom extract was mixed with 0.5 mL of borate buffer, 0.5 mL of 6% phenol and 1.5 mL 6% sodium hypochlorite. This mixture was boiled for 10 min and cooled before the absorbance was measured at 630 nm. The standard GABA (γ -C₄H₉NO₂) solution was prepared by dissolving 0.1 g in 100 mL distilled water (1000 µg/mL). Concentrations between 200 and 1000 µg/mL were analyzed and used to create the standard curve. The GABA content is reported as mg/g dw.

2.6 Amino acid determination

The amino acid profile of the mushrooms was measured by HPLC (Wandelen & Cohen, 1997). The mushroom powder was digested with 5 mL of 6 N HCl in a 110 °C heating block for 22 h. The internal standard was filtered and diluted with distilled water. The sample was filtered through a 0.45 μ m membrane filter. The mixture was mixed with AccQ-fluor derivatization buffer and AccQ-fluor reagent to obtain amino acid derivatives. The samples were heated at 55 °C for 10 min and measured by HPLC (Water Alliance 2695) using sodium acetate buffer in the mobile phase. The temperature of the column (Hypersil gold column C18 Z4.6*150 mm, 3 μ m) was adjusted to 30 °C. A Jasco FP2020 fluorescence detector (EX: 250, EM: 395 nm) was used. H-STD (Sigma) was used for the amino acid standards.

2.7 Statistical analysis

The results of all experiments were reported as the mean \pm standard deviation of three replicates. The results were evaluated using ANOVA (SPSS version 15.0) with $\alpha < 0.05$.

3 Results and discussion

3.1 Wild mushroom identification based on morphology

Twenty-two species of wild edible mushrooms were collected from the Nasrinual community forest. Based on morphology studies (Chandrasrikul et al., 2008), the species were placed in 9 families: Russulaceae (8), Boletaceae (5), Tricholomataceae (2), Agaricaceae (1), Amanitaceae (1), Pluteaceae (1), Auriculariaceae (1), Hymeogasteraceae (1), Diplocystaceae (1) and Sclerodermataceae (1), as shown in Table 1. Most of these species are well known and have been consumed by the local people for a long time. However, one species of Scleroderma sp. had first been found in northeast Thailand. It has a unique and pleasant odor that is different than that of other species of edible mushrooms. Although the mushrooms are abundant in the rainy season, between June and September, of each year, in some years, when rainfall is plentiful through October, these mushrooms found in the Nasrinual forest. Various types of edible mushrooms are collected by mushroom hunters for their own use and for sale. Figure 1 shows some edible mushrooms collected from the Nasrinual community forest.

3.2 Antioxidant activity

DPPH free radical scavenging

Free radical scavenging activity by the mushroom extracts obtained by water and 60% methanol were determined by the DPPH reagent. The DPPH• radical (purple) accepts an electron or hydrogen radical from antioxidant molecules. Upon electron transfer, the color changes from purple to yellow. Figure 2 shows the DPPH radical scavenging activity of the mushroom

Table 1. Mushrooms collected from the community forest in the Nasrinual District.

No.	Local Name	Code	Order	Family	Scientific Name
1	Hed-Koh-na-dang	RAMY	Russulales	Russulaceae	Russula emetica
2	Hed-Kai-na-kao	RLIC	Russulales	Russulaceae	Russula delica
3	Kai-na-luang	FLAV	Russulales	Russulaceae	Russula flavida
4	Hed-Kai-na-kaew	RHET	Russulales	Russulaceae	Russula virescens
5	Hed-Tan-lek	RDEN	Russulales	Russulaceae	Russula densifolia
6	Hed-Na-kao	RALB	Russulales	Russulaceae	Russula alboareolata
7	Hed-Dang-ku-lab	REME	Russulales	Russulaceae	Russula emitica
8	Hed-Nam-noom	LACT	Russulales	Russulaceae	Lactarius sp.
9	Hed-Ra-ngok-luang	AHEM	Agaricales	Pluteaceae	Amanita hemibapha
10	Hed-Sai-duan	AVAT	Agaricales	Amanitaceae	Amanita vaginata
11	Hed-Pluak-jig	TCLY	Agaricales	Tricholomataceae	Termitomyces clypeatus
12	Hed-Pluak-tab	TALB	Agaricales	Tricholomataceae	Termitomyces albuminosa
13	Hed-Ta-poo	CMIS	Agaricales	Agaricaceae	Calvatia boninensis
14	Hed-Hoo-noo	AUAR	Auriculariales	Auriculariaceae	Auricularia auricular
15	Hed-Peung-No1	BOL1	Boletales	Boletaceae	Boletus sp.
16	Hed-Peung-No.2	BOL2	Boletales	Boletaceae	Indoporus shorea
17	Hed-Peung-No.3	BOL3	Boletales	Boletaceae	Boletus sp.
18	Hed-Peung-No.4	BOL4	Boletales	Boletaceae	Hygrophorus canthavellus
19	Hed-Peung-No.5	BOL5	Boletales	Boletaceae	Boletus sp.
20	Hed-Klum-Ma	ALPO	Boletales	Hymeogasteraceae	Mycoamaranthus cambodgensis
21	Hed-Poa	ASTA	Boletales	Diplocystaceae	Astraeus hygrometricus
22	-	TUBER	Boletales	Sclerodermataceae	Scleroderma sp.



