# Terpenoids, Flavonoids and other Constituents of *Eupatorium betonicaeforme* (Asteraceae)

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Um novo diterpeno kaureno, caracterizado como ácido  $15\alpha$ -decanoiloxi-kaur-16-en-19-óico, juntamente com os compostos conhecidos: ácido pentacosanóico,  $24\alpha$ -etil- $5\alpha$ -colesta-7,22*E*-dien- $3\beta$ -ol, ácido  $15\alpha$ -hidroxi-kaur-16-en-19-óico,  $8\beta$ -angeloil- $9\beta$ ,10 $\beta$ -diidroxi-1-oxo-germacra-4E,11(13)dien-12, $6\alpha$ -olideo,  $3\beta$ -hidroxiicosan-1, $5\beta$ -olideo, acetate de taraxasterila, 7-O-metilcampferol, caempferol e nepetina foram isolados das flores de *Eupatorium betonicaeforme*. Adicionalmente, foram isolados das partes aéreas, o acetato de taraxasterila e a mistura binária de  $\alpha$ - e  $\beta$ -amirina, enquanto foram obtidos das raízes, a mistura de  $\beta$ -sitosterol e estigmasterol, e 6-acetil-2,2-dimetilcroman-4-ona. A elucidação estrutural de todos os compostos isolados neste trabalho foi realizada por análise espectroscópica e comparação com dados descritos na literatura.

A new acylated kaurene diterpene, characterized as  $15\alpha$ -decanoyloxy-kaur-16-en-19-oic acid, along with nine known compounds: pentacosanoic acid,  $24\alpha$ -ethyl- $5\alpha$ -cholesta-7,22*E*dien- $3\beta$ -ol,  $15\alpha$ -hydroxy-kaur-16-en-19-oic acid,  $8\beta$ -angeloyloxy- $9\beta$ ,  $10\beta$ -dihydroxy-1-oxogermacra-4E, 11(13)dien- $12,6\alpha$ -olide,  $3\beta$ -hydroxyeicosan- $1,5\beta$ -olide, taraxasteryl acetate, 7-*O*methylkaempferol, kaempferol, and nepetin were isolated from the flowers of *Eupatorium betonicaeforme* (Asteraceae). In addition, from the aerial parts were isolated taraxasteryl acetate and  $\alpha$ - and  $\beta$ -amyrin, while the mixture of  $\beta$ -sitosterol and stigmasterol, and 6-acetyl-2,2dimethylchroman-4-one were isolated from the roots. The structure elucidation of all compounds was performed by spectroscopic analysis and comparison with published data from literature.

Keywords: *Eupatorium betonicaeforme*, Asteraceae, triterpenes, diterpenes, flavonoids, lactones

## Introduction

The genus *Eupatorium* (family Asteraceae, tribe Eupatorieae, subtribe Eupatoriinae), is a taxonomically complex group of species with distribution centers mainly in Europe, eastern Asia and North America.<sup>1</sup> Several studies on *Eupatorium* species have been reported, revealing a great number and diversity of secondary metabolites, where the sesquiterpenes lactones of the guaianolide,<sup>2</sup> germacranolide,<sup>3</sup> heliangolide,<sup>4</sup> and eudesmanolide types<sup>5</sup> are the most characteristic compounds. Flavonoids,<sup>6</sup> diterpenes,<sup>7</sup> benzofurans,<sup>8</sup> pyrrolizidine alkaloids,<sup>9</sup> chromenes<sup>10</sup> and thymol derivatives<sup>11</sup> are also common. Triterpenes,<sup>12</sup> sesquiterpenes,<sup>13</sup> coumarins<sup>14</sup> and  $\delta$ -lactones of the long chain type<sup>15</sup> have been occasionally isolated. A great number of *Eupatorium* species are employed in traditional medicine in the treatment of different pathologies. Indeed, a significant number of bioactive natural compounds have been reported from this renowned genus.<sup>16-19</sup>

As part of a continuing search on native plants from Northeast of Brazil, particularly of the Asteraceae family, the volatile oils from the roots and leaves of *E*. *betonicaeforme* yielded 2,2-dimethyl-6-vinylchroman-4one (**1**) and 2-senecioyl-4-vinylphenol (**2**), the sesquiterpenes  $\beta$ -caryophyllene,  $\gamma$ -muurolene and bicyclogermacrene as the main constituents. Both oils showed larvicidal effects against *Aedes aegypti* larvae, particularly the oil from the roots.<sup>20</sup> The major constituents (**1** and **2**) were isolated and tested. Compound **1** was the most active compound and is probably responsible for the larvicidal effect.<sup>20</sup> Continuing with the phytochemical investigation of *E. betonicaeforme* this paper reports now its nonvolatile chemical composition.

