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Comparative study of inclusion removal from liquid steel in the RH and argon purging processes

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Abstract

In order to improve steel quality, it is necessary to increase the cleanliness of the liquid steel; that is, to reduce the number and size of the inclusions in the liquid steel, as well as to control their chemical composition. For this purpose, processes (with different operating costs) are used, such as the bubbling of inert gas in the steel ladle and its treatment in the vacuum degasser RH. This article deals with inclusion removal through an argon purging process and RH treatment, and provides a comparison of their effectiveness using industrial data and a mathematical model. The inclusion count shows a strong exponential decay for RH treated heats with chemical heating, as expected. The same behavior is not seen for RH treated heats without chemical heating as well as an argon purging process. It is suggested that in the later cases, there could exist competition between inclusion removal and inclusion generation from sources, such as refractory-slag-metal interaction and open eye atmosphere metal interaction. It has been concluded that the RH and the argon purging processes are both able to reduce the amount of inclusions. However, the bubbling process would require a much smaller gas flow rate to avoid open eye formation.

Keywords: inclusions, RH, bubbling station.

1. Introduction

The composition, quantity and size distribution of nonmetallic inclusions strongly influence the chemical and physical properties of steel, such as fatigue, machinability and corrosion resistance, whereby improving steel quality implies increased liquid steel cleanliness (Yang *et al.*, 2014). Many processes in secondary metallurgy, such as lade furnace, VOD, RH and argon purging are used with the purpose of impurities removal.

The inert gas injection is commonly practiced in ferrous and nonferrous metal secondary metallurgy processes and this technique is used for obtaining temperature and chemical homogeneity of the liquid metal and for helping the removal of secondary phases and impurities dispersed in liquid metals (Wang *et al.*, 1996). Although inert gas injection is a low-cost process, it can jeopardize the steel quality, since if the gas is not well controlled, the steel can be exposed to an oxidizing atmosphere, which could imply in inclusion generation.

Beyond the argon purging process (APP), the attention given to the RH process (which is normally used for dehydrogenation, denitrogenation and decarburation) has been growing due to its strong ability for removing inclusions (Yang *et al.*, 2013). One advantage of this processes in comparison to the APP is that it does not present the same oxidizing atmosphere exposition risk as long as vacuum exists inside the treatment chamber.

The development of the APPs has been focusing on two main conditions: obtaining small bubbles and good mixing. Small bubbles imply in a large gas-liquid interface, resulting in a high probability of adhesion between the inclusion and bubbles; a good mixing condition improves the mass transfer efficiency (Zhang and Taniguchi, 2000). The larger the gas bubble, the lower the particle removal rate is, so smaller bubbles can favor inclusion removal (Zhang *et al.*, 2016).

Söder *et al.* (2004) studied the growth and removal of inclusions in the APP through a static model and suggested that the turbulent collision is the most important phenomena, improving as the size difference between the inclusions increases.

Cao and Nastac (2018) have studied the inclusions transport and removal in a gas stirred ladle trough, a 3D computational fluid dynamics (CFD) model. They concluded that the absorption of the inclusions by the slag is the main inclusion removal mechanism, and that raising the gas flow rate favors this mechanism.

Marins (2011) highlighted that there is a possibility of steel contamination (possible steel re-oxidation and hydrogen and nitrogen pick up) during bubbling due to the steel-slag-atmospheric air interaction, so it is important to control the gas flow rate in order to avoid the metal-slag emulsion occurrence and the open eye onset.

Shirabe and Szekely (1983) performed a mathematical simulation of an RH system and observed that the flow in the ladle is highly turbulent (mainly below the down leg), emphasizing that the RH is an excellent mixer.

High turbulence and non-oxidizing atmosphere during treatment may indicate that this equipment could be ideal for inclusion removal.