Figure 1. Some wild edible mushrooms collected from Nasrinual community forest. (A) AHEM; (B) ALPO; (C) RHET; (D) TUBER; (E) BOL2; (F) BOL3.

extracts obtained by water and methanol. It was found that the methanol extracts (1.79-34.98 mg AAE/g dw) had higher activity than the water extracts (1.91-12.58 mg AAE/g dw). This result was in agreement with the studies performed by Gan et al. (2013), and Boonsong et al. (2016). The highest value was from Hed-Peung-No.3 (BOL3) at 34.98 \pm 0.2833 mg AAE/g dw. The second highest level of DPPH scavenging radical activity was found in the methanol extract from Hed-Klum-Ma (ALPO) at 24.01 \pm 0.2341 mg AAE/g dw. Hed-Hoo-noo (AUAR) is a type of jelly fungi, and the extracts from it had the lowest DPPH scavenging activity at 1.69 \pm 0.061 and 2.34 \pm 0.038 mg/g dw for the water and methanol extracts, respectively. In addition, the DPPH scavenging activity in the extract of Boletaceae was higher than that in the samples from the other families, as shown in Figure 2.

Total phenolic content

In contrast with the results of the DPPH scavenging assay, most of the phenolic contents in the wild mushroom water extracts were higher than those in the methanol extracts. The total phenolic content in the water and methanol extracts was between 8.14-141.18 and 1.56-97.32 mg GAE/g dw, respectively. The highest total phenolic content was found in the water and methanol extracts of Hed-Peung-No.2 at 141.18 \pm 1.274, and 97.317 \pm 0.806 mg GAE/g dw, respectively (Figure 3). The lowest total phenolic content was in the water and methanol extract of Hed-Hoo-noo (AUAR) at 8.94 \pm 0.171 and 1.56 \pm 0.069 mg GAE/g dw, respectively. In this study, the total phenolic content of the water extract was higher than that of the methanol extract, which was consistent with



Figure 2. DPPH scavenging activity of the edible mushroom extracts obtained by water and 60% methanol; vertical bars indicate the standard deviation (n = 3).



Figure 3. Total phenolic content in mushroom extracts obtained by water and 60% methanol; vertical bars indicate the standard deviation (n = 3).

the finding of Boonsong et al. (2016). Furthermore, the total phenolic content in some species of the wild mushrooms (RAME, RDEN, TALB, CMIS, BOL1, BOL2, and BOL3) was higher than that of cultivated mushrooms (*Lentinus edodes, Volvariella volvacea, Pleurotus eous, Pleurotus sajor-caju, and Auricularia auricular*).

Reducing power as determined by the FRAP method

The results of the reducing power assay by the RFAP method for the mushroom extracts obtained by water and methanol are shown in Figure 4. It was found that the FRAP value for the water extract was higher than it was for the methanol extract at 1.76-15.00 and 0.815-14.19 mg AAE/g, respectively. The reducing power of the water extract from

ALPO was the highest at 15.00 ± 0.405 mg AAE/g dw. The highest reducing power among the methanol extracts was 14.19 mg AAE/g dw from Hed-Peung-No.2. This study revealed that the FRAP value for the mushroom extract obtained by water was higher than for the extract obtained by methanol. Chirinang & Intarapichet (2009) and Puttaraju et al. (2006) also reported that the FRAP values of *Pleurotus ostreatus* and *P. sajor-caju* water extracts were higher than those of their ethanol extracts.