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### **Results and Discussion**

The extract of the flowers of E. betonicaeforme, after successive column chromatography on Si gel, afforded the new secondary metabolite  $15\alpha$ -decanoyloxy-kaur-16-en-19oic acid (3), together with the known compounds: pentacosanoic acid (4),  $24\alpha$ -ethyl- $5\alpha$ -cholesta-7,22E-dien- $3\beta$ -ol (5),<sup>21</sup> 15\alpha-hydroxy-kaur-16-en-19-oic acid (6),<sup>22</sup> 8\betaangeloyloxy-9 $\beta$ ,10 $\beta$ -dihydroxy-1-oxo-germacra-4E,11(13)dien-12,6 $\alpha$ -olide (7),<sup>23</sup> 3 $\beta$ -hydroxyeicosan-1,5 $\beta$ olide (8),<sup>24</sup> taraxasteryl acetate (9),<sup>25</sup> 7-O-methylkaempferol (10), kaempferol (11), and nepetin (12).<sup>26</sup> From the hexane extract of the aerial parts (leaves and stems) were isolated taraxasteryl acetate (9) and  $\alpha$ - and  $\beta$ -amyrin (13 and 14),<sup>25</sup> while from the hexane extract of roots were obtained a mixture of  $\beta$ -sitosterol and stigmasterol (15 and 16) and 6-acetyl-2,2-dimethylchroman-4-one (17),<sup>27</sup> Figure 1. All of the known compounds were identified by comparison of their spectral data with those reported in the literature. Compound 7 was previously isolated from Trichogonia salviaefolia (Asteraceae) by Bohlmann et al., and its molecular structure has been deduced by <sup>1</sup>H NMR and MS.<sup>23</sup> Complete <sup>1</sup>H and

Table 1. <sup>1</sup>H and <sup>13</sup>C NMR data for compounds 3 (CDCl<sub>3</sub>) and 7 (acetone-d<sub>6</sub>)

<sup>13</sup>C NMR chemical shifts of **7**, which were assigned by a combination of 1D and 2D NMR techniques, are listed in Table 1.

Compound 3 was isolated as a colorless waxy solid. The molecular formula  $C_{20}H_{48}O_4$  was deduced by a combination of EIMS ([M]+, m/z 472) and <sup>13</sup>C NMR data. The <sup>13</sup>C NMR spectrum showed a total of 30 carbon signals, among the characteristic signals at  $\delta_c$  29.9-29.3 and 14.3 consistent with the presence of aliphatic methylene groups of a fatty acyl moiety. The deshielded carbon signals at  $\delta_c$  184.8 and 174.1 indicated the presence of two carbonyl groups compatible with the presence of carboxylic acid and ester functions, respectively. IR absorption bands at 1730 and 1694 cm<sup>-1</sup> further supported these findings. The additional deshielded carbon signals at  $\delta_{\rm c}$  at 155.7 and 110.0, as well the proton signal at  $\delta_{11}$  5.26 (2H, brs) clearly demonstrated the presence of an exomethylene moiety. In the HMBC spectrum were observed correlations for the exocyclic double-bond protons  $(\delta_{\rm u}, 5.26)$  with the oxymethine carbon signal at  $\delta_{\rm c}, 82.9$ revealing the contiguity of these units. Comparison of the chemical shifts with the literature data indicated the presence of a kaur-16-en-19-oic acid bearing an oxygen atom at C-