Miki et al. (1997) developed a mathematical model for predicting the evolution of inclusion size and spatial distribution during the steel treatment in RH. According to them, one minute after the aluminum addition, it is possible to find large dendritic inclusions, but after fifteen minutes, these inclusions disappear, and larger aggregates are to be found in exchange. The authors concluded that the large inclusions are able to float, while the smaller ones aggregate. These aggregates formed by the small inclusions float with increasing difficulty because, as the steel fills the blank spaces between the particles, their resulting density approximates the steel density. Later, Lascosqui (2006) evaluated the influence of circulation time on the steel cleanliness, concluding that after 300 seconds, the inclusion concentration is almost independent of the initial concentration.

These studies suggest that both processes can be used for inclusion removal from liquid steel. However, although the inclusion removal mechanisms have been discussed, comparison between both methods are scarce and little is discussed about the necessity of using one process or the other for achieving a certain level of steel cleanliness. This last consideration is important, since the RH process is usually more costly than the bubbling process. This article provides a comparison between the two processes.

2. Materials and methods

Mathematical simulations and industrial data collection were made in order to evaluate the inclusion removal efficiency of APP and RH processes. The mathematical simulation was proposed to analyze the steel flux in the

2.1 Mathematical simulations

Mathematical simulations of steel flow in a 224 t liquid steel ladle have been performed for APP (gas flow rate of 250 NL/min) and RH (gas flow rate of 100 Nm³/h) processes. Figure 1 shows the main dimensions of the ladle and RH reactor.

vessel and ladle. The velocity and dissipation of turbulent kinetic energy rate fields were obtained for the argon purging and RH process. The industrial data were collected manually in six heats during the APP and RH treatments.

Simulations have been performed using CFX 19.1 (Ansys®) software. The model was the k-ε turbulence model for the continuous phase (liquid), while for the dispersed phase (gas), the zero-equation model is adopted. For the RH reactor, the Ishii-Zuber models were adopted for the

Then, an inclusion count was made to determine the inclusion removal during the processes. Combining these results, it is possible to understand the processes efficiency and the probable causes for differences observed.

drag force with the virtual mass coefficient equaling 0.25. The Frank model for the wall lubrication force and Favre average for the turbulent dispersion force, according to Peixoto (2019) were also used. For the ladle APP simulation, only the drag and turbulent dispersion forces were considered.

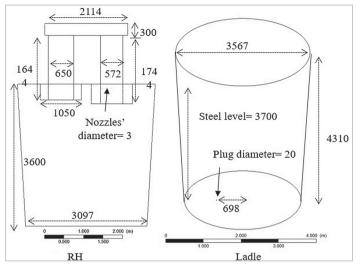


Figure 1 - Main dimensions of ladle and RH (mm).

The values of argon (25 °C) and steel (1600 °C) properties used in this simulation are the same given by Peixoto (2019) and are described in Table 1. It is assumed that

Where: G - gas flow rate (Nm³/s); g - acceleration of gravity (m/s²);

For the RH reactor in a symmetrical condition, only half of the reactor was considered. The boundary conditions applicable to the problem are: (i) Non-slip condition applied to all walls, regions where the fluid has zero velocity; (ii) Symmetrical Condition: in this region the velocity component's normal to the boundary is zero; (iii) Injection nozzles: gas

the gas bubble diameter is constant and is calculated from the experimental correlation given by Equation (1), from Johansen and Boysan (1998). The bubble size was

$$d_b = 0.35 \left(\frac{G^2}{g}\right)^{0.2}$$

flow rate in kg/s (considering symmetrical conditions), subsonic flow regime with a turbulence intensity of 5%; (iv) Free slip condition on ladle free surface; (v) Vacuum chamber surface: opening condition, with pressure equal to the applied vacuum. For APPs, the boundary conditions appli-

cable to the problem are: (i) Non-slip condition applied to all walls, regions where the fluid has zero velocity; (ii) porous plug: gas flow rate in kg/s, subsonic flow regime

around 7.5 mm for APP and ranged from 17 mm to 47 mm for RH reactor due to gas expansion caused by the sharp pressure drop, in accordance with Peixoto (2019).

with a turbulence intensity of 5%; (iii) ladle free surface: opening condition, with relative pressure equal 0 Pa.