Total flavonoid contents

Figure 5 shows the total flavonoid content in the mushroom extracts. The results indicated that the flavonoid content in the mushroom extract obtained by methanol was slightly



Figure 4. Reducing power determined by FRAP method on mushroom extracts obtained by water and 60% methanol in terms of ascorbic acid equivalency (AAE); vertical bars indicate the standard deviation (n = 3).



Figure 5. Total flavonoid content of mushroom extracts obtained by water and 60% methanol; vertical bars indicate the standard deviation (n = 3).

higher than that in the extract obtained by water. These results were similar to those of Gan et al. (2013), who also found that the flavonoid content of Brazilian and button mushroom in extracts obtained by methanol was higher than that of extracts obtained by water. In this study, the total flavonoid content of all mushroom extracts, obtained by both methanol and water, was relatively low. The flavonoid content in the water extracts were between 0.88 and 188.43 mg RE/g dw, and that in the 60% methanol extracts were between 1.72 and 216.46 mg RE/g dw. The highest flavonoid content was in Hed-Peung-No.2 (BOL2), in which it was higher in the methanol extract compared to the water extract, but it was low in most of the mushrooms.

y-Aminobutyric acid (GABA)

GABA content was also determined for the mushroom extracts by water and methanol. The results are shown in Table 2. The GABA content was higher in the water extracts than it was in the methanol extracts. The highest GABA level was found in the water extract of Hed-Sai-duan at 23.304 ± 0.027 mg/g dw, and the lowest GABA level was found in the methanol extract of Hed-Sai-duan at 0.999 ± 0.072 mg/g dw. Chen et al. (2012) studied GABA from the fruiting body and mycelium of various mushrooms. They found that the fruiting bodies of *Flammulina velutipes* and *Boletus edulis* contained the greatest amounts of GABA (0.229 and 0.202 mg/g dw, respectively). In contrast, mycelia from *Cordyceps cicadae* contained the highest GABA content at 0.254 mg/g dw. This finding indicated that wild edible mushrooms are a good source of GABA.

Amino acids

Mushrooms are well known as high protein sources. Furthermore, all types of essential amino acids are found in mushrooms. Due to the method used to determine the amino acids in this study, only 15 types of essential amino acids were found; 8 of them were essential amino acids (EAAs), and 7 were non-essential amino acids. The eight EAAs were arginine, leucine, isoleucine, lysine, valine, threonine, phenylalanine, and histidine. Nonessential amino acids included aspartic acid, serine, glutamic acid, glycine, alanine, proline and tyrosine. All types of amino acids in these mushrooms are shown in Table 3. The amount of amino acid detected was between 0.11 and 5.66 g/100 g dried weight. Glutamic acid was found in the greatest amounts in the mushrooms at 0.77-5.66 g/100 g. These results correspond with the report of Kouassi et al. (2016) that indicated that glutamic acid and valine are the most predominant amino acids in *R. lepida*, R. mustelina and, R. delica. However, a food protein is expected to have a high ratio of essential amino acids to total amino acids. In this study, the EAA:total amino acid ratio was 0.389-0.462. The high ratio of essential amino acid:total amino acid indicated that these mushrooms have a high protein content. In this study, only Hed-Klum-Ma had a low essential amino acid:total amino acid ratio (0.389).

Table 2. Total amino acid and GABA analysis of wild mushroom extracts obtained bywater and methanol.

