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# Effect of Cage Configuration in Structural and Optical Properties of TiN Films Grown by Cathodic Cage Discharge

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Received: October 17, 2012; Revised: December 17, 2012

Cathodic cage discharge was developed recently in order to eliminate phenomena as edge effect and overheating, which occurs during conventional processes. In this study, the effect of cage configuration in active species during the deposition process and optical properties of TiN film were studied. TiN compound was chosen because its optical properties are very sensitive to slight variations in microstructure and film thickness, becoming a good monitoring tool in fabrication process control. Cages were made of titanium and have different holes numbers and holes diameter. Electrical efficiency of the system and optical properties of TiN films were strongly influenced by experimental conditions. It was found that with more holes at the top of cage, deposition rate and crystallinity were higher, if compared to cages with a small number of holes at the top. On the other hand, the opposite behavior was observed when more holes were located at the sidewall of cage.

**Keywords:** *cathodic cage, thin films, titanium nitride, Optical Emission Spectroscopy, plasma deposition, optical properties* 

## 1. Introduction

Cathodic cage plasma technique was recently developed in order to eliminate edge effect and overheating, which occurs during thermochemical treatment with plasma. These effects take place due to high superficial area/ volume ratio, mainly in pieces with complex geometry<sup>1-3</sup>. The literature reports provided evidence that cathodic cage reduce the edge effect in steel samples<sup>3,4</sup>, improve the temperature distribution throughout the sample and allow the nitriding of pieces with different dimensions<sup>1</sup>. The most recent researches have been regarding the efficiency of cathodic cage in order to modify surfaces of titanium alloys<sup>5</sup>, polymer<sup>6</sup> and deposit silver particles on stainless steel substrates<sup>7</sup>.

This technique is used to protect electrically the workpiece during the process, resembling a Faraday cage device. An isolator is placed between the workpiece and the cathode, in order to keep the workpiece in a floating potential<sup>4</sup>. In this configuration, only the cage is in cathodic potential, producing a hollow cathode effect in each hole. Due to the higher density of ions formed in the holes of the cage; there is a higher sputtering rate in that region. The combinations of the gas species with the sputtered species present in plasma are directed towards the substrate surface, where they are deposited and diffused onto the sample<sup>8</sup>. Thus, it is possible to produce a hybrid process of deposition and diffusion. Furthermore, the literature reported by Nishimoto<sup>9</sup> and Sousa<sup>10</sup> provided evidence that the distance

between the sample and cage influence in the properties of the nitrited sample, since nitrited layer thickness increased as the distance between sample and cage decreased.

Optical properties of titanium nitride films are very sensitive to even small variations in chemical composition and thickness. Therefore, this sensibility is a useful tool to monitoring the fabrication process control. Moreover, titanium nitride films are used in various industrial applications, for example, as coatings for high hardness and low friction in metallurgical industry<sup>11</sup>, decorative coatings replacing gold, since different color tones may be achieved varying Ti/N ratio<sup>12</sup>, as well as coatings for solar cells, solar control windows<sup>13</sup>, biomaterials<sup>14</sup> and microelectronic semiconductors<sup>15</sup>.

Because this technique was developed recently, there are few studies about the effect of process parameters on the characteristics of the grown film. Therefore, this study proposed to investigate the efficiency of this technique to obtain TiN thin films on glass substrates. The effects of cage configuration on the deposition rate, electrical parameters, microstructure, topography and optical properties of the grown film were determined by Optical Emission Spectroscopy (OES), Atomic Force Microscopy (AFM), X-Ray Diffraction, Spectrophotometry and Ellipsometry.

#### 2. Material and Methods

Rectangular samples of borosilicate glass (each with  $25 \times 10 \text{ mm}^2$  surface area and 2 mm thickness) were used as substrate in this study. The chemical composition of

borosilicate glass was 72.1% of silicon dioxide, 14.3% of sodium oxide, 6.3% of calcium oxide, 4.1% of magnesium oxide, 1.1% of aluminum oxide, 1.1% of potassium oxide and 1.0% of other oxides. The deposition process was conducted in an ion nitriding reactor adapted to cathodic cage configuration. The substrates were in a floating potential because they were electrically isolated from the cathode trough an alumina disc, as illustrated in Figure 1.