	3			7
С	$\delta_{ m c}$	$\delta_{\rm H}$ (m, J in Hz)	$\delta_{ m c}$	$\delta_{_{\rm H}}$ (m, J in Hz)
1	37.8	2.15 (d, J 13.1)	216.6	-
2	19.2	1.44 (m)	39.1	3.64 (td, J 13.6; 3.0)
		1.60 (m)		2.25 (dt, J 13.6; 3.7)
3	34.9	1.44 (m)	36.8	2.95 (m) 2.15 (dt, J 11.9; 3.0)
4	43.9	-	145.4	-
5	56.8	1.09 (d, J 11.9)	126.4	4.90 (br d, <i>J</i> 10.4)
6	20.9	1.86 (m)	76.3	5.21 (dd, J 7.8; 10.4)
		1.77 (m)		
7	40.7	1.86 (m)	50.4	2.83°
8	47.6	-	71.7	5.86 (dd, J 2.5; 4.0)
9	53.1	1.29-1.25	80.1	4.35 (dd, <i>J</i> 4.0; 6.4)
10	40.0	-	82.0	-
11	18.6	1.60 (m)	138.2	-
12	32.8	1.60 (m); 1.49	169.8	-
13	42.7	2.77 (br s)	120.4	6.11 (d, J 3.2)5.43 (d, J 3.2)
14	37.4	1.96 (d, J 11.5); 1.44	26.1	1.48 (s)
15	82.9	5.26 (s)	19.4	1.87 (d, <i>J</i> 1.3)
16	155.7	-	-	-
17	110.0	5.07 (s)	-	-
18	29.0	1.23 (s)	-	-
19	184.8	-	-	-
20	15.9	0.95 (s)	-	-
1'	174.1	-	167.1	-
2'	34.9	2.91 (t, J 7.5)	128.7	-
3'	25.3	1.60 (m)	139.2	6.13 (m)
4'			16.1	1.93 (dq, J 1.4; 7.3)
5'			20.8	1.82 (quint. J 1.4)
6'	29.9-29.3ª	1.29-1.25 (m) <sup>b</sup>	-	-
7'			-	-
8'	32.0		-	-
9'	22.8		-	-
10'	14.3	0.87 (t, <i>J</i> 6.4)	-	-

<sup>a</sup> Chemical shifts for C-4' - C-7'; <sup>b</sup> Chemical shifts for H-4' - H-9'; <sup>c</sup> Multiplicity not determined due partial overlapping with the water peak.



Figure 1. Chemical constituents isolated from *E. betonicaeforme*.

15.<sup>22,28</sup> This fact was also reinforced by the EIMS, which showed an intense ion peak at m/z 318 ( $C_{20}H_{30}O_3$ ) related to a proton rearrangment followed by loss of the fatty side chain. In accordance with published data, it was assumed that diterpene **3** belongs to the *ent*-series, like the other kaurenes previously isolated from *Eupatorium*.<sup>28,29</sup> Additional support for this affirmation was the small negative specific rotation ( $[\alpha]_D^{20}$ -0.66°, *c* 0.05, CHCl<sub>3</sub>), which is characteristic of *ent*kaurenes.<sup>30</sup> The location of the acyl moiety was supported by the long-range correlation of the oxymethine proton at  $\delta_H$ 5.26 (H-15) with the ester carbonyl signal at  $\delta_C$  174.1, while the  $\alpha$ -orientation of this unit was inferred by comparison of the <sup>13</sup>C NMR data with those from cinannamoylgrandifloric acid.<sup>22</sup> Thus, the structure of **3**, a new *ent*-kaurene derivative, was established as  $15\alpha$ -decanoyloxy-16-kauren-19-oic acid. Although the labdanes are the most frequently encountered diterpenes in plants of the genus *Eupatorium*, *ent*-kauranes are not uncommon.<sup>28,29</sup>

# **Experimental**

### General experimental procedures

Melting points were determined using a digital Mettler Toledo FP90 apparatus. The optical rotations were measured on a Perkin-Elmer 341 digital polarimeter. IR spectra were



Figure 2. Selected HMBC correlations for compound 3.