Edible mushrooms —	Water extract	Methanol extract	Total amino acid,	EAA*/Total amino acid ratio				
	Gaba mg/g dw	Gaba mg/g dw	g/100 g dw					
Hed-Koh-na-dang	$7.505\pm0.166^{\rm h}$	$2.725\pm0.014^{\rm ef}$	15.98 ± 0.06	0.419				
Hed-Kai-na-kao	$8.959\pm0.280^{\rm g}$	1.713 ± 0.072^{ghij}	14.38 ± 0.022	0.445				
Kai-na-luang	${\bf 3.799 \pm 0.140^k}$	$4.163\pm0.057^{\mathrm{bc}}$	16.01 ± 0.021	0.423				
Hed-Kai-na-kaew	$5.186\pm0.054^{\text{j}}$	$1.818\pm0.028^{\text{ghij}}$	16.24 ± 0.016	0.443				
Hed-Tan-lek	$14.404 \pm 0.055^{\rm e}$	$8.282\pm0.366^{\text{a}}$	12.01 ± 0.012	0.423				
Hed-Na-kao	$15.816 \pm 0.374^{\rm d}$	$2.569\pm0.014^{\text{efg}}$	16.8 ± 0.022	0.419				
Hed-Dang-ku-lab	$15.855 \pm 0.495^{\rm d}$	$1.700\pm0.042^{\rm ghij}$	18.98 ± 0.018	0.434				
Hed-Nam-noom	$7.337\pm0.243^{\rm hi}$	$0.988\pm0.027^{\rm j}$	14.49 ± 0.018	0.452				
Hed-Ra-ngok-luang	2.699 ± 0.322^{1}	$2.081\pm0.187^{\rm fghi}$	21.82 ± 0.021	0.449				
Hed-Sai-duan	$23.304 \pm 0.027^{\rm a}$	0.999 ± 0.072^{j}	23.46 ± 0.03	0.414				
Hed-Pluak-jig	$4.017\pm0.054^{\rm k}$	1.840 ± 0.000^{jghij}	22.65 ± 0.015	0.432				
Hed-Pluak-tab	$3.501\pm0.039^{\rm k}$	1.235 ± 0.029^{ij}	19.53 ± 0.018	0.412				
Hed-Ta-poo	$10.508 \pm 0.240^{\rm f}$	8.038 ± 0.579^{a}	24.51 ± 0.029	0.437				
Hed-Hoo-noo	$3.818\pm0.152^{\rm k}$	1.315 ± 0.159^{ij}	6.41 ± 0.007	0.462				
Hed-Peung-No1	$6.646\pm0.263^{\mathrm{i}}$	$2.859\pm0.202^{\rm ef}$	13.94 ± 0.013	0.428				
Hed-Peung-No.2	$20.328\pm0.418^{\circ}$	$2.246\pm0.014^{\rm fgh}$	21.99 ± 0.035	0.462				
Hed-Peung-No.3	$10.617 \pm 0.108^{\rm f}$	$4.677 \pm 0.299^{\text{b}}$	19.87 ± 0.026	0.429				
Hed-Peung-No.4	ND	$3.829\pm0.846^{\rm cd}$	27.36 ± 0.022	0.454				
Hed-Peung-No.5	$21.445 \pm 0.589^{\mathrm{b}}$	$1.438\pm0.102^{\rm hij}$	28.41 ± 0.028	0.444				
Hed-Klum-Ma	$2.153 \pm 0.052^{\rm l}$	$3.321\pm0.162^{\rm de}$	5.76 ± 0.007	0.389				
Hed-Poa	4.808 ± 0.007^{j}	3.7905 ± 0.157^{cd}	8.64 ± 0.009	0.441				

*EAA = Essential amino acid; ND = Not Detected. Means followed by same letter in the same column are not significantly different (p<0.05 according to Duncan's test). Data are presented as the mean \pm standard deviation of three replicates (n = 3).