The cages were made of commercially pure titanium (grade II), they have 1 mm of thickness and 70 mm of diameter. Five different cages were used and denominated as  $L_1T_4F_{12}$ ,  $L_1T_8F_{12}$ ,  $L_2T_4F_{12}$ ,  $L_2T_8F_{12}$  and  $L_1T_8F_8$ . The L is referent to the number of holes lines in the sidewall of cage and T is referent to the number of holes lines in the top (covering) of the cage and F referent to hole diameter in millimeter. Table 1 shows the experimental conditions.

Pressure, temperature and duration time were fixed at 1.5 mbar, 450 °C and 120 minutes, respectively. The plasma atmosphere was compounded by 50% of Ar, 37.5% of  $N_2$  and 12.5% of  $H_2$ . Argon was used to increase the titanium sputtering rate and to control nitriding rate of the cage<sup>16</sup>. Hydrogen was used to reduce the presence of superficial oxides and to increase process efficiency<sup>17</sup>.

The deposition process was monitored by Optical Emission Spectroscopy (OES), using an Ocean Optics USB 4000 spectrograph. The X-Ray Diffraction analyses were performed with a Shimadzu XRD-6000 diffractometer using Cu radiation K $\alpha$  and an accessory for Grazing Incidence X-ray Diffraction (GIXRD) and angle of incidence fixed at 0.5°. The grown films were also analyzed by Atomic Force Microscopy (AFM) in contact mode with a Shimadzu microscope model SPM 9600. Optical properties were analyzed by transmittance, reflectance using a Varian Cary-5000 spectrophotometer with integration sphere. Dispersion curves, n( $\lambda$ ) and k( $\lambda$ ), were obtained by spectral ellipsometry, using Sopra GES-5E Spectral Ellipsometer.

### 3. Results and Discussion

Figure 2 shows a typical OES spectrum obtained during cathodic cage deposition at the  $L_1T_8F_{12}$  configuration, where some lines corresponding to Ar,  $H_2$  and  $N_2$  transitions are highlighted.

The intensities lines relative to argon (750.3 nm) of  $N_2^+$ ,  $N_2$  and H $\alpha$  species, with respective emissions lines at 391.4 nm, 337 nm and 656.3 nm and the powder supplied at different cage configurations are shown in Figure 3. The H $\alpha$  and  $N_2^+$  intensities varied in function of cathodic cage configuration. The plasma efficiency in the heat transfer was estimated by the ratio between the power supplied in the system (electric power) and the average of the species



Figure 1. Schematic illustration of plasma reactor in cathodic cage configuration.

Table 1.	Experimental	conditions	used in	this	study
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Sample Cu	Current (A)	Voltage	Power	Cage height	Hole diameter (mm)	Number of holes	
		( <b>V</b> )	(W)	(mm)		Wall	Тор
$L_1 T_4 F_{12}$	0.30	860	258	34	12	16	8
$L_1 T_8 F_{12}$	0.31	820	254	34	12	16	12
$L_2 T_4 F_{12}$	0.33	811	267	45	12	31	8
$L_2 T_8 F_{12}$	0.36	878	316	45	12	31	12
$L_{1}T_{8}F_{8}$	0.30	805	285	34	8	23	17

energy (line intensities). These results indicated that the optimal configurations were the  $L_1T_8F_8$  and  $L_1T_8F_{12}$  cages.

The plasma efficiency increased as the amount of holes at the cage top increased. However, the highest efficiency was achieved when the lowest amount of holes was present at the sidewall of cage. Apparently, this result was contradictory, since it is expected that heating improve due to the higher area with hollow cathode effect. This result was justified by the farther distance between the pieces and the cage top, when the holes number in the sidewall was increased, because the heat from these holes was transferred to the chamber walls, to the inner and to the outer cage. Therefore, there was an appropriate height at 34 mm which the maximum efficiency was promoted due to combination of heating by hollow cathode effect and distance between the sample and the cage top.

The XRD patterns of deposited films at different cage configurations (Figure 4) exhibited two peaks at 37.3° and 43.3° corresponding to  $\delta$ -TiN (111) and  $\delta$ -TiN (200), respectively. The most intense TiN peak was observed at the  $L_1T_8F_{12}$  and  $L_1T_8F_8$  cage configuration. On the other hand, the  $L_1T_4F_{12}$  configuration exhibited the lowest deposition rate evidenced by low signal and high noise ratio.