recorded using a Perkin-Elmer FT-IR 1000 spectrometer. EIMS was acquired with the direct insertion probe on a GC-MS Shimadzu spectrometer at 70 eV. NMR spectra were recorded on a Bruker Avance DRX-500 (500 MHz for <sup>1</sup>H and 125 MHz for <sup>13</sup>C) or DPX-300 (300 MHz for <sup>1</sup>H and 75 MHz for <sup>13</sup>C) spectrometers. Chemical shifts, given on the  $\delta$ scale, were referenced to the residual undeuterated portion of the deuterated organic solvent, for proton [(CD<sub>2</sub>)<sub>2</sub>CO:  $\delta_{\mu}$ 2.05; CDCl<sub>3</sub>:  $\delta_{\rm H}$  7.27], and the center peak of the deuterated solvent [(CD<sub>3</sub>)<sub>2</sub>CO:  $\delta_{c}$  206.68, 29.92; CDCl<sub>3</sub>:  $\delta_{c}$  77.23]. Column chromatography was run using silica gel 60 (70 -230 mesh, VETEC) and silica gel 60 (0.063-0.200 mm, MERCK). TLC was performed on precoated silica gel polyester sheets (Kieselgel 60 F<sub>254</sub>, 0.20 mm, Merck) by detection with a spray reagent of vanillin/perchloric acid/ EtOH solution followed by heating at 120 °C.

#### Plant material

*Eupatorium betonicaeforme* in the flowering stage was collected in May 2002, from Acarape County, Ceará, and identified by Prof. Elnatan B. Souza of the Universidade Estadual Vale do Acaraú - Ceará. Voucher specimen (# 29.104) has been deposited at the Herbario Prisco Bezerra (EAC) of the Departamento de Biologia, Universidade Federal do Ceará.

#### Extraction and isolation

Air-dried and powdered *E. betonicaeforme* parts were extracted with hexane: flowers (0.8 kg,  $2 \times 4$  L), aerial parts (2.5 kg,  $2 \times 9$  L) and roots (0.7 kg,  $2 \times 3$ L), followed by EtOH with same amounts of solvent. After evaporation of the solvents under reduced pressure afforded the crude hexane and EtOH extracts, respectively: flowers (15.9 g and 82.9 g), aerial parts (51.6 g and 75.9 g), and roots (4.6 g and 18.0 g). The hexane extract from the flowers (15.9 g) was fractioned over Si gel to yield the follow fractions: hexane (0.4 g), hexane/EtOAc 8:2 (14.0 g), hexane/EtOAc 1:1 (0.8 g) and EtOAc (0.3 g). The hexane/EtOAc fraction 8:2

(14.0 g) was chomatographed over Si gel, eluting with hexane/EtOAc (9:1, 1:1), EtOAc, and MeOH to afford four fractions (F1-F4). F4 (8.7 g, hexane/EtOAc 9:1) was subjected to successive Si gel flash column chromatography using hexane/EtOAc 9.5:0.5 to yield 3 (80.0 mg) and pentacosanoic acid 4 (12.0 mg). Similarly, F2 (1.7 g, hexane/EtOAc 1:1) was subjected to repeated Si gel flash column chromatography using hexane/EtOAc 9:1 mixture to yield 5 (13.0 mg) and 6 (5.8 mg). The hexane/EtOAc fraction 1:1 (0.8 g) was subjected to Si gel flash column chromatography using CHCl<sub>2</sub>/EtOAc 9:1 to give 145 subfractions of 10 mL each. 7 (16.0 mg) was isolated from fractions 65-72, while 8 (21.0 mg) was obtained from fractions 89-105. The EtOH extract of the flowers (82.0 g) was suspended in MeOH/H<sub>2</sub>O 3:1 and partitioned with CH<sub>2</sub>Cl<sub>2</sub>, EtOAc, and n-BuOH. The CH<sub>2</sub>Cl<sub>2</sub> extract was evaporated under reduced pressure yielding a residue (24.0 g), which was fractioned over Si gel and eluted with hexane/EtOAc (7.5:2.5, 1:1, 2.5:7.5) and EtOAc. The hexane/EtOAc (7.5:2.5) fraction (4.0 g) was chromatographed over Si gel eluting with hexane/EtOAc (9:1, 8:2, 7:3, 1:1), EtOAc and MeOH to afford seven fractions (F1-F7). F2 (1.2 g) was subjected to Si gel flash column chromatography using hexane/EtOAc (9.9:0.1) to afford 9 (40.0 mg), while from fraction F6 (0.1 g), using the same chromatographic procedure and the solvent system hexane/EtOAc (7:3) was isolated 10 (3.0 mg). The hexane/EtOAc (1:1) fraction (1.1 g) was subjected to Si gel flash column chromatography eluting with a CHCl<sub>3</sub>/EtOAc (8:2) gradient to yield 11 (4.2 mg) and 12 (370.0 mg). The hexane extract from the aerial parts (51.0 g) was fractioned over Si gel eluting with hexane, CHCl<sub>3</sub> and EtOAc. The hexane fraction (25.0 g) was chromatographed over Si gel using petroleum ether, CH<sub>2</sub>Cl<sub>2</sub> and EtOAc as eluent. An aliquot of the petroleum ether fraction (1.5 g) was subjected to hydrolysis with KOH solution to yield a free fatty acids mixture (0.07 g) and an unsaponifiable fraction (0.7 g)g). Methylation of the fatty acids with MeOH/HCl (1.0%) followed by GC/MS analysis of the methyl esters mixture allowed the identification of: hexadecanoic (65.0%), heptadecanoic (1.2%), octadecanoic (7.3%), eicosanoic (3.5%), docosanoic (9.0%), tricosanoic (1.0%), and tetracosanoic acids (3.5%). TLC analysis indicated that the CH<sub>2</sub>Cl<sub>2</sub> fraction (9.8 g) was rich in compound 9, which was recrystallized in acetone to afford taraxasteryl acetate (736.0 mg). The CH<sub>2</sub>Cl<sub>2</sub> fraction (21.0 g) after successive Si gel column chromatography using hexane and increasing amounts of EtOAc afforded a binary mixture of 13 and 14 (7.0