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	*əninslalıqnəd	$0.66\pm0.012^{\rm f}$	$0.69\pm0.036^{\rm ef}$	$0.69 \pm 0.016^{\mathrm{ef}}$	$0.75\pm0.022^{\rm e}$	$0.55\pm0.001^{\rm gh}$	$0.73\pm0.036^{\rm e}$	$0.83 \pm \mathbf{0.003^d}$	$0.66\pm0.009^{\rm f}$	$1.01\pm0.016^{\rm b}$	$1.00\pm0.027^{\mathrm{b}}$	$1.00\pm0.016^{\rm b}$	$0.92\pm0.009^{\rm c}$	$1.09\pm0.035^{\mathrm{a}}$	$0.36\pm0.006^{\rm i}$	$0.50\pm0.010^{\rm h}$	$1.01\pm0.058^{\rm b}$	$0.59\pm0.017^{\mathrm{g}}$	$1.02\pm0.014^{\rm b}$	$1.01\pm0.028^{\rm b}$	0.22 ± 0.006	$0.33\pm\mathbf{0.008^{i}}$	$0.18\pm0.005^{\rm i}$	$0.58 \pm \mathbf{0.008^{s}}$	$0.54\pm0.006^{\mathrm{gh}}$	3).
	*9niou9.I	1.17 ± 0.011^{i}	1.16 ± 0.011^{i}	$1.21\pm0.015^{\rm i}$	$1.27\pm0.022^{\rm h}$	$0.92\pm0.004^{\rm k}$	$1.28\pm0.01\mathrm{l^h}$	$1.49\pm0.021^{\rm f}$	1.17 ± 0.012^{i}	$1.76\pm0.015^{\rm d}$	1.77 ± 0.013^{d}	$1.76\pm0.016^{\rm d}$	$1.40\pm0.005^{\mathrm{g}}$	$1.93\pm0.021^{\rm c}$	$0.55\pm0.003^{\mathrm{m}}$	0.91 ± 0.009^k	$1.75\pm0.038^{\rm d}$	$1.55\pm0.017^{\rm e}$	$\textbf{2.22}\pm\textbf{0.021}^{a}$	$\textbf{2.16}\pm\textbf{0.032}^{b}$	$0.42\pm0.010^{\rm n}$	$0.78\pm0.008^{\text{l}}$	$0.33\pm0.008^\circ$	$1.03\pm0.006^{\rm j}$	0.75 ± 0.011^{l}	eplicates (n =
	soleucine	0.67 ± 0.014^{ijk}	0.65 ± 0.002^k	0.71 ± 0.011^{hi}	0.73 ± 0.015^{h}	0.52 ± 0.001^{m}	0.70 ± 0.002^{hij}	$0.87\pm0.007^{\mathrm{g}}$	0.61 ± 0.015^{1}	$0.94 \pm 0.011^{\circ}$	$1.04 \pm 0.019^{\circ}$	0.99 ± 0.012^{d}	0.78 ± 0.002^{8}	1.14 ± 0.017^{b}	$0.26\pm0.001^\circ$	0.50 ± 0.000^{m}	$0.95\pm0.020^{\rm de}$	$0.92 \pm 0.019^{\circ}$	1.30 ± 0.012^{a}	1.33 ± 0.023^{a}	$0.23\pm0.004^\circ$	0.46 ± 0.008^{n}	0.19 ± 0.005^{p}	$0.66 \pm 0.008^{\rm ik}$	0.43 ± 0.010^{n}	ation of three r
	*9nisyJ	$0.98\pm0.026^{\rm h}$	0.71 ± 0.013^{1}	0.75 ± 0.024^{kl}	0.82 ± 0.016^{i}	0.71 ± 0.022^{1}	0.90 ± 0.013^{i}	1.08 ± 0.027^{fg}	$1.03\pm0.025^{\rm gh}$	$1.39\pm0.024^{\rm d}$	1.37 ± 0.025^{d}	$1.47 \pm 0.015^{\circ}$	1.06 ± 0.020^{fg}	$1.30\pm0.006^{\circ}$	0.36 ± 0.012^{m}	0.82 ± 0.026^{jk}	$1.38\pm0.017^{\rm d}$	1.11 ± 0.031^{f}	$1.59 \pm 0.024^{\rm b}$	1.73 ± 0.036^{a}	0.23 ± 0.003^{n}	0.36 ± 0.008^{m}	0.22 ± 0.001^{n}	0.77 ± 0.003^{jkl}	0.38 ± 0.017^{m}	standard devi