The element sizing of the grid was set like Peixoto (2019) for the RH simulation. For APP a densely packed mesh was applied in the region above the porous plug. Thus, the calculation domain had approximately 380 thousand grid nodes (930 thousands grid elements), which is sufficiently fine to give reasonably grid—independent results.

Table 1	-	Materials'	properties.
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	Density (kg.m ⁻³)	Viscosity (mPa.s)	Interfacial tension (N.m ⁻¹)
Steel	7000	5.7	Steel/argon 1.54
Argon	1.623	0.04848	

2.2 Industrial tests

The present article evaluates inclusion behavior during aluminum and silicon killed steel processing; these are steels used to produce civil construc-

tion plates with 0.13-0.17% carbon and 0.8-0.9% manganese. Six heats were produced in the routes below in secondary metallurgy treatments as

showed in Table 2: BOF – RH degasser – slab continuous casting machine BOF – bubbling station - slab continuous casting machine.

Table 2 - Secondary refining processes applied to the six heats.

Heat	Secondary refining process		
1	Argon purging process (APP)		
2	Argon purging process (APP)		
3	RH with chemical heating		
4	RH with chemical heating		
5	RH without chemical heating		
6	RH without chemical heating		

3. Results and discussion

Lollipop samples were taken manually in all the heats in different treatment moments, as can be seen in section 3.2. The samples were prepared by sanding and polishing according to the ASTM E3 standard. Then, an inclusion count and analyses were made using a MEV/EDS (ASPEX) with an FEI inclusion analyzing software ("Aspex Explorer") in a scanned area of $2.33 \times 10^7 \ \mu m^2$. The inclusions with less than $2 \mu m$ diameter were declassified.

3.1 Mathematical simulation

Figures 2 and 3 show the results of the mathematical simulation for the steel flow in the stirred ladle and

in the RH, respectively. The resulting profiles are quite different, and they may indicate that the inclusion removal efficiency of these processes may differ considerably.

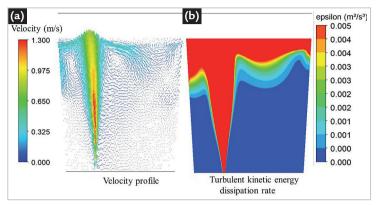


Figure 2 - Mathematical simulation of the steel flow in a gas stirred ladle.

It can be noticed that the region with the larger steel velocities in the stirred ladle is located mainly above the porous plug (Figure 2); this same region depicts the larger rates of dissipation of turbulent kinetic energy, which is due to gas bubble swarming.

The higher dissipation of turbulent kinetic energy rate in this area may indicate that it favors the collision and aggregation of inclusions, but the gas bubble traffic time to the surface may not be large enough, so this phenomena could occur before the inclusions reach the slag layer. In addition, some vortices are observed close to the slag layer, which may indicate that the inclusions, after reaching this layer, may be carried back to the center of the bath.

In the RH (Figure 3), the regions with larger velocities and rate of dissipation of turbulent kinetic energy are better distributed as compared with the gas stirred ladle. There are three main regions (at the up leg, at the bottom of the ladle and under the down leg, as

also pointed out by Shirabe and Szekely (1983) and Lascosqui (2006)) where the rate of dissipation of turbulent kinetic energy is high, favoring the collision and agglomeration of inclusions.

The flow pattern suggests that then inclusions or clusters are carried away towards the floating slag layer and vortices that could carry the inclusions back to the bath after reaching the slag layer are not observed. There is a favorable combination of stirring intensity and flow pattern, which is not observed in the ladle.