mg). The hexane fraction from the roots (4.6 g) was fractioned over Si gel eluting with hexane, hexane/ $CH_2Cl_2$  1:1,  $CH_2Cl_2$  and EtOAc. From the fraction hexane/ $CH_2Cl_2$  (1:1) fraction (1.6 g) was obtained a mixture of **15** and **16** (30.0 mg). The EtOH extract was fractioned over Si gel eluting with hexane/EtOAc (8:2, 6:4, 4:6, 2:8), EtOAc and MeOH. The hexane/EtOAc (8:2) fraction (2.7 g) was subjected to successive Si gel column chromatography using increasing amounts of EtOAc in hexane to yield **17** (42.0 mg).

15α-Decanoyloxy-kaur-16-en-19-oic acid (3). Colorless waxy solid;  $[\alpha]_{\rm D}^{20}$ -0.66° (*c* 0.05, CHCl<sub>3</sub>); IR (KBr)  $\nu_{\rm max}$ /cm<sup>-1</sup>: 3350-2700, 1730, 1694, 1464, 1242, 1165; EIMS (70 eV) *m*/*z* 472 [M]<sup>+</sup> (5), 318 (25), 300 (80), 285 (90), 255 (50), 104 (65), 91 (45), 71 (82), 57 (100); <sup>1</sup>H (500 MHz) and <sup>13</sup>C NMR (125 MHz) data, Table 1.

8β-Angeloyl-9β, 10β-dihydroxy-1-oxo-germacra-4E,11(13)dien-12,6α-olide (7). Colorless needles; mp 164-165 °C;  $[α]_D^{20}$ -0.24° (*c* 0.05, acetone); IR (KBr)  $ν_{max}$ /cm<sup>-1</sup>: 3437, 2929, 1736, 1699, 1154, 1107; EIMS *m/z* 378 [M]<sup>+</sup> (absent), 312 (10), 295 (5), 277 (15), 260 (12), 232 (20), 207 (12), 108 (13), 83 (100); <sup>1</sup>H (500 MHz) and <sup>13</sup>C NMR (125 MHz) data, Table 1.

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## References

- King, R M.; Robinson, H.; *The Genera of the Eupatorieae* (Asteraceae). Monographs in Systematic Botany, Missouri Botanical Garden: United States, 1987, vol. 22, p. 65.
- Bohlmann, F.; Mahantan, P. K.; Suwita, A.; Natu, A. A.; Zdero, C.; Dorner, W.; Ehlers, D.; Grenz, M.; *Phytochemistry* 1977, *16*, 1973.
- Bohlmann, F.; Banerjee, S.; King, R. M.; Robinson, H.; Phytochemistry 1984, 23, 1189.
- Boeker, R.; Jakupovic, J.; Bohlmann, F.; King, R. M.; Robinson, H.; *Phytochemistry* 1986, 25, 1669.
- González, A.G.; Barrera, J. B.; Hernández, A. C. Y.; Rosas, F. E.; Dominguez, X. A.; *Phytochemistry* 1985, 24, 1847.
- Oliveira, B. H.; Nakashima, T.; Souza Filho, J. D.; Frehse, F.; J. Braz. Chem. Soc. 2001, 12, 243.
- Carreras, C. R.; Rossomando, C.; Giordano, O. S.; *Phytochemistry* 1998, 48, 1031.