	*snihsV	$0.83\pm0.024^{\rm j}$	$0.82\pm0.010^{\rm i}$	0.87 ± 0.016^{hi}	$0.91\pm0.012^{\rm h}$	0.61 ± 0.003^{m}	0.87 ± 0.010^{hi}	$1.00\pm0.007^{\mathrm{g}}$	0.77 ± 0.020^k	$1.12\pm0.016^{\rm f}$	$1.15\pm0.023^{\rm ef}$	$1.18\pm0.008^{\circ}$	$1.03\pm0.005^{\mathrm{g}}$	$1.33\pm0.018^{\rm c}$	$0.40\pm0.000^\circ$	$0.67\pm0.006^{\rm i}$	$1.17\pm0.022^{\rm ef}$	$1.23\pm0.035^{\rm d}$	$1.60\pm0.004^{\rm b}$	$1.66\pm0.025^{\mathrm{a}}$	0.29 ± 0.006^p	0.50 ± 0.007^{n}	$0.23\pm0.007\text{s}$	$0.73\pm0.008^{\rm k}$	$0.48\pm0.010^{\rm n}$	as the mean ±
	*anibitaiH	$0.43\pm0.008^{\rm hi}$	$0.36\pm0.016^{\rm k}$	$0.39\pm0.010^{\mathrm{ini}}$	0.44 ± 0.005^{gh}	$0.33\pm0.011^{\rm M}$	$0.40\pm0.016^{\text{hij}}$	$0.51\pm0.005^{\rm f}$	$0.39\pm0.010^{\mathrm{hij}}$	$0.66\pm0.010^{\rm d}$	$0.72\pm0.028^{\rm c}$	$0.65\pm0.007^{\rm d}$	$0.59\pm0.014^{\rm e}$	$0.80\pm0.021^{\rm b}$	0.17 ± 0.013 n	$0.47\pm0.011^{\mathrm{fg}}$	0.65 ± 0.036^{d}	0.49 ± 0.009^{i}	$0.72\pm0.010^{\circ}$	$0.84\pm0.014^{\mathrm{a}}$	0.17 ± 0.003^{n}	$0.21\pm0.005\mathrm{mn}$	$0.11\pm0.002^\circ$	0.30 ± 0.006^{1}	$0.23\pm0.011\mathrm{m}$	are presented a
g dry weight	*əninoərdT	0.94 ± 0.012^{g}	$1.00\pm0.025^{\rm f}$	$1.04\pm0.007^{\rm f}$	$1.11\pm0.016^{\rm e}$	$0.69 \pm 0.009^{\circ}$	$1.00\pm0.025^{\rm f}$	$1.25\pm0.006^{\rm d}$	$0.89\pm0.009^{\rm sh}$	$1.32\pm0.007^{\circ}$	$1.33\pm0.030^{\circ}$	$1.29\pm0.008^{\rm cd}$	$1.16 \pm 0.016^{\circ}$	$1.56\pm0.024^{\mathrm{b}}$	$0.45\pm0.004^{\rm k}$	0.84 ± 0.009^{h}	$1.32 \pm 0.040^{\circ}$	$1.15\pm0.016^{\circ}$	$1.83\pm0.037^{\mathrm{a}}$	1.83 ± 0.028 ^a	0.35 ± 0.006^{1}	$0.58 \pm 0.012^{\circ}$	0.31 ± 0.006^{1}	$0.87 \pm 0.012^{\rm h}$	$0.56 \pm 0.010^{\circ}$	can's test. Data
Amino acid content, g/100 g	*əniniyaA	$1.01 \pm 0.013^{\circ}$	$1.01 \pm 0.045^{\circ}$	$1.12\pm0.016^{\rm ghi}$	1.16 ± 0.022^{fgh}	0.77 ± 0.006^{k}	$1.16\pm0.045^{\rm fgh}$	1.21 ± 0.00^{fg}	1.03 ± 0.011^{ij}	$1.59 \pm 0.016^{\circ}$	$1.33\pm0.042^{\circ}$	$1.45\pm0.013^{\rm d}$	$1.11\pm0.015^{\rm hij}$	$1.57\pm0.041^{\circ}$	0.41 ± 0.009^{m}	$1.25\pm0.007^{\rm ef}$	$1.93\pm0.096^{\rm b}$	$1.48\pm0.030^{\rm d}$	2.14 ± 0.048^{a}	2.06 ± 0.037 ^a	0.33 ± 0.010^{mn}	0.59 ± 0.018^{1}	0.28 ± 0.008^{n}	$0.75\pm0.018^{\rm k}$	0.50 ± 0.006^{1}	ording to Dunc