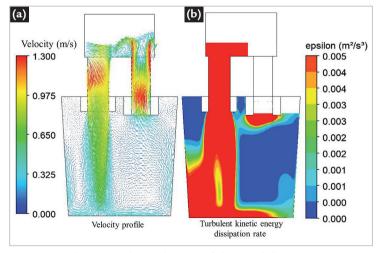


Figure 3 - Mathematical simulation of the steel flow in a RH reactor.

3.2 Industrial tests

Concerning the industrial tests, the following figures show the evolution of inclusion content along the treatment. Q/Q

is the fraction of inclusions remaining in the steel (as compared with the initial inclusion count). The results are presented as the global count (concerning all the inclusion sizes) and number of inclusions by inclusion size ranges.

3.2.1 Heats treated in APP station

Data for Heat #1 (APP) are given in Figure 4. Data for Heat #2 (APP) are given in Figure 5.

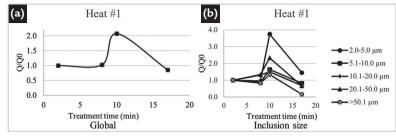


Figure 4 - Ratio between the inclusion counting and the initial inclusion quantity Q/Q_0 - Heat # 1: (a) global Q/Q_0 ; (b) according to the inclusions size.

In both treatments, an increase in the inclusion quantity and a further decrease

is observed, which indicate that inclusions were generated during the treatment.

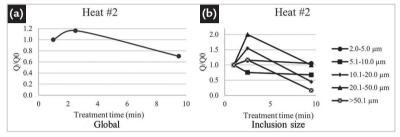


Figure 5 – Ratio between the inclusion counting and the initial inclusion quantity Q/Q_0 – Heat # 2: (a) global Q/Q_0 ; (b) according to the inclusions size.

Heats #1 and #2, produced in the APP station, show an initial increase of the inclusion counting which may

be due to steel re-oxidation by the open eye and stronger interaction with the floating slag. The heats show a small decrease of 22% (average) of the number of inclusions at the end of the treatment.

3.2.2 Heats treated in RH with chemical heating

Heats #3 and #4 refer to RH treatment with chemical heating (aluminum addition

plus oxygen blowing). The shown data are for samples taken after oxygen blowing.

Data for Heat #3 are given in Figure 6. Data for Heat #4 are given in Figure 7.

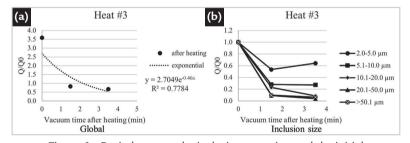


Figure 6 – Ratio between the inclusion counting and the initial inclusion quantity Q/Q_{\circ} – Heat # 3: (a) global Q/Q_{\circ} ; (b) according to the inclusion size.

As heats 3 and 4 have gone through chemical heating (aluminothermy), after this

procedure the inclusion content increases more than 400% (average), dropping exponentially right afterwards. This behavior was observed in every inclusion size range.

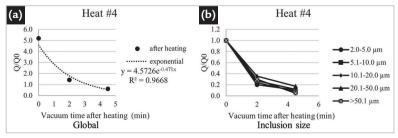


Figure 7 – Ratio between the inclusion counting and the initial inclusion quantity Q/Q_0 – Heat # 4: (a) global Q/Q_0 ; (b) according to the inclusion size.

The equations in Figures 8 and 9 relate to the work of Aoki et al. (2005),

where the authors studied the inclusion removal during argon purging in a steel ladle. The authors evaluated a deoxidation constant such as:

$$\frac{d[O]}{dt} = -k_o[O] \tag{2}$$

3.2.3 Heats treated in RH without chemical heating

Heat # 5 and #6 refer to RH treated steels but without chemical heating; however, some aluminum is

added to complete the deoxidation process. The initial inclusion counting is given by the previous deoxidation process. Data for Heat #5 are given in Figure 8. Data for Heat #6 are given in Figure 9.