- Rios, M. Y.; Aguilar-Guadarrama, A. B.; Navarro, V.; *Planta Med.* 2003, 69, 967.
- Lang, G.; Passreiter, C. M.; Medinilla, B.; Castillo, J. J.; Witte, L.; *Biochem. Syst. Ecol.* 2001, 29, 143.
- Bandara, B. M. R.; Hewage, C. M.; Karunaratne, V.; Wannigama, G. P.; Adikaram, N. K. B.; *Phytochemistry* 1992, *31*, 1983.
- Tori, M.; Ohara, Y.; Nakashima, K.; Sono, M.; J. Nat. Prod. 2001, 64, 1048.
- Domínguez, X. A.; Quintanilla, G. J. A.; Rojas, M. P.; *Phytochemistry* 1974, 13, 673.
- Rücker, G.; Schenkel, E. P.; Manns, D.; Mayer, R.; Heiden, K.; Heinzmann, B. M.; *Planta Med.* **1996**, *62*, 565.
- Herz, W.; Govindan, S.V.; Kumar, N.; *Phytochemistry* **1981**, 20, 1343.
- 15. Herz, W.; Ramakrishnan, G.; Phytochemistry 1978, 17, 1327.
- Yang, S. P.; Huo, J.; Wang, Y.; Lou, L. G.; Yue, J. M.; *J. Nat. Prod.* 2004, 67, 638.
- El-Seedi, H. R.; Ohara, T.; Sata, N.; Nishiyama, S.; J. Ethnopharmacol. 2002, 81, 293.
- 18. François, G.; Passreiter, C. M.; Phytother. Res. 2004, 18, 184.
- Muschietti, L.; Gorzalezany, S.; Ferraro, G.; Acevedo, C.; Martino, V.; *Planta Med.* 2001, 67, 743.
- Albuquerque, M. R. J. R.; Silveira, E. R.; Uchôa, D. E. A.; Lemos, T. L. G.; Souza, E. B.; Santiago, G. M. P.; Pessoa, O. D. L.; *J. Agric. Food Chem.* 2004, *52*, 6708.
- Itoh, T.; Kikuchi, Y.; Tamura, T.; Matsumoto, T.; *Phytochemistry* 1981, 20, 761.
- Nascimento, A. M.; Oliveira, D. C. R.; J. Braz. Chem. Soc. 2001, 12, 552.
- Bohlmann, F.; Zdero, C.; Jakupovic, J.; Gerke, T.; Wallmeyer, M.; King, R. M.; Robinson, H.; *Liebigs Ann. Chem.* 1984, 162.
- Gao, F.; Wang, H.; Mabry, T. J.; Watson, W. H.; Kashyap, R. P.; Phytochemistry 1990, 29, 551.
- Ahmad, V. U.; Rahman, A. U.; *Pentacyclic Triterpenoids*, *Handbook of Natural Products Data*, Elsevier: Amsterdam, vol. 2, 1994.
- Agrawal, P. K.; Thakur, R. S.; Bansal, M. C.; In *Studies in Organic Chemistry 39 Carbon-13 NMR of Flavonoids*; Agrawal, P. K., ed.; Elsevier Science Publishing Company Inc.: New York, USA, 1989.
- Bohlmann, F.; Zdero, C.; Lonitz, M.; *Phytochemistry* 1977, 16, 575.
- 28. Herz, W.; Sharma, R. P.; J. Org. Chem. 1976, 41, 1021.
- 29. Herz, W.; Govindan, S. V.; Blount, J. F.; J. Org. Chem. 1979, 44, 2999.
- Velandia, J. R.; Carvalho, M. G.; Braz-Filho, R.; *Quim. Nova* 1998, 21, 397.