	anisoryT	$0.31\pm0.004^{\rm h}$	$0.32\pm0.023^{\rm h}$	$0.37\pm0.007^{\rm fg}$	$0.39\pm0.016^{\mathrm{fg}}$	1.00 ± 0.013 ^a	0.38 ± 0.023^{fg}	$0.40\pm0.006^{\rm f}$	$0.41\pm0.007^{\rm f}$	$0.55\pm0.007^{\rm de}$	$0.53\pm0.017^{\circ}$	$0.57 \pm \mathbf{0.008^{de}}$	$0.59\pm0.005^{\rm d}$	$0.59\pm0.015\mathrm{d}$	0.16 ± 0.006	$0.34\pm0.006^{\mathrm{gh}}$	$0.78\pm0.054^{\rm c}$	$0.54\pm0.013^{\rm de}$	$0.93\pm0.018^{\rm b}$	$0.95\pm0.013^{\rm b}$	$0.14\pm0.005^{\rm i}$	0.21 ± 0.005^i	0.12 ± 0.003^{j}	$0.31\pm0.011^{\rm h}$	$0.24\pm0.004^{\rm i}$	(p < 0.05) acco
	Proline	$0.79\pm0.014^{\rm i}$	$0.92\pm0.016^{\mathrm{g}}$	$0.92\pm0.009^{\rm g}$	$1.00\pm0.009^{\ell}$	$0.65\pm0.010^{\circ}$	$0.92\pm0.016^{\rm g}$	$0.85\pm0.006^{\mathrm{g}}$	0.76 ± 0.025^i	1.03 ± 0.009^{cf}	$1.06\pm0.003^{\rm e}$	$1.02\pm0.008^{\rm cf}$	$0.83\pm0.009^{\rm h}$	$1.55\pm0.024^{\rm ab}$	$0.38\pm0.005^{\rm l}$	$0.77 \pm 0.009^{\circ}$	$1.13\pm0.012^{\rm d}$	$1.38\pm0.023^{\circ}$	$1.54\pm0.003^{\mathrm{b}}$	$1.58\pm0.018^{\mathrm{a}}$	0.26 ± 0.005^{m}	$0.45\pm0.002^{\rm k}$	$0.22\pm0.007^{\mathrm{n}}$	$0.63\pm0.004^{\rm j}$	$0.45\pm0.005^{\rm k}$	cantly different
	əninslA	1.15 ± 0.017^{g}	$0.86 \pm \mathbf{0.005^k}$	$1.05 \pm 0.029^{\rm h}$	0.98 ± 0.003^{1}	$0.71\pm0.008^{\rm i}$	1.00 ± 0.005^{i}	$1.35\pm0.025^{\circ}$	$0.93 \pm 0.006^{\circ}$	$1.50\pm0.029^{\rm cd}$	$1.28\pm0.006^{\rm f}$	1.73 ± 0.013^{a}	$1.35\pm0.006^{\circ}$	$1.48\pm0.008^{\rm d}$	0.55 ± 0.007^{n}	$0.91 \pm 0.014^{\circ}$	$1.35\pm0.011^\circ$	1.14 ± 0.014^{8}	$1.53\pm0.008^\circ$	$1.63 \pm 0.014^{\rm b}$	$0.37\pm0.003^\circ$	0.66 ± 0.006^{m}	$0.33\pm0.003^\circ$	0.85 ± 0.011^k	0.67 ± 0.005^{m}	are not signifi
	Glycine	0.76 ± 0.006^{i}	$0.86\pm0.026^{\rm gh}$	$0.90 \pm 0.014^{\mathrm{fg}}$	$0.97\pm0.020^{\circ}$	$0.86 \pm 0.011^{\rm gh}$	0.90 ± 0.026^{fgh}	$0.94 \pm 0.010^{\rm ef}$	$0.70 \pm 0.011^{\circ}$	$1.06 \pm 0.014^{\rm d}$	$1.17 \pm 0.029^{\circ}$	$1.21 \pm 0.022^{\circ}$	$1.06\pm0.020^{\rm d}$	1.56 ± 0.011^{b}	0.35 ± 0.006^{m}	$0.84 \pm 0.011^{\rm h}$	$1.21\pm0.034^{\circ}$	$1.19 \pm 0.015^{\circ}$	1.75 ± 0.021^{a}	1.78 ± 0.028^{a}	0.31 ± 0.002^{m}	0.47 ± 0.006^{1}	0.24 ± 0.004^{n}	$0.62 \pm 0.014^{\rm k}$	0.44 ± 0.008^{1}	e same column
	Glutamic acid	$3.52\pm0.038^{\rm f}$	$\textbf{2.19}\pm\textbf{0.033}^{\text{I}}$	2.69 ± 0.046	$2.46\pm0.033^{\rm k}$	$1.53\pm0.016^{\mathrm{m}}$	$3.51\pm0.033^{\rm g}$	$3.92 \pm \mathbf{0.088^d}$	$\textbf{2.47}\pm\textbf{0.055}^k$	$3.50\pm0.046^{\rm g}$	$5.66 \pm 0.104^{\rm a}$	$4.32\pm0.028^{\rm b}$	$4.05\pm0.054^{\rm cd}$	$3.18\pm0.035^{\rm i}$	0.77 ± 0.010^{p}	$2.21\pm0.046^{\rm i}$	$3.36\pm0.015^{\rm h}$	$2.49 \pm 0.054^{\mathrm{k}}$	$3.78\pm0.043^{\rm e}$	$4.09\pm0.055^{\rm c}$	$0.91\pm0.007^{\rm op}$	$1.30\pm0.014^{\rm n}$	$0.98\pm0.004^\circ$	$\textbf{2.15}\pm\textbf{0.033}^l$	$1.39\pm0.037^{\rm n}$	me letter in th