Table 2 shows  $I_{H}\alpha/I_{Ar}$  line intensity values obtained from OES spectra (Figure 3), the TiN (200) peak intensity (Figure 4) and roughness values (Figure 5). There was a correlation between the luminous intensity of H $\alpha$  species



Figure 2. Plasma spectra obtained during deposition process at  $L_1T_8F_{12}$  configuration.



Figure 3. Luminous intensities of plasma active species at different cage configurations (bars) and values of the power supplied to keep the system at 450 °C (black dots).

and the grown film characteristics. The highest deposition rate, (demonstrated by the intensity of TiN (200) diffraction peak) occurred when the  $I_{H}\alpha/I_{Ar}$  ratio was highest. Therefore the H $\alpha$  species plays a key role in the deposition rate of TiN, similar to literature reports<sup>18-21</sup>: there was an increasing in deposition rate since the hydrogen reduced the presence of superficial oxides on the cage (cage poising), promoting a formation of more stoichimonetry film.

Furthermore, it was possible notice (Table 2) that the increasing in the number of holes in the sidewall promoted a reduction of film roughness, which was directly related to the decreasing of sputtering rate on the substrate, seeing as the increasing of the distance between the pieces and the cage top reduced the probability plasma particles collide with the substrate.

The few variations in refraction index and extinction coefficient (Figure 6) were related to different microstructure, because the refraction index vary in function of thickness,

**Table 2.** Luminous intensity of  $H\alpha$  species, characteristics of the crystal structure and topography of films deposited at different cage configuration.

Cage	$I_{H}\alpha/I_{Ar}$	Intensity	Roughness	
configuration		δ-TiN (200)	Ra	Rms
$L_{1}T_{8}F_{12}$	53643	6676	4.27	3.53
$L_2 T_8 F_{12}$	44900	3862	1.77	2.24
$L_1 T_4 F_{12}$	41989	2520	3.66	2.95
$L_1T_8F_8$	46629	6118	5.35	6.61



Figure 4. X-ray pattern for films deposited at different cage configurations.

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Figure 5. Nanotopography of the TiN films grown at (a)  $L_1T_8F_8$ , (b)  $L_1T_8F_{12}$ , (c)  $L_2T_8F_{12}$  and (d)  $L_1T_4F_{12}$  cage configurations.



Reflectance (%) L1T8F8 L1T8F12 L1T4F12 1000 1100 Wavelengh (nm)

Figure 6. Dispersion plots, real part n ( $\lambda$ ) and imaginary part k ( $\lambda$ ), obtained by spectral ellipsometry for film deposited at three different cage configurations.

Figure 7. Spectra of light reflectance for films obtained at different cage configurations.



Figure 8. Transmittance spectra for films obtained at different cage configurations.

chemical composition and porosity<sup>22,23</sup>. Moreover, those variations in refraction index resulted in reflectance (Figure 7) and transmittance spectra (Figure 8) modifications.

The transmittance values were different for films deposited at each cage configuration (Figure 8). The lowest transmittance indicated that there was the highest thickness since the crystal phase formed was almost the same for every condition as evidenced by X-Ray pattern (Figure 4). These results accorded to other results obtained by optical emission spectroscopy and X-Ray diffraction:

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the films obtained at  $L_1T_8F_8$  and  $L_1T_8F_{12}$  cage configuration had the lowest transmittance values, thus the highest deposition rates.

#### 4. Conclusions

Based on the experimental results of this study, the following conclusions can be drawn:

- TiN films can be grown on glass substrates by cathodic cage discharge;
- The deposition rates, microstructures and optical properties of grown films were influenced by the gas species density and cage configuration;
- The L<sub>1</sub>T<sub>8</sub>F<sub>8</sub> and L<sub>1</sub>T<sub>8</sub>F<sub>12</sub> cage configurations, in this order, were the most efficient when the ratio between supplied power and average energy of the species was considered;
- In general, configurations with more holes at the top of the cage exhibited higher efficiency. However, it was not observed when the holes at sidewall of the cage were more numerous. Actually, in this case, the efficiency was lower;
- There was a correlation between the luminous intensity of H $\alpha$  species and the characteristics of the grown film. The highest deposition rate and the highest crystallinity were obtained with the highest  $I_{\mu}\alpha/I_{Ar}$  ratio; and
- The optical properties varied in function of microstructure and deposition rate of grown film.
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