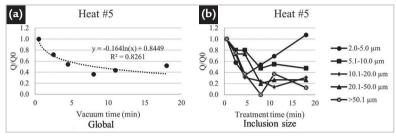


Figure 8 – Ratio between the inclusion counting and the initial inclusion quantity Q/Q_0 – Heat # 5: (a) global Q/Q_0 ; (b) according to the inclusion size.

Heat number 5 shows an increase in the number of inclusions of the 2-5µm range right after the

aluminum addition until the end of the treatment and this behavior is not observed in Heat #6. In the other ranges, a decrease in the initial inclusion amount is observed in Heats 5 and 6.

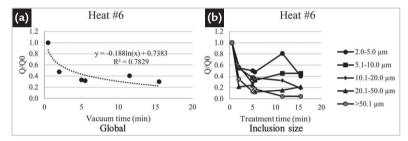


Figure 9 – Ratio between the inclusion counting and the initial inclusion quantity Q/Q_{\circ} – Heat # 6: (a) global Q/Q_{\circ} ; (b) according to the inclusion size.

Although the initial inclusion counting of Heat #6 is almost twice that of the initial inclusion content of Heat #5, the final inclusion counting is almost the same, which suggests that the initial inclusion content does not influence the final result. For vacuum times of approximately 20 min (typical for most of the treatments), it is possible to achieve a reduction of 60% of the initial inclusion amount.

According to Yang *et al.* (2015), the inclusions removal in an RH reactor is directly proportional to the treatment time, so the inclusion removal would always decrease as long as the vacuum and steel mixing go on. In disagreement to Yang *et*

al. (2015), the present article shows that after some minutes under treatment, the inclusion number asymptotically tends to a fixed value.

The initial amount of inclusions in Heats #3 to #5 is quite similar. However, an exponential decay is not observed for Heats #5 and #6, and a logarithmic fitting is more appropriate, even though the samples were taken from the same equipment. This difference can be assigned to the timing of deoxidation, which in Heats 3 and 4 was performed after the chemical heating and in Heats 5 and 6 during tapping of the BOF. This is the reason why the initial inclusion count of the various

heats is four times lower than after chemical heating.

No clear tendency was observed as far as bubbling station data are concerned, so it is not possible to compare this process to the ones studied by Aoki *et al.* (2005). Nevertheless Aoki *et al.* (2005) suggest that the bubbling station is as capable of removing the inclusions as the RH, but this performance would only be achieved with a gas flow rate seven times lower than the one used in this study. This excessive flow rate may have caused competition between the inclusion removal process and incorporation phenomena, such as emulsification and reoxidation in Heats 1 and 2.

4. Conclusions

The APP and RH processes can be used for inclusion removal from liquid steel, even when aluminothermic chemical heating is required. As shown by the

analysis of industrial data, this removal comes with a much higher predictability and efficiency in the RH as compared to the bubbling station; a decrease of 60%

is achievable in about twenty minutes under vacuum.

The mathematical simulations showed that the higher dissipation of

turbulent kinetic energy rate in the RH in comparison with the APP indicates that this equipment has a greater efficiency in aggregating inclusion. Besides, recirculating flow was observed in the APP simulation indicating that inclusions may return to the metal bath after reaching the slag layer. Also, concerning the RH treatment, industrial data showed that in spite of the different initial inclusion content, after some minutes under treatment, the

inclusion numbers in the heats are almost the same, which means that the initial amount of inclusions does not influence the final amount.

An exponential decay of inclusion count is noticed for RH treated steels with chemical heating, just as expected. This behavior has not been found for RH treated steels without chemical heating. For the latter, it is presumed that there occurs a competition process between in-

clusion generation and inclusion removal, which leads to an asymptotic decay to a specific value.

From comparison with Aoki *et al.* (2005) data, it is suggested that it would be possible to obtain an inclusion removal at the bubbling station as efficient as at the RH. Nevertheless, for achieving this purpose, a flow rate around seven times smaller than the one used in this study is required.

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