	Serine	$0.87\pm\mathbf{0.006^{i}}$	$0.93\pm0.024^{\rm gh}$	$1.03\pm0.020^{\rm f}$	$1.07\pm0.014^{\mathrm{f}}$	0.72 ± 0.013^{i}	$1.01\pm0.024^{\rm f}$	$1.15\pm0.015^{\rm e}$	$0.95\pm0.012^{\mathrm{g}}$	$1.38\pm0.020^{\rm d}$	$1.36\pm0.032^{\rm d}$	1.37 ± 0.032^{d}	$1.15\pm0.014^{\rm e}$	$1.58\pm0.026^{\circ}$	$0.41\pm0.009^{\rm i}$	$0.88 \pm \mathbf{0.006^{hj}}$	1.35 ± 0.044^{d}	$1.15\pm0.009^{\circ}$	$1.90\pm0.021^{\rm b}$	$\textbf{2.01} \pm \textbf{0.023}^{a}$	$0.36\pm0.012^{\rm i}$	0.55 ± 0.014^{k}	0.29 ± 0.006^{m}	0.85 ± 0.017^{i}	0.57 ± 0.015^{k}	followed by sa
	Aspartic acid	$1.89 \pm 0.041^{\circ}$	1.90 ± 0.042^{ij}	$2.27\pm0.070^{\rm g}$	$2.18\pm0.010^{\mathrm{gh}}$	1.44 ± 0.049^{1}	$2.04\pm0.042^{\rm hi}$	$2.13\pm0.040^{\mathrm{gh}}$	1.72 ± 0.039^k	$3.01\pm0.070^{\rm d}$	$2.69\pm0.053^\circ$	$2.64\pm0.022^{\circ}$	$2.47\pm0.072^{\rm f}$	3.85 ± 0.128^{a}	0.83 ± 0.007^{n}	$2.03 \pm \mathbf{0.025^{hi}}$	$2.65\pm0.022^{\circ}$	$3.46\pm0.087^{\rm c}$	$3.51\pm0.040^\circ$	3.75 ± 0.052^{b}	1.17 ± 0.020^{m}	1.19 ± 0.016^{m}	$0.63\pm0.006^\circ$	$1.76 \pm 0.006^{ m k}$	$1.23\pm0.045^{\rm lm}$	(EAA). Means
Edible mushrooms		Hed-Koh-na-dang	Hed-Kai-na-kao	Kai-na-luang	Hed-Kai-na-kaew	Hed-Tan-lek	Hed-Na-kao	Hed-Dang-ku-lab	Hed-Nam-noom	Hed-Ra-ngok-luang	Hed-Sai-duan	Hed-Pluak-jig	Hed-Pluak-tab	Hed-Ta-poo	Hed-Hoo-noo	Hed-Peung-No1	Hed-Peung-No.2	Hed-Peung-No.3	Hed-Peung-No.4	Hed-Peung-No.5	Hed-Klum-Ma	Hed-Poa (spore)	Hed-Poa (bark)	Bamboo Mushroom ^a	Bamboo Mushroom ^b	*Essential amino acid

4 Conclusion

These findings show that the wild edible mushrooms in the Nasrinual community forest are not only highly diverse but also of high nutritional value, which is consistent with their use as food sources for a long time. Wild edible mushrooms contain essential amino acids and substantial glutamic acid, imparting an umami taste. Various antioxidant activities and high levels of GABA were also found in the wild edible mushrooms. Therefore, they are good for human health